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# **Phenolic Composition and Antioxidant Activity of** *Alchemilla* Species

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**Abstract:** The genus *Alchemilla*, belonging to the Rosaceae family, is a rich source of interesting secondary metabolites, including mainly flavonoids, tannins, and phenolic acids, which display a variety of biological activities, such as anti-inflammatory, antimicrobial, and antioxidant. *Alchemilla* species are used in traditional medicine for treatment of acute diarrhea, wounds, dysmenorrhea, and menorrhagia. In this review, we focus on the phenolic compound composition and antioxidative activity of *Alchemilla* species. We can assume that phytomedicine and natural products chemistry are of significant importance due to the fact that extract combinations with various bioactive compounds possess the activity to protect the human body rather than disturb damaging factors.

Keywords: Alchemilla; Rosaceae; polyphenols; flavonoids; tannins; antioxidant activity



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# 1. Introduction

Alchemilla is commonly called "Lady's Mantle" or "lion's foot" and representatives of this genus occur mostly across Europe and Asia which include northeastern Anatolia (Turkey), north of Iraq, and northwest of Iran. More than 300 species have been described from Europe, where large mountain ranges such as the Caucasus, the Alps, the Carpathians, and others with many endemic species are probably their main distribution centers [1].

The genus *Alchemilla* L. (F. Rosaceae Juss., subfam. Rosoidae Focke) includes a large number of forms that are not easy to identify. All *Alchemilla* species were considered for many years as one species—*Alchemilla vulgaris* L., within which at most a few vaguely defined varieties were distinguished. *Alchemilla* species are apparently very similar to each other, but apart from morphological diversity, they differ greatly in ecological, phytosociological, and geographic terms.

One of the earliest mentions of the *Alchemilla* species in scientific literature is dated to 1929 and was published by Harvard University Herbaria [2]. As reported by Ergene and co-authors [3], the *Alchemilla* genus consists of approximately 1000 species. Furthermore, according to The Plant List [4], 1722 plant name records match the search criteria for *Alchemilla*.

It has been reported that in Turkish folk medicine, species belonging to the *Alchemilla* are known locally as findik out or aslan pençesi [5]. As reported by Afshar et al. [6], 24 species have been discovered within Iran and 14 among them are known to be endemic.

Aerial parts of miscellaneous *Alchemilla* species (Figure 1) are known to be excellent healing agents towards asthma, bronchitis, cough, and disorders connected with skin and liver inflammation [5].

Because of the fact that *Alchemilla vulgaris* possesses significant anti-inflammatory as well as astringent properties, it is a valued remedy for such ailments as ulcers, eczema, and menstrual disorders [7–9].



Figure 1. Alchemilla peristerica Pawł. aerial parts.

Moreover, *Alchemilla* species diminish symptoms of sore throat and alleviate nausea and vomiting [10]. *Alchemilla* species have been reported to possess a wide variety of biological activities, such as antioxidant, antibacterial, antiviral, anti-inflammatory, and ability to heal wounds in rats [11]. European Pharmacopoeia 6.0 describes *Alchemillae herba* as a medicinal agent with a variety of pharmacodynamics properties [12].

Among *Alchemilla* species that are the most widely researched for antimicrobial and antioxidant properties, *A. vulgaris*, *A. xanthochlora*, *A. diademata*, *A. rizeensis*, and *A. mollis* can be distinguished [13,14].

Various studies showed that *Alchemilla* species include miscellaneous compounds such as terpenes, hydrocarbons, fatty acids, and their esters as well as aldehydes, responsible for their pharmacological activities [15].

Other active compounds responsible for antioxidant and antimicrobial activities are tannins (composed of gallic and ellagic acid) and flavonoids (quercetin, luteolin, and proanthocyanidins) [16].

Due to the pharmaceutical and cosmetic importance of some *Alchemilla* species, in the present review, phenolic compounds occurring in the genus and antioxidant activity is discussed.

#### 2. Methodology of Evidence Acquisition

This review focuses primarily on the content of phenolic compounds as well as antioxidant activity within plants belonging to the *Alchemilla* genus. All relevant literature databases were searched up to 19 June 2022. The database Google Scholar was used for searching articles with definite search terms, namely: *Alchemilla* phenolic compounds, *Alchemilla* polyphenols, *Alchemilla* antioxidant activity. Moreover, evidence was acquired with the use of the database Pubmed. Publications were identified using the search terms *Alchemilla* (all fields) and phenolic (all fields) or antioxidant (all fields). Articles that focused on and discussed the phenolic compounds and antioxidant activity of *Alchemilla* species, dating up to June 2022, were selected.

This review was done to highlight the diversity of phenolic compounds in the *Alchemilla* genus, and thus to reveal its healing potential. Furthermore, the aim of this paper was to provide up-to-date information on antioxidant activity of *Alchemilla* species, available in the scientific literature. Finally, we decided to discuss occurring knowledge gaps and propose recommendations concerning future research directions.

To the best of our knowledge, such a review has previously not been undertaken, thus the aim of the present study was to fill this knowledge gap.

#### 3. Phenolic Compounds in the Alchemilla Species

Phenolic compounds are the most widely distributed secondary metabolites, ubiquitously present in the plant kingdom. Higher plants synthesize several thousand different phenolic compounds. The leaves contain, among others, amides and glycosides of hydroxycinnamic acids, esters, glycosylated flavonoids, and proanthocyanins and their relatives. Some soluble phenolics are widely distributed, but the distribution of many others is restricted to specific genera or families, making them easy biomarkers for taxonomic studies [17,18].

According to Choi and co-authors, plants from the genus *Alchemilla*, like other representatives of the Rosaceae family, are rich in polyphenols which are responsible for various pharmacological activities [7].

Another noteworthy observation is that various studies revealed the presence of phenolic compounds within the aerial parts of *Alchemilla* species, namely tannins, phenolic acids, flavonoids, and coumarins [6,19].

Flavonoids are an exceptionally large group of natural products (over 8000) that are found in many plant tissues, present inside the cells or on the surfaces of different plant organs [17].

Phytochemical investigations of *Alchemilla* species have led to the isolation diverse types of flavonoids, represented mostly by flavonols and flavones. Their structures are summarized in Figures 2–11 and Tables 1–6.

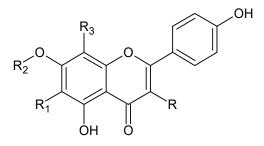


Figure 2. Kaempferol and kaempferol O-glycosides.

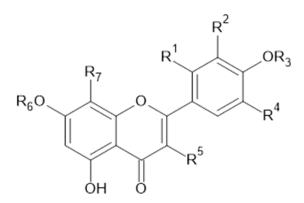


Figure 3. Quercetin and derivatives.

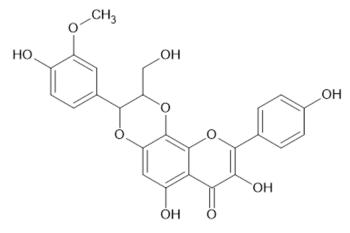


Figure 4. Rhodiolgin (36, MW 464).

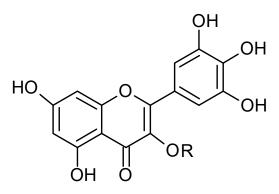


Figure 5. Chemical structure of compound 38 (R = H; MW = 318).

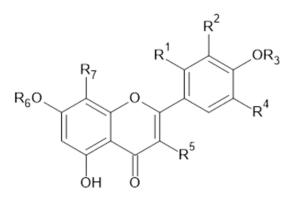


Figure 6. Luteolin and isorhamnetin, and derivatives.

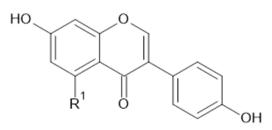


Figure 7. Chemical structure of compound 39 ( $R_1 = OH$ ; MW = 270) and 40 ( $R_1 = H$ ; MW = 254).

X

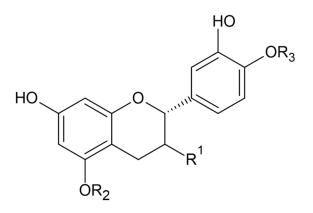


Figure 8. Catechin and epicatechin.

