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# The Flavonoids of Orophea polycarpa A. DC. (Annonaceae)

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#### ABSTRAK

Kajian ke atas bahagian neutral *Orophea polycarpa* A. DC. telah menghasilkan dua flavonoid yang dikenali sebagai 2'-hidroksi-4,4',6'-trimetoksikalkon dan 5,7,4'-trimetoksiflavanon.

#### ABSTRACT

Investigation on the neutral fraction of the leaves of *Orophea polycarpa* A. DC. led to the isolation of two flavonoids identified as 2'-hydroxy-4,4',6'-trimethoxychalcone and 5,7,4'-trimethoxyflavanone.

Keywords: chalcone; flavanone; Orophea polycarpa; <sup>1</sup>H nuclear magnetic resonance spectrometry; <sup>13</sup>C nuclear magnetic resonance spectrometry; infra-red & uv-visible spectrophotometry; flavonoids.

## INTRODUCTION

Continuing investigations on the medicinal plants of Peninsular Malaysia, the chemical constituents, including the isolation of flavonoids, of *Orophea polycarpa* A. DC. (Annonaceae) were studied. *Orophea polycarpa* has not previously been investigated chemically but two aristolactams, enterocarpam I and enterocarpam II have recently been reported from *O. enterocarpam* Maingay ex Hk.f. & Thoms (Mahmood *et al.* 1986). *Orophea polycarpa* is a small tree, approximately 8 m tall which grows in the north of Peninsular Malaysia and is reported to have bechic and sudorific properties (Perry and Metzger 1980).

#### MATERIALS AND METHODS

#### General

The leaves of this species were collected from the foot of a limestone hill in Langkawi Island, off the northwest coast of Peninsular Malaysia. Voucher specimen No: 4718 was deposited in the herbarium of the Biology Department, Universiti Pertanian Malaysia.

Melting points were determined on a Kofler hot stage and were uncorrected. Infrared and ultraviolet spectra were recorded on Beckmann Acculab 3 and Hitachi 200-20 spectrophotometers, respectively. <sup>1</sup>H NMR and <sup>13</sup>C NMR were recorded on a Bruker CPX 300 instrument at 300 and 75.5 MHz, respectively. Mass spectra were obtained from a Hewlett-Packard 5970 Mass Selective Detector coupled to a Hewlett Packard 5890 gas chromatography fitted with an open split capillary interface and HP-1 (polymethylsiloxane,  $0.17\mu$ ; 25 × 0.32 mm i.d.) capillary column with helium carrier gas (2 ml/min. flow rate; 15 psi). Column and thin-layer chromatography utilized Merck 7734 and 7730 silica gel, respectively.

### Extraction

The air-dried leaves (1.1 kg) were ground to powder and defatted with petroleum ether (2  $\times$  2 L). The combined methanol extracts were evaporated under reduced pressure to give 24.1 g of crude extract which was fractionated between chloroform and 2N aqueous sulphuric acid. The chloroform fraction was evaporated to give 5.0 g of dark green semi-solid which was subjected to column chromatography (5 cm i.d.  $\times$  35 cm) using silica gel as adsorbent and chloroform:methanol mixture as eluent in increasing polarity manner.

The combined major fractions consisted of two major components which were separated on preparative thin-layer chromatography using silica gel and developed with 2% methanol in chloroform. The more polar compound-I (30 mg) which crystallized from  $\text{CHCl}_3$ , was found to convert into the less polar compound-II (40 mg) on standing in N<sub>9</sub> atmosphere.

Compound-II, 5,7,4'-trimethoxyflavanone. Bright yellow crystals: m.p. 115-117 °C (lit; 123.5-124.5°C, Kaufmann and Lam 1967; 195°C, Mishra and Misra 1980). Spectral data were in accord with literature.

UV	:	Kaufmann and Lam 1967; Lam and Wrang 1975.
IR	:	Kaufmann and Lam 1967; Dutta and Som 1978.
<sup>1</sup> H NMR	:	Kaufmann and Lam 1967; Dutta and Som 1978.
<sup>13</sup> C NMR	:	Duddeck et al. 1978.
MS	:	Lam and Wrang 1975.

*Compound-I, 2'-hydroxy-4,4',6'-trimethoxychalcone.* Pale yellow crystals: m.p. 95-97° (lit. 115°C, Tuntiwachwuttikul *et al.* 1987). Spectral data were in accord with literature.

UV : Hansel *et al.* 1963; Lam and Wrang 1975; Tuntiwachwuttikul *et al.* 1987.

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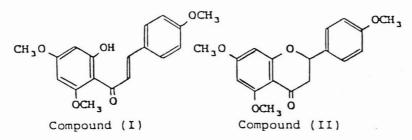
IR	:	Hansel et al. 1973; Dutta and Som 1978; Tuntiwachwuttikul
		et al. 1987.
<sup>1</sup> H NMR	:	Dutta and Som 1978; Tuntiwachwuttikul et al. 1987.
<sup>13</sup> C NMR	:	Duddeck et al. 1978.
MS	:	Lam and Wrang 1975; Dutta and Som 1978;
		Tuntiwachwuttikul et al. 1987.

# **RESULTS AND DISCUSSION**

Two compounds were isolated from the leaves of *Orophea polycarpa*, one of them being converted to the other on standing at room temperature. The former was isolated on column followed by preparative thin-layer chromatography and appeared as faint yellowish crystals, m.p. 95-97°C, which turned to bright yellow on standing even in nitrogenous atmosphere. The mass spectrum having M<sup>+</sup> at m/z 340 was consistent with molecular formula of C<sub>18</sub> H<sub>18</sub> O<sub>5</sub>. Further scrutiny of the UV, IR, <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra and their comparison with literature confirmed the structure as 2'-hydroxy-4,4',6'-trimethoxychalcone I.

Compound II was isolated as a bright yellow solid, m.p. 115-117°C, having identical molecular weight as the former. The UV, IR, <sup>1</sup>H NMR and <sup>13</sup>C NMR data were consistent with the reported data for 5,7,4'-trimethoxyflavanone. It is worthwhile noting that interconversion between 2'-hydroxychalcones and the corresponding flavanones in the presence of acid is a known phenomenon (Sheshadri 1962). Our observation of the conversion of 2'-hydroxy-4,4',6'-trimethoxychalcone to 5,7,4'-trimethoxyflavanone on standing may imply the possibility of the former as an artifact. This possibility was however not verified in this study.

Naturally occurring 5,7,4'-trimethoxyflavanone or naringenin trimethyl ether has been isolated from *Dahlia tenuicaulis* (syn. *D. lehmanni*) (Kaufmann and Lam 1967; Lam and Wrang 1975). Isolation of 2'-hydroxy-4,4',6'-trimethoxychalcone was, however, first reported from *Piper methysticum* (Hansel *et al.* 1963; Dutta and Som 1978). Subsequently it has also been reported to occur in *D. tenuicaulis* (Lam and Wrang 1975), *Boesenbergia pandurata* (Herunsalee *et al.* 1987), *Boesenbergia* sp. nov. (Tuntiwachwuttikul *et al.* 1987) and *Kaempferia angustifolia* (Pancharoen *et al.* 1989). Our literature search indicates that these flavonoids have not previously been isolated from the genus *Orophea*.



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