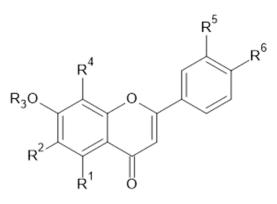


Figure 9. Apigenin and derivatives.

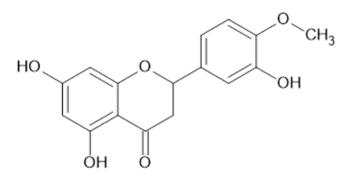


Figure 10. Chemical structure of compound 50 (MW = 302).

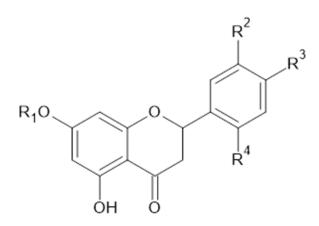


Figure 11. Chemical structure of compound 43 and 44.

Compound	R	<b>R</b> <sub>1</sub>	<b>R</b> <sub>2</sub>	R <sub>3</sub>	MW (g/mol)
1	OH	Н	Н	Н	286
2	<i>O</i> -glc	Н	Н	Н	448
3	<i>p</i> -coumaroyl-robinobioside	Н	Η	Н	918
4	xyl	Н	Η	Н	418
5	2″-O-α-L-rha-β-D-glc	Н	Η	Н	594
6	glcA	Н	Η	Н	462
7	6"-O-(E)-p-coumaroyl)glc	Н	Н	Н	594
8	<i>O</i> -rha-glc	Н	Η	Н	594
51	<i>O</i> -rha	Н	rha	Н	578
53	2- <i>p</i> -coumaroyl-glc	Н	Η	Н	594

 Table 1. Kaempferol and kaempferol O-glycosides.

Table 2. Quercetin and derivatives.

Compound	<b>R</b> <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	$R_5$	R <sub>6</sub>	<b>R</b> <sub>7</sub>	MW (g/mol)
9	Н	OH	Н	Н	OH	Н	Н	302
10	Н	Н	Н	OH	O-α-L-ara	Н	Н	434
11	Н	OH	Н	Н	$O$ - glc(6 $\rightarrow$ 1)rha	Н	Η	610
12	Н	Н	Н	OH	<i>O</i> -glucuronide	Н	Н	478
13	Н	OH	Н	Н	O- gal	Н	Н	464
14	Н	OH	Н	Н	<i>O</i> - glc	Н	Η	464
15	Н	OH	Н	Н	<i>O</i> - rha	Н	Η	448
19	Н	OH	Н	Н	$O$ - $\beta$ -D-xyl-(2→1)- $\beta$ -D-glc	Н	Н	596
20	Н	OH	Н	Н	O-β-D-xyl-(2→1)-β-D-glc	glc	Н	758
21	Н	OH	Н	Н	O- α-D-ara-furanoside	H	Н	434
54	Н	OH	Н	Н	OH	CH <sub>3</sub>	Н	316
56	Н	OH	Н	Н	O-ara-furanoside	H	Н	434

#### Table 3. Luteolin and isorhamnetin, and derivatives.

Compound	<b>R</b> <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	<b>R</b> <sub>7</sub>	MW (g/mol)
24	Н	OH	Н	Н	Н	Н	Н	286
25	Η	OH	Н	Н	Н	Н	C-β-D-glc	448
26	Η	OH	Н	Н	Н	$O$ - $\beta$ -D-glc	H	448
27	Η	OH	Н	Н	Н	rha-glc	Н	594
28	Η	OCH <sub>3</sub>	Н	Н	OH	H	Н	316
29	Η	OCH <sub>3</sub>	Н	Н	<i>O</i> -glc	Η	Н	478
30	Η	OCH <sub>3</sub>	Н	Н	<i>O</i> -rha glc	Η	Н	624
42	Н	OCH <sub>3</sub>	Н	Н	Н	Η	Н	300

Table 4. Catechin and epicatechin.

Compound	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	MW (g/mol)
34	OH	Н	OH	290
35	ОН	Н	OH	290

Table 5. Apigenin and derivatives.

Compound	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	MW (g/mol)
31	OH	Н	Н	Н	Н	OH	270
32	OH	Н	Н	glc	Н	OH	432
33	OH	Н	glc	H	Н	OH	432
57	OH	Н	H	Н	Н	OCH <sub>3</sub>	284

X

Compound	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	MW (g/mol)
43	Н	Н	OH	Н	272
44	Н	OH	OH	Н	288

Table 6. Chemical structure of compound 43 and 44.

D'Agostino et al. (1998) isolated four flavonoid glycosides, namely 3-O-kaempferol-6"-O-(*p*-coumaroyl)- $\beta$ -D-glucopyranoside (7), quercetin-3-O- $\beta$ -D-rutinoside (11), quercetin-3-O- $\beta$ -D-glucopyranoside (14), and quercetin-3-O- $\alpha$ -D-arabinofuranoside (21) from the methanol extract of *Alchemilla vulgaris* L. (Campania region, Italy) [20].

From the aqueous methanolic extract from the leaves of *A. speciosa* (Germany) astragalin (2), kaempferol 3-*O*- $\beta$ -(2"-*O*- $\alpha$ -L-rhamnopyranosyl)-glucopyranoside uronic acid (5), kaempferol 3-*O*- $\beta$ -D-glucuronide (6), 7, 11, miquelianin (12), 14, quercitrin (15), quercetin 3-*O*- $\beta$ -(2"-*O*- $\alpha$ -L-rhamnopyranosyl)-glucopyranoside uronic acid (18), hyperin (=hyperoside) (13), quercetin 3-*O*- $\beta$ -D-sambubioside (19), quercetin 3-*O*- $\beta$ - $\nu$ -sambubioside-7-*O*- $\beta$ -Dglucoside (20), cynaroside (26), and scolymoside (27) were isolated [21].

Kaya et al. used TLC and HPLC techniques to separation of **11**, **13**, **14**, **15**, orientin (**25**), and vitexin (**32**) in 50% aqueous ethanol extracts from leaves of *A. hirtipedicellata*, *A. procerrima*, *A. sericata*, and *A. stricta* collected in the northeastern Black Sea region of Turkey [22]. The same authors identified **11**, **13–15**, **25**, and **32** in the extracts from the aerial parts of *A. bursensis*, *A. cimilensis*, *A. hirsutiflora*, *A. ikizdereensis*, *A. orduensis*, and *A. oriturcica* [23].

Lamaison and co-authors isolated miquelianin (12) from the aerial parts of *A. xan-thochlora* [24]. This glucuronide of quercetin was later found also in the aerial parts of *A. barbatiflora* [25], *A. caucasica* [10], *A. achtarowii* [26], *A. mollis* [27], *A. persica* [6], *A. vul-garis* [28], *A. coriacea*, *A. filicaulis*, and *A. glabra* [29].

From the aerial flowering parts of *A. mollis* (Bulgaria), **7**, **13**, **14**, rhodiolgin (**36**, Figure 4), gossypetin-3-O- $\beta$ -D-galactopyranosyl-7-O- $\alpha$ -L-rhamnopyranoside (**37**, MW 626), and sinocrassoside D2 (**45**, MW 626) were isolated [27]. One year later, Trendafilova et al. isolated astragalin (**2**), kaempferol 3-O-(4"-*E*-*p*-coumaroyl)-robinobioside (variabiloside G, **3**), **7**, quercetin-3-O- $\alpha$ -D-arabinopyranoside (**10**), **13**, and **14** from the aerial parts of *A. achtarowii* [26].

Phytochemical studies on active fractions of the water subextract led to the isolation of kaempferol-3-O- $\beta$ -D-xylopyranoside (4), 7, 10, 12, 13, and 34 from the aerial parts of *A. barbatiflora* (Turkey) [25]. Guaijaverin (10) was also identified in the aerial parts of *A. xanthochlora* (France) [30].

Neagu et al. applied liquid chromatography coupled to tandem mass spectrometry to identify polyphenolic compounds extracted by water and aqueous ethanol (70% v/v) from *A. vulgaris*. Flavonols (kaempferol **1**, quercetin **9**, **11**, myricetin **38**, Figure 5), flavanols (**35**), flavones (luteolin **24**, Figure 6), isoflavones (genistein **39**, daidzein **40**, Figure 7), and flavonoid glucosides (**14**) were detected in the plant samples [31].

Using high performance liquid chromatography (HPLC) analysis, Denev et al. (2014) found that aerial parts of *A. glabra* (Plovdiv, Bulgaria) contained **11**, **34**, and **35** [32]. Akkol and co-authors used the same analytical method to identify **13** and **14** in the aerial parts of *A. mollis* and *A. persica* (Turkey) [33].

Duckstein et al. investigated acetone/water extracts from the leaves, including stalks, of *A. vulgaris* and *A. mollis* (Germany) for their phenolic composition by liquid chromatography-tandem mass spectrometry (LC-MS/MS). Compounds **12**, methyl-quercetin glucuronide (**16**), quercetin hexoside (**17**), quercetin-feruloyl hexose (**22**), and quercetin hexoside-deoxyhexoside (**23**) were reported [**34**].

Compounds **1**, **9**, **11**, and **34** were found in the ethanol extracts from the aerial parts of *A. vulgaris* (Russia) [35]. Karatoprak et al. reported that different (hexane, ethyl acetate, methanol, butanol) extracts from the aerial parts of *A. mollis* (Turkey) contained **11**, **26**, cosmosiin (**33**), **34**, and **35** (Figure 8) [36].

Twenty flavonoids were identified in aqueous ethanol (80% v/v) from the leaves of *A. vulgaris* (Egypt). Among them, **1**, **9**, **11**, **24**, **31** (Figure 9), **33**, **34**, **35**, naringenin (43), luteolin 6-arabinose 8-glucose (**46**, MW 610), luteolin 6-glucose 8-arabinose (**47**, MW 610), apigenin 6-arabinose 8-galactose (**48**), apigenin 6-rhamnose 8-glucose (**49**), apigenin 7-*O*-neohespiroside (**50**, MW 578; Figure 10), kaempferol 3,7-dirhamoside (**51**), hesperetin (**52**), kaempferol 3-(2-*p*-comaroyl)glucose (**53**), and rhamnetin (**54**) were determined [37].

In the aerial parts of *A. vulgaris*, **1**, **2**, **7**, **9**, **11**, **13**, **14**, **15**, **24**, **26**, **31**, **33**, **34**, **39**, morin (**41**, MW 302), chrysoeriol (**42**), and **43** were reported [38,39]. Afshar et al. used HR-MS Q-TOF for the identification of nicotiflorin (**8**), catechin (**34**), epicatechin (**35**), and aromadendrin glucoside (**55**) in the aerial parts of *A. persica* (Eastern Azarbaijan) [6]. Catechin is regarded as one of the most powerful antioxidants. Moreover, some research has shown the a strong relationship between this compound and inhibition of carcinogenesis (e.g., breast and ovarian cancer cell growth) [39].

In the latest study from 2022, Dos Santos Szewczyk et al. reported that the aerial parts and roots of *A. acutiloba* Opiz (Poland) contained **1**, **2**, **8**, **9**, **11**, **13**, **14**, **15**, **24**, isorhamnetin (**28**), isorhamnetin-3-glucoside (**29**), narcissoside (**30**), and eriodictyol (**44**, Figure 11) [40].

Recent studies have revealed the presence of such flavonoids as **2**, **7**, **11–14**, and **16** in methanol extracts from the aerial parts of *A. viridiflora* (Bulgaria) [41].

The phenolic compounds identified in the *Alchemilla* genus are summarized in Table 7.

Constituent Name	Species	Part of Plant	References
1 Karaa (m.)	A. acutiloba	aerial parts, roots	[40]
<b>1.</b> Kaempferol –	A. vulgaris	aerial parts	[31,35,37–39]
	A. acutiloba	aerial parts, roots	[40]
2. Astragalin	A. achtarowii	aerial parts	[26]
	A. speciosa	leaves	[21]
	A. viridiflora	aerial parts	[41]
	A. vulgaris	aerial parts	[38,39]
<b>3.</b> Variabiloside G	A. achtarowii	aerial parts	[26]
<b>4.</b> Kaempferol-3- <i>O</i> -β-D-xylopyranoside	A. barbatiflora	aerial parts	[25]
<b>5.</b> Kaempferol 3- <i>Ο</i> -β-(2"- <i>O</i> -α-L-rhamnopyranosyl)-glucopyranoside uronic acid	A. speciosa	leaves	[21]
<b>6.</b> Kaempferol 3- <i>O</i> -β-D-glucuronide	A. speciosa	leaves	[21]
	A. achtarowii	aerial parts	[26]
_	A. barbatiflora	aerial parts	[25]
— <b>7.</b> Kaempferol 3- <i>O</i> -β-D-(6"- <i>O</i> -( <i>E</i> )- <i>p</i> -coumaroyl)	A. mollis	aerial parts	[27]
glucopyranoside	A. speciosa	aerial parts	[21]
_	A. viridiflora	aerial parts	[41]
	A. vulgaris	aerial parts	[20,38]
8. Nicotiflorin	A. acutiloba	aerial parts, roots	[40]
_	A. persica	aerial parts	[6]

Table 7. The overview on the phenolic compounds identified in the Alchemilla genus.

Constituent Name	Species	Part of Plant	References
	A. acutiloba	aerial parts, roots	[40]
9. Quercetin	A. vulgaris	aerial parts leaves	[31,35,37–39]
	A. achtarowii	aerial parts	[26]
10. Guaijaverin	A. barbatiflora	aerial parts	[25]
	A. xanthochlora	aerial parts	[30]
	A. acutiloba	aerial parts	[40]
	A. hirtipedicellata A. procerrima A. sericata A. stricta	leaves	[22]
	A. glabra	aerial parts	[32]
11. Rutin	A. bursensis A. cimilensis A. hirsutiflora A. ikizdereensis A. orduensis A. oriturcica	aerial parts	[23]
	A. mollis	aerial parts	[27,36]
	A. speciosa	aerial parts	[21]
	A. viridiflora	aerial parts	[41]
	A. vulgaris	aerial parts	[20,31,35,37-39
	A. barbatiflora	aerial parts	[25]
	A. caucasica	aerial parts	[10]
	A. achtarowii	aerial parts	[26]
	A. mollis	aerial parts	[27]
<b>12.</b> Miquelianin	A. persica	aerial parts	[6]
	A. speciosa	aerial parts	[21]
	A. viridiflora	aerial parts	[41]
	A. vulgaris	aerial parts	[28]
	A. xanthochlora	aerial parts	[24]
	A. achtarowii	aerial parts	[26]
	A. acutiloba	aerial parts, roots	[40]
<b>13.</b> Hyperoside	A. hirtipedicellata A. procerrima A. sericata A. stricta	leaves	[22]
	A. barbatiflora	aerial parts	[25]
	A. coriacea A. filicaulis A. glabra	aerial parts	[29]

Constituent Name	Smarin-	Dout of Direct	Doference
Constituent Name	Species A. armeniaca	Part of Plant	References
	A. bursensis A. cimilensis A. hirsutiflora A. ikizdereensis A. orduensis A. oriturcica	aerial parts	[23]
-	A. mollis	aerial parts	[27,33]
-	A. persica	aerial parts	[33]
-	A. speciosa	leaves	[21]
-	A. viridiflora	aerial parts	[41]
-	A. vulgaris	aerial parts	[38]
	A. achtarowii	aerial parts	[26]
14. Isoquercitrin	A. acutiloba	aerial parts, roots	[40]
	A. hirtipedicellata A. procerrima A. sericata A. stricta	leaves	[22]
	A. bursensis A. cimilensis A. erzincanensis A. orduensis A. oriturcica	aerial parts	[23]
	A. mollis	aerial parts	[27,33]
-	A. persica	aerial parts	[33]
	A. speciosa	aerial parts	[21]
	A. viridiflora	aerial parts	[41]
	A. vulgaris	aerial parts	[20,31,38]
	A. acutiloba	aerial parts, roots	[40]
- 15. Quercitrin	A. hirtipedicellata A. procerrima A. sericata A. stricta	leaves	[22]
-	A. hirsutiflora A. orduensis	aerial parts	[23]
-	A. speciosa	aerial parts	[21]
	A. vulgaris	aerial parts	[39]
16. Methyl-quercetin glucuronide	A. mollis	leaves	[34]
	A. viridiflora	aerial parts	[41]
17. Quercetin hexoside	A. mollis	leaves	[34]
<b>18.</b> Quercetin 3- <i>Ο-β-</i> (2''- <i>Ο-</i> α-L-rhamnopyranosyl)-glucopyranoside uronic acid	A. speciosa	leaves	[21]
<b>19.</b> Quercetin 3- $O$ - $\beta$ -D-sambubioside	A. speciosa	leaves	[21]

Table 7. Cont.

Table 7. Cont.

Constituent Name	Species	Part of Plant	References
<b>20.</b> Quercetin 3- $O$ - $\beta$ - $v$ -sambubioside-7- $O$ - $\beta$ -D-glucoside	A. speciosa	leaves	[21]
<b>21.</b> Quercetin-3- <i>O</i> - <i>α</i> -D-arabinofuranoside	A. vulgaris	aerial parts	[20]
22. Quercetin-feruloyl hexose	A. vulgaris	leaves	[34]
23. Quercetin hexoside-deoxyhexoside	A. vulgaris	leaves	[34]
<b>24.</b> Luteolin	A. acutiloba	aerial parts, roots	[40]
-	A. vulgaris	aerial parts	[31,37–39]
	A. hirtipedicellata A. procerrima A. sericata A. stricta	leaves	[22]
<b>25.</b> Orientin	A. armeniaca A. cimilensis A. hirsutiflora A. ikizdereensis A. orduensis	aerial parts	[23]
	A. mollis	aerial parts	[36]
<b>26.</b> Cynaroside	A. speciosa	aerial parts	[21]
_	A. vulgaris	aerial parts	[38,39]
27. Scolymoside	A. speciosa	aerial parts	[21]
28. Isorhamnetin	A. acutiloba	aerial parts, roots	[40]
29. Isorhamnetin-3-glucoside	A. acutiloba	aerial parts	[40]
<b>30.</b> Narcissoside	A. acutiloba	aerial parts, roots	[40]
	A. caucasica	aerial parts	[10]
<b>31.</b> Apigenin	A. vulgaris	aerial parts leaves	[37,39]
	A. hirtipedicellata A. procerrima A. sericata A. stricta	leaves	[22]
<b>32.</b> Vitexin	A. armeniaca A. erzincanensis A. ikizdereensis A. orduensis	aerial parts	[23]
	A. mollis	aerial parts	[36]
<b>33.</b> Cosmosiin	A. vulgaris	aerial parts	[38,39]
	21. Unizurio _	leaves	[37]
	A. barbatiflora	aerial parts	[25]
-	A. caucasica	aerial parts	[10]
-	A. glabra	aerial parts	[32]
<b>34.</b> Catechin	A. mollis	aerial parts	[36]
-	A. persica	aerial parts	[6]
-	A. vulgaris	aerial parts leaves roots	[35,37,39]

Constituent Name	Species	Part of Plant	References
	A. glabra	aerial parts	[32]
-	A. mollis	aerial parts	[36]
<b>35.</b> Epicatechin	A. persica	aerial parts	[6]
-	A. vulgaris	aerial parts leaves	[31,37]
36. Rhodiolgin	A. mollis	aerial parts	[27]
<b>37.</b> Gossypetin-3- <i>O</i> -β-D-galactopyranosyl-7- <i>O</i> -α-L- rhamnopyranoside	A. mollis	aerial parts	[27]
<b>38.</b> Myricetin	A. vulgaris	aerial parts	[31]
<b>39.</b> Genistein	A. vulgaris	aerial parts	[31,39]
40. Daidzein	A. vulgaris	aerial parts	[31]
<b>41.</b> Morin	A. vulgaris	aerial parts	[38]
42. Chrysoeriol	A. vulgaris	aerial parts	[39]
43. Naringenin	A. vulgaris	aerial partsleaves	[37,39]
44. Eriodictyol	A. acutiloba	aerial parts	[40]
<b>45.</b> Sinocrassoside D2	A. mollis	aerial parts	[27]
46. Luteolin 6-arabinose 8-glucose	A. vulgaris	leaves	[37]
47. Luteolin 6-glucose 8-arabinose	A. vulgaris	leaves	[37]
48. Apigenin 6-arabinose 8-galactose	A. vulgaris	leaves	[37]
49. Apigenin 6-rhamnose 8-glucose	A. vulgaris	leaves	[37]
50. Apigenin 7-O-neohesperidoside	A. vulgaris	leaves	[37]
51. Kaempferol 3,7-dirhamoside	A. vulgaris	leaves	[37]
52. Hesperetin	A. vulgaris	leaves	[37]
53. Kaempferol 3-(2- <i>p</i> -comaroyl)glucose	A. vulgaris	leaves	[37]
54. Rhamnetin	A. vulgaris	leaves	[37]
55. Aromadendrin glucoside	A. persica	aerial parts	[6]
56. Avicularin	A. vulgaris	aerial parts	[35]
57. Acacetin	A. vulgaris	leaves	[37]
	A. mollis	leaves	[34]
-	A. persica	aerial parts	[6]
58. Agrimoniin	A. viridiflora	aerial parts	[41]
-	A. vulgaris	leaves	[34]
-	A. xanthochlora	aerial parts	[42]
	A. mollis	leaves	[34]
-	A. persica	aerial parts	[6]
<b>59.</b> Pedunculagin	A. vulgaris	leaves	[34]
-	A. viridiflora	aerial parts	[41]
-	A. xanthochlora	aerial parts	[42]
<b>60.</b> Laevigatin F	A. xanthochlora	aerial part	[42]

Table 7. Cont.

Constituent Name	Species	Part of Plant	References
61 Castalagin (vascalagin isomor	A. mollis	leaves	[34]
<b>61.</b> Castalagin/vescalagin isomer	A. vulgaris	leaves	[34]
	A. mollis	leaves	[34]
62. Galloyl-HHDP hexose	A. persica	aerial parts	[6]
	A. vulgaris	leaves	[34]
<b>63.</b> Trigalloyl hexose	A. mollis	leaves	[34]
	A. mollis	leaves	[34]
	A. persica	aerial parts	[6]
64. Sanguiin	A. viridiflora	aerial parts	[41]
	A. vulgaris	leaves	[34]
	A. mollis	aerial parts	[36]
<b>65.</b> Methyl gallate	A. persica	aerial parts	[6]
66. Casuarictin	A. persica	aerial parts	[6]
67. Digalloyl-galloyl galloside	A. persica	aerial parts	[6]
68. HHDP-hexoside	A. viridiflora	aerial parts	[41]
69. Brevifolincarboxylic acid	A. viridiflora	aerial parts	[41]
<b>70.</b> Tellimagrandin I	A. viridiflora	aerial parts	[41]
<b>71.</b> Tellimagrandin II	A. viridiflora	aerial parts	[41]
0	A. vulgaris	leaves	[37,43]
72. Benzoic acid	A. jumrukczalica	leaves	[43]
	A. acutiloba	aerial parts, roots	[40]
	A. glabra	aerial parts	[32]
<b>73.</b> Caffeic acid	A. jumrukczalica	leaves	[43]
701 cullet dela	A. mollis	aerial parts	[36]
	A. vulgaris	aerial parts leaves	[37,39,43]
	A. glabra	aerial parts	[32]
	A. mollis	leaves	[34]
74. Chlorogenic acid	A. persica	aerial parts	[6]
	A. vulgaris	aerial parts leaves	[31,34,37,39
75. 2,5-Dihydroxybenzoic acid	A. vulgaris	aerial parts	[39]
76. 3,4-Dihydroxybenzoic acid	A. glabra	aerial parts	[32]
	A. mollis	leaves	[34]
77. Ellagic acid	A. persica	aerial parts	[6]
77. Ellagic aciu	A. vulgaris	aerial parts leaves	[31,38,44] [31,34,37]
	A. acutiloba	aerial parts	[40]
78. Ferulic acid	A. vulgaris	aerial parts leaves	[37–39]

Table 7. Cont.

Constituent Name	Species	Part of Plant	References
	A. acutiloba	aerial parts roots	[40]
	A. glabra	aerial parts	[32]
	A. jumrukczalica	leaves	[43]
<b>79.</b> Gallic acid	A. mollis	aerial parts leaves	[34,36]
	A. persica	aerial parts	[6]
	A. vulgaris	aerial parts leaves roots	[31,34,35] [37,39,43] [45]
	A. acutiloba	aerial parts, roots	[40]
<b>80.</b> Gentisic acid	A. jumrukczalica	leaves	[43]
	A. mollis	aerial parts	[36]
	A. vulgaris	leaves	[43]
	A. acutiloba	aerial parts, roots	[40]
81. Protocatechuic acid	A. jumrukczalica	leaves	[43]
	A. vulgaris	aerial parts leaves	[37,39,43]
	A. acutiloba	aerial parts roots	[40]
<b>82.</b> <i>p</i> -Coumaric acid	A. jumrukczalica	leaves	[43]
	A. vulgaris	aerial parts leaves	[31,37,39,43]
	A. acutiloba	aerial parts roots	[40]
83. 4-Hydroxybenzoic acid	A. jumrukczalica	leaves	[43]
	A. vulgaris	aerial parts leaves	[37,39,43]
	A. jumrukczalica	leaves	[43]
84. Mandelic acid	A. vulgaris	leaves	[43]
85. 3,4,5-Methoxycinnamic acid	A. vulgaris	leaves	[37]
	A. jumrukczalica	leaves	[43]
<b>86.</b> $\beta$ -Resorcylic acid	A. vulgaris	leaves	[43]
87. Rosmarinic acid	A. acutiloba	aerial parts roots	[40]
	A. vulgaris	aerial parts leaves	[31,37]
	A. acutiloba	aerial parts roots	[40]
88. Salicylic acid	A. jumrukczalica	leaves	[43]
	A. vulgaris	leaves	[37,43]

Table 7. Cont.

Constituent Name	Species	Part of Plant	References
	A. jumrukczalica	leaves	[43]
89. Sinapic acid	A. vulgaris	aerial parts leaves	[31,43]
	A. acutiloba	aerial parts roots	[40]
90. Syringic acid	A. jumrukczalica	leaves	[43]
	A. vulgaris	leaves	[43]
	A. jumrukczalica	leaves	[43]
91. Trans-cinnamic acid	A. vulgaris	leaves	[37,43]
	A. jumrukczalica	leaves	[43]
<b>92.</b> 3,4,5-Trimethoxymandelic acid	A. vulgaris	leaves	[43]
	A. acutiloba	aerial parts roots	[40]
93. Vanillic acid	A. jumrukczalica	leaves	[43]
	A. vulgaris	leaves	[37,43]
94. Quinic acid	A. vulgaris	aerial parts	[39]

Table 7. Cont.

The plant tannins are a unique group of phenolics of relatively high molecular weight with the ability to complex strongly with carbohydrates and proteins. In higher plants, tannins consist of two major groups of metabolites: the hydrolysable tannins and condensed tannins [17]. *Alchemilla* species as members of Rosaceae family also produce, apart from flavonoids, tannins.

Geiger et al. identified in the aerial parts of *A. xanthochlora* (Germany) ellagitannins such as agrimoniin (58, MW 1871), pedunculagin (59, MW 784)), and laevigatin F (60, MW 802) [42].

Duckstein and co-authors used liquid chromatography-tandem mass spectrometry (LC-MS/MS) to identify tannins extracted by acetone/water (8/2, v/v) from the leaves (including stalks) from *A. vulgaris* and *A. mollis* (Germany). Compounds **58**, **59**, castalagin/vescalagin isomer (**61**, MW 934), galloyl-HHDP hexose (**62**, MW 618), trigalloyl hexose (**63**, MW 636), and sanguiin (**64**, MW 1871) were detected in studied plant samples [34].

In the aerial parts of *A. mollis* [36] and *A. persica* [6], methyl gallate (65, Figure 12) was identified. The authors [6] found also that extracts of *A. persica* contained 58, 59, 62, 64, casuarictin (66, Figure 13), and digalloyl-galloyl galloside (67, MW 1084).

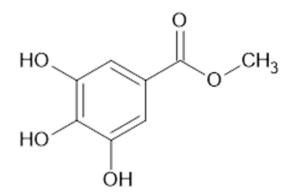


Figure 12. Chemical structure of compound 65 (MW = 184).

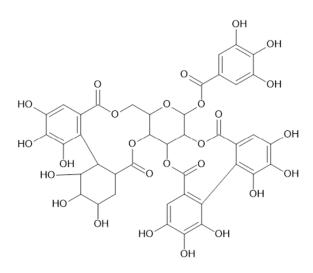


Figure 13. Chemical structure of compound 66 (MW = 936).

Recent studies have revealed the presence in the aerial parts of *A. viridiflora* such tannins as HHDP-hexoside (68), brevifolincarboxylic acid (69, Figure 14), and tellimagrandin I (70, Figure 15) and II (71, MW 938) which have been reported for the first time in *Alchemilla* species [41].

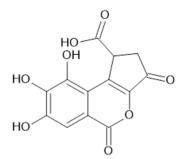


Figure 14. Chemical structure of compound 69 (MW = 292).

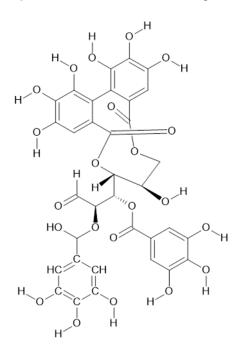


Figure 15. Chemical structure of compound 70 (MW = 786).

Phenolic acids are also common in higher plants, and they are usually present in the bound soluble form conjugated with sugars or organic acids [17] (Figures 16 and 17, Tables 8 and 9).

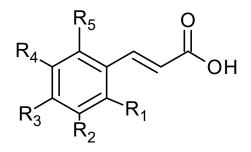


Figure 16. Hydroxycinnamic acid derivatives.

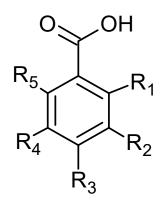


Figure 17. Benzoic acid derivatives.

Table 8. Hydroxycinnamic acid derivatives.

Compound	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	MW
73	Н	OH	OH	Н	Н	180
78	Н	OCH <sub>3</sub>	Н	OH	Н	194
82	Η	Н	OH	Н	Н	164
85	Н	OCH <sub>3</sub>	OCH <sub>3</sub>	OCH <sub>3</sub>	Н	238
89	Н	OCH <sub>3</sub>	OH	OCH <sub>3</sub>	Н	224
91	Н	Н	Н	Н	Н	148

Table 9. Benzoic acid derivatives.

Compound	R <sub>1</sub>	<b>R</b> <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	MW
72	Н	Н	Н	Н	Н	122
79	Н	OH	OH	OH	Н	170
80	OH	Н	Н	OH	Н	154
81	Н	OH	OH	Н	Н	154
83	Н	Н	OH	Н	Н	138
86	OH	Н	OH	Н	Н	154
88	OH	Н	Н	Н	Н	138
90	OH	OCH <sub>3</sub>	OH	OCH <sub>3</sub>	Н	198
93	Н	OCH <sub>3</sub>	OH	Н	Н	168

In the aqueous extracts and ethanolic extracts (70% (v/v) ethanol) of *A. vulgaris* caffeic acid (73), chlorogenic acid (74, Figure 18), elagic acid (77), ferulic acid (78), gallic acid (79), *p*-coumaric acid (82), rosmarinic acid (87, Figure 19), and sinapic acid (89) were identified using HPLC-MS analysis [31].

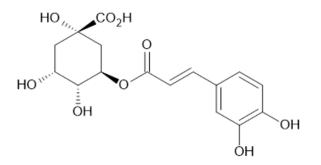


Figure 18. Chemical structure of compound 74 (MW = 354).

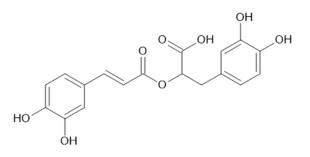


Figure 19. Chemical structure of compound 87 (MW = 360).

Denev et al. reported that the aerial parts of *A. glabra* contained **73**, **74**, **3**,**4**-dihydroxybenzoic acid (**76**), and **79** [**32**]. In the leaves, including stalks, of *A. vulgaris* and *A. mollis*, three phenolic acids (**74**, **77**, **79**) were found [**34**]. Moreover, in different extracts from the aerial parts of *A. mollis*, **73**, **79**, and gentisic acid (**80**) were noticed [**36**].

In the 80% methanol extracts from the leaves of *A. jumrukczalica* and *A. vulgaris* (Bulgaria), free and bonded phenolic acids were identified. Among reported phenolic acids, **73**, **80**, protocatechuic (**81**), salicylic (**88**), *trans*-cinnamic (**91**), and vanilic (**93**) acids were the major compounds [43].

Fifteen phenolic acids (benzoic acid (72), 73, 74, 78, 79, 81, 82, 4-hydroxybenzoic acid (83), 3,4,5-methoxycinnamic acid (85), 87, 88, 91, 93) in the leaves [37] and 77 [38], and 73, 74, 2,5-dihydroxybenzoic acid (75), 78, 79, 81–83, as well as quinic acid (94, Figure 20) [39] were found in the aerial parts of *A. vulgaris*. The advantages of identified phenolic acids are associated with several health benefits such as antioxidant, anti-diabetic, and anticancer effects [39].

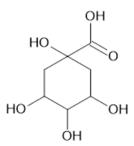


Figure 20. Chemical structure of compound 94 (MW = 192).

Moreover, Dos Santos Szewczyk et al. [40] reported that aerial parts and roots of *A. acutiloba* contained **73**, **78–83**, **87**, **88**, syringic acid (90), and **93**.

#### 4. Antioxidant Activity

It has been proven that oxidative stress participates in the formation of various diseases such as chronic obstructive pulmonary disease (COPD), Alzheimer's disease, atherosclerosis, and cancer [46].

The evidence described above confirms the importance of searching for new effective and safe antioxidant agents. As reported by Forman and Zhang [46], two major mechanisms connected with diseases formation which contribute to cellular damage can be distinguished, namely: generation of reactive oxygen species (•OH, ONOO–, HOCl) which directly oxidize macromolecules, especially membrane lipids, enzymes, proteins, as well as nucleic acids, and leads to death resulting from aberrant cell function. Furthermore, the second pathway relates with aberrant redox signaling.

Based on all the above, in this review, we sought to clarify in more detail the antioxidant potential of *Alchemilla* species (Table 10).

Species	Plant Part/Extract	Antioxidant Assay	Antioxidant Effect	References
	aerial parts, 60% methanol	DPPH	$IC_{50} = 18.69 \ \mu g/mL \ DE$	
	aerial parts, 60% methanol	ABTS	$IC_{50} = 6.17 \ \mu g/mL \ DE$	
	aerial parts, 60% methanol	CHEL	$IC_{50} = 21.60 \ \mu g/mL \ DE$	
	roots, 60% methanol	DPPH	$IC_{50} = 29.87 \ \mu g/mL \ DE$	
	roots, 60% methanol	ABTS	$IC_{50} = 14.29 \ \mu g/mL \ DE$	
	roots, 60% methanol	CHEL	$IC_{50} = 25.76 \ \mu g/mL \ DE$	
	aerial parts, butanol fraction	DPPH	$IC_{50} = 8.96 \ \mu g/mL \ DE$	
	aerial parts, butanol fraction	ABTS	$IC_{50} = 1.42 \ \mu g/mL \ DE$	
	aerial parts, butanol fraction	CHEL	$IC_{50} = 11.43 \ \mu g/mL \ DE$	
A. acutiloba	roots, butanol fraction	DPPH	$IC_{50} = 12.08 \ \mu g/mL \ DE$	[40]
	roots, butanol fraction	ABTS	$IC_{50} = 8.78 \ \mu g/mL \ DE$	
	roots, butanol fraction	CHEL	$IC_{50} = 12.33 \ \mu g/mL \ DE$	
	aerial parts, diethyl acetate fraction	DPPH	$IC_{50} = 8.83 \ \mu g/mL \ DE$	
	aerial parts, diethyl acetate fraction	ABTS	$IC_{50} = 6.54 \ \mu g/mL \ DE$	
	aerial parts, diethyl acetate fraction	CHEL	$IC_{50} = 18.89 \ \mu g/mL \ DE$	
	roots, diethyl acetate fraction	DPPH	$IC_{50} = 15.37 \ \mu g/mL \ DE$	
	roots, diethyl acetate fraction	ABTS	$IC_{50}$ =10.39 µg/mL DE	
	roots, diethyl acetate fraction	CHEL	$IC_{50} = 19.30 \ \mu g/mL \ DE$	
	aerial parts, diethyl ether fraction	DPPH	$IC_{50} = 41.46 \ \mu g/mL \ DE$	
	aerial parts, diethyl ether fraction	ABTS	$IC_{50} = 16.28 \ \mu g/mL \ DE$	

Table 10. The overview on the antioxidant activities in Alchemilla species.

Table 10. Cont.	
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Species	Plant Part/Extract	Antioxidant Assay	Antioxidant Effect	References
	aerial parts, diethyl ether fraction	CHEL	$IC_{50} = 25.51 \ \mu g/mL \ DE$	
	roots, diethyl ether fraction	DPPH	$IC_{50} = 51.42 \ \mu g/mL \ DE$	
	roots, diethyl ether fraction	ABTS	$IC_{50} = 24.82 \ \mu g/mL \ DE$	
	roots, diethyl ether fraction	CHEL	$IC_{50} = 44.12 \ \mu g/mL \ DE$	
A. alpina	aerial parts, methanol	DPPH	% Inhibition = 45.4–94.4%	[47]
	leaves, methanol	DPPH	$IC_{50} = 97.72 \ \mu g/mL$	
A. arvensis	leaves, hexane		$IC_{50} = 11.22 \ \mu g/mL$	[48]
	leaves, acetone		$IC_{50} = 4.86 \ \mu g/mL$	
	aerial parts, methanol		% Inhibition = 83.44–95.35%	
	aerial parts, hexane fraction	DPPH	% Inhibition = 18.6–59.62%	
	aerial parts, chloroform fraction		% Inhibition = 67.17–91.11%	
	aerial parts, water fraction	SOD	% Inhibition = 83.06–97.17%	
	aerial parts, methanol		% Inhibition = 83.34-85.83%	
	aerial parts, hexane fraction		% Inhibition = 9.80%	
A. barbatiflora	aerial parts, chloroform fraction		% Inhibition = 12.84–42.73%	[25]
	aerial parts, water fraction aerial parts,		% Inhibition = 81.07%	
	methanol aerial parts,	PRA	Absorbance 0.932–1.280	
	hexane fraction aerial parts,		Absorbance 0.355–0.612	
	chloroform fraction aerial parts,		Absorbance 0.640–0.820	
	water fraction aerial parts,		Absorbance 1.158–1.516	
	methanol aerial parts,	FRAP	44.32 mg BHAE/g DE	
	chloroform fraction aerial parts,		15.76 mg BHAE/g DE	
	water fraction		93.46 mg BHAE/g DE	
A. bulgarica	aerial parts, 80% methanol	DPPH	$IC_{50} = 75.63 \ \mu g/mL$	[49]
A. crinita	aerial parts, 80% methanol	DPPH	$IC_{50} = 46.03 \ \mu g/mL$	[49]
A. ellenbergiana	aerial parts, hexane	DPPH	$IC_{50} = 7.1 \ \mu g/mL$	[50]

Species

References

[55]

TEAC = 247.58 mmol TE/g DW

TEAC = 308.44 mmol TE/g DW

**Antioxidant Effect** 

#### aerial parts, $IC_{50} = 243.6 \,\mu g/mL$ DPPH ethanol [51] A. ellenbergiana aerial parts, $IC_{50} = 243.1 \ \mu g/mL$ methanol aerial parts, [50] A. erythropoda DPPH $IC_{50} = 30.67 \,\mu g/mL$ 80% methanol ORAC $IC_{50} = 1337 \ \mu mol \ TE/g$ aerial parts, A. glabra TRAP $IC_{50} = 1815 \ \mu mol \ TE/g$ [32] 80% acetone in 0.2% formic acid HORAC $IC_{50} = 1999 \ \mu mol GAE/g$ aerial parts, A. glabra DPPH $IC_{50} = 34.89 \ \mu g/mL$ [49] 80% methanol aerial parts, DPPH $IC_{50} = 36.10 \ \mu g/mL$ [49] A. glaucescens 80% methanol leaves, A. jumrukczalica DPPH $IC_{50} = 12.09 \ \mu g/mL$ [43] 80% methanol shoots grown in vitro on different $IC_{50} = 18.6 - 38.1 \,\mu g/mL$ nutrient media leaves of ex vitro adapted plants in $IC_{50} = 13.1 \ \mu g/mL$ Bulgarian mountains Vitosha DPPH [52] A. mollis one year old in vivo plantsgrown $IC_{50} = 27.5 \ \mu g/mL$ in Bulgarian mountains Viotsha one year old in vivo plantsgrown $IC_{50} = 22.2 \ \mu g/mL$ in Bulgarian mountains Rhodopes $IC_{50} = 42.4 \ \mu g/mL$ DPPH leaves, [53] A. mollis $IC_{50} = 7.8 \,\mu g/mL$ ABTS 50% ethanol aerial parts, $IC_{50} = 0.264 \text{ mg/mL}$ water DPPH aerial parts, $IC_{50} = 0.146 \text{ mg/mL}$ deodorized water aerial parts, [54] A. mollis $IC_{50} = 0.161 \text{ mg/mL}$ 50% methanol aerial parts, 0.90 mmol/L/Trolox water ABTS aerial parts, deodorized water 0.4 mmol/L/Trolox aerial parts,50% methanol 0.4 mmol/L/Trolox aerial parts, $IC_{50} = 0.21 \text{ mg/mL}$ 70% methanol DPPH $IC_{50} = 0.24 \text{ mg/mL}$ aerial parts,water aerial parts, TEAC = 0.75 mmol/Trolox A. mollis [36] 70% methanol ABTS aerial parts, water TEAC = 0.83 mmol/Trolox aerial parts, hexane, ethyl acetate, Inhibition of methanol, butanol, 70% β-carotene/linoleic no data methanol, water acid co-oxidation dry stalks, FRAP TEAC = 382.78 mmol TE/g DWaqueous ethanol dry stalks, CUPRAC TEAC = 363.79 mmol TE/g DWaqueous ethanol

DPPH

ABTS

Antioxidant Assay

Table 10. Cont.

**Plant Part/Extract** 

A. mollis

dry stalks,

aqueous ethanol dry stalks,

aqueous ethanol

## Table 10. Cont.

Species	Plant Part/Extract	Antioxidant Assay	Antioxidant Effect	Reference
	aerial parts,		$IC_{50} = 31.7 \ \mu g/mL$	
	methanol aerial parts,		u u u u u u u u u u u u u u u u u u u	
	ethyl acetate fraction		$IC_{50} = 9.8 \ \mu g/mL$	
A. mollis	aerial parts,	DPPH	$IC_{50} = > 200 \ \mu g/mL$	[27]
	petroleum fraction		$1C_{50} = 200 \ \mu g/ 11L$	
	aerial parts,		$IC_{50} = > 200 \ \mu g/mL$	
	chloroform fraction aerial parts,			
	water residue fraction		$IC_{50} = 42.5 \ \mu g/mL$	
	aerial parts,		$IC_{50} = 0.055 M$	
<u>а</u>	80% methanol	DPPH	1050 - 0.000 W	[0]
A. persica	roots, 80% methanol		$IC_{50} = 0.151 M$	[3]
	aerial parts,			
	80% methanol	TBARS	MDA = 5.9  nmol/mL	
	roots,		MDA = 19.08 nmol/mL	
	80% methanol			
A. monticola	aerial parts,	DPPH	$IC_{50} = 32.72 \ \mu g/mL$	[49]
	80% methanol aerial parts,			
A. obtusa	80% methanol	DPPH	$IC_{50} = 26.35 \ \mu g/mL$	[49]
A. sericata	aerial parts,	DPPH	$IC_{50} = 185 \ \mu g/mL$	[56]
71. <i>Ser leulu</i>	hexane	DITII	1C50 = 105 µg/ III	[00]
A. vulgaris	leaves,	DPPH	% Inhibition = 71.8%	[57]
	50% ethanol	DDDIA		
A. vulgaris	aerial parts,	DPPH ABTS	$IC_{50} = 5.40 \ \mu g/mL$	[58]
	methanol	ADIS	$IC_{50} = 60.10 \ \mu g/mL$	
	aerial parts, methanol	DPPH	$IC_{50} = 5.96 \ \mu g/mL$	
	roots,	DITT		
	methanol		$IC_{50} = 11.86 \ \mu g/mL$	
	aerial parts,		$IC_{50} = 14.80 \ \mu g/mL$	
	methanol	ABTS	-30	
	roots, methanol		$IC_{50} = 32.49 \ \mu g/mL$	
	aerial parts,	T Tanduna and and and		
	methanol	Hydroxyl radical scavenging activity	$IC_{50} = 13.06 \ \mu g/mL$	
A. vulgaris	roots,	seavenging activity	$IC_{50} = 18.44 \ \mu g/mL$	[11]
	methanol aerial parts,			
	methanol	Inhibition of lipid	$IC_{50} = 31.91 \ \mu g/mL$	
	roots,	peroxidation	$IC_{50} = 475.13 \ \mu g/mL$	
	methanol		1C50 = 475.15 µg/ IIL	
	aerial parts, methanol	Reducing power	$IC_{50} = 632.99 \text{ mg TE/g DE}$	
	roots,	Reducing power		
	methanol		$IC_{50} = 607.52 \text{ mg TE/g DE}$	
	aerial parts,	Total antioxidant	$IC_{50} = 265.62 \text{ mg AA/g DE}$	
	methanol	activity	20002 mg mm g DE	
	roots, methanol	5	$IC_{50} = 316.47 \text{ mg AA/g DE}$	
	leaves,			
A. vulgaris	80% ethanol	DPPH	% inhibition = 131.74%	[37]
A mularia	roots,	TEAC	68.21 mmol TE/g DW	
A. vulgaris	50% ethanol	FRAP	40.12  mmol TE/g DW	[59]

Species	Plant Part/Extract	Antioxidant Assay	Antioxidant Effect	References
A. vulgaris	aerial parts, cyclohexane	DPPH	$IC_{50} = 23.12 \ \mu g/mL$	[44]
		DPPH	153.30 mg TE/g DE	
		ABTS	143.55 mg TE/g DE	
	aerial parts,	CUPRAC	216.14 mg TE/g DE	
	80% methanol	PRAP	1.77  mmol TE/g DE	
		CHEL	42.58 mg EDTAE/g DE	
		FRAP	7899.45 mg AAE/g DE	
		DPPH	95.99 mg TE/g DE	
		ABTS	119.62 mg TE/g DE	
	aerial parts,	CUPRAC	203.53 mg TE/g DE	
	70% ethanol	PRAP	1.57 mmol TE/g of DE	
		CHEL	42.32 mg EDTAE/g DE	
1 milania		FRAP	6405.75 mg AAE/g DE	[20]
A. vulgaris		DPPH	502.56 mg TE/g DE	[39]
		ABTS	174.05 mg TE/g DE	
	aerial parts,	CUPRAC	283.16 mg TE/g DE	
	70% ethyl-acetate	PRAP	2.22 mmol TE/g DE	
	-	CHEL	37.96 mg EDTAE/g DE	
		FRAP	8745.31 AAE/g DE	
		DPPH	89.25 mg TE/g DE	
		ABTS	37.50 mg TE/g DE	
	aerial parts,	CUPRAC	78.56 mg TE/g DE	
	water	PRAP	0.53 mmol TE/g DE	
		CHEL	39.23 mg EDTAE/g DE	
		FRAP	3240.09 mg AAE/g DE	
	aerial parts,		$IC_{50} = 0.11 \ \mu g/mL$	
	ethanol		$1C_{50} = 0.11 \mu\text{g/mL}$	
A. vulgaris	aerial parts,		$IC_{50} = 27.22 \ \mu g/mL$	[38]
<i>А. бигдин</i> 5	water	DPPH	$1C_{50} = 27.22 \ \mu g/ \Pi L$	[30]
	aerial parts,		$IC_{50} = 2.88 \ \mu L/mL$	
	propylene glycolic		$R_{50} = 2.00 \ \mu L/ mL$	
A 1 '	aerial parts,	DDDI	87.95% (at 3 mg/mL) and 80.71%	[01]
A. vulgaris	70% ethanol	DPPH	(at 1.5 mg/mL)	[31]
	leaves,	DDDI		[ (0]
A. vulgaris	80% methanol	DPPH	$IC_{50} = 19.62 \ \mu g/mL$	[43]
A. xanthochlora	aerial parts,	DPPH	$IC_{50} = 41.78 \ \mu g/mL$	[49]
1. <i>xunthochtoru</i>	80% methanol	DITT	$1050 - 41.70 \mu g/ mL$	[יד]
	leaves, hexane		no data	
	leaves, chloroform		no data	
	leaves,	TLC-DPPH		
A. xanthochlora	ethylacetate	analysis, DPPH	no data	[13]
	leaves,	anary 515, D1 1 11		
	methanol		no data	
	leaves,			
	water		no data	

Table 10. Cont.

DPPH, 2.2-diphenyl-1-picryl-hydrazyl free radical scavenging activity; ABTS, 2,2'-azinobis [3-ethylbenzthiazoline]-6-sulfonic acid decolorization assay; CHEL, metal chelating activity; DE, dry extract; SOD, superoxide radical scavenging; PRAP, phosphomolibdenum-reducing antioxidant power; FRAP, ferric-reducing antioxidant power; BHAE, butylated hydroxyanisole equivalents; ORAC, oxygen radical absorbance capacity; TE, Trolox equivalents; TRAP, total peroxyl-radical antioxidant parameter; HORAC, hydroxyl radical averting capacity; GAE, gallic acid equivalents; CUPRAC, cupric reducing antioxidant capacity; DW, dry weight; TBARS, thiobarbituric acid reactive substances assay; MDA, malondialdehyde level; TEAC, Trolox equivalents; no data—results on the figures, without values. Ondrejovič et al. evaluated the antioxidant activities of various solvent extracts and fractions acquired by solid-liquid and liquid-liquid extraction and column chromatography from leaves of *A. xanthochlora*. The most prominent antioxidant activity was observed in methanolic extract. Isolated fraction showed antioxidant activity of 535.2 mg DPPH per g of fraction residue [13].

Antioxidant activity of the extracts from the aerial parts and roots of *A. persica* (Turkey) was evaluated using the DPPH radical scavenging assay and measurement of malondialdehyde (MDA) levels. The extracts were found to exhibit DPPH (1,1-diphenyl-2picrylhydrazyl) free radical scavenging activity with IC<sub>50</sub> values of 0.055 M and 0.151 M for the aerial parts and roots, respectively. The MDA level of the aerial parts was found to be 5.9 nmol/mL, and 19.08 nmol/mL for the roots [3].

Nikolova et al. evaluated antioxidant activity of the methanol extracts from leaves of *A. jumrukczalica* and *A. vulgaris* by the scavenging effect on DPPH radicals. The extracts showed good antiradical activity with  $IC_{50}$  values of 12.09 µg/mL and 19.62 µg/mL, respectively. Obtained values were comparable with those of butylated hydroxytoluene (BHT)—12.65 µg/mL and syringic acid—4.40 µg/mL, used as standard substances [43].

Boroja and co-authors [58] analyzed antioxidant activity of the methanol extract from aerial parts of *A. vulgaris*. They found that studied extract possessed DPPH inhibition activity with  $IC_{50} = 5.40 \ \mu g/mL$ . Moreover, the ABTS results demonstrated  $IC_{50}$  value of 60.10  $\mu g/mL$ . The same authors evaluated antioxidant efficacy of extracts from the roots and aerial parts of *A. vulgaris* (Central Serbia) as total antioxidant capacity, metal chelation and reducing power ability, inhibition of lipid peroxidation, as well as their potential to neutralize DPPH, ABTS, and OH radicals. They found that roots exert a higher total antioxidant activity than aerial parts (316.5 and 265.6 mg ascorbic acid/g, respectively). Comparable results for both extracts were obtained in the reducing power assay (633.0—aerial parts and 607.5 mg Trolox/g—roots). In the ferrous ion chelating test, all studied samples failed to chelate Fe<sup>2+</sup> at concentration 1 mg/mL [11].

Antioxidant capacity of the methanolic extract and fractions of A. mollis was also measured by their ability to scavenge the DPPH radical. The EtOAc fraction was found to be the most active radical scavenger (IC<sub>50</sub> = 9.8  $\pm$  1.8  $\mu$ g/mL) but this value was less than that of quercetin (IC<sub>50</sub> =  $3.2 \pm 0.4 \,\mu$ g/mL) [27]. The antioxidant capacity of stalks of A. mollis aqueous ethanol extracts was also investigated. The activity was determined by four different assays (FRAP (ferric-reducing antioxidant power), CUPRAC (cupric ion reducing antioxidant capacity), DPPH, and ABTS) and was expressed as mmol Trolox equivalent per dm<sup>3</sup> extract. The maximum values of extracts were  $382.78 \pm 1.16$ ;  $363.79 \pm 0.74$ ;  $247.58 \pm 2.26$ ; and  $308.44 \pm 6.74$  for FRAP, CUPRAC, DPPH, and ABTS assays, respectively [55]. Karatoprak et al. evaluated antioxidant activity of hexane, ethyl acetate, methanol, butanol, water, and 70% methanol extracts from the aerial parts of A. mollis. In the DPPH assay, the  $IC_{50}$ values were found to be 0.21 mg/mL and 0.24 mg/mL, respectively for 70% MeOH and water extracts. ABTS<sup>+•</sup> radical scavenging effects of the extracts were studied at the doses of 0.25 and 0.5 mg/mL. All the extracts showed the highest level of activity at 0.5 mg/mL. The TEAC values of 70% methanol and water extracts (0.75 and 0.83 mM/Trolox) were found higher than the methanol, ethyl acetate, and hexane extracts [36]. The authors studied antiradical activity of various extracts from the herb of A. mollis also in different research. The IC<sub>50</sub> values in the DPPH test were found 0.264 mg/mL, 0.146, and 0.161 mg/mL, respectively, for water, deodorized water, and 50% MeOH extracts. All extracts also managed to inhibit the ABTS<sup>++</sup> radical. The TEAC values of water extract (0.90 and 1.55 mM/L/Trolox) were found higher than the deodorized water and 50% MeOH extracts [54].

Denev et al. evaluated extracts of the aerial parts of *A. glabra* by means of several assays, including ORAC, TRAP, HORAC, and inhibition of lipid peroxidation. Between all extracts studied, *A. glabra* extract revealed the second highest chelating ability expressed as a HORAC value of 1999.4 µmol GAE/g [32].

Hamid and co-authors analyzed antioxidant activities of *A. vulgaris* roots using Trolox equivalent antioxidant capacity (TEAC), ferric reducing antioxidant power (FRAP), and

TBARS (thiobarbituric acid reactive substances) assays. The antioxidant activity measured with TEAC was 68.21 mmol of Trolox (TE)/g of dry weight (DW). Whereas FRAP assay was 40.12 mmol of TE/g DW [59].

Vlaisavljević et al. evaluated the various extracts (80% methanol, 70% ethanol, 70% ethylacetate, and distilled water) of the aerial parts of *A. vulgaris* by means of different in vivo assays. The authors found that ethylacetate extract demonstrated the highest antioxidant potential, where the most pronounced and significant antioxidant effect of tested extracts was observed for DPPH and FRAP assays (DPPH: 502.56 mg TE per g extract; FRAP: 8745.31 mg EAA per g of dry extract), followed by methanol extract [39].

Antioxidant effects of the extracts, fractions, and isolated compounds from the aerial parts of *A. barbatiflora* were estimated using several methods as 1,1-diphenyl-2-picryl-hydrazyl (DPPH), superoxide radical scavenging (SOD), phosphomolibdenum-reducing antioxidant power (PRAP), and ferric-reducing antioxidant power (FRAP) assays. Crude methanol extract showed remarkable DPPH and SOD radical scavenging activities with 83.44% and 83.34% at 125  $\mu$ g/mL. Among the tested sub-extracts, water sub-extract showed the best results with 83.06%, 96.08%, and 97.17% at 125, 250, and 500  $\mu$ g/mL, respectively, for DPPH scavenging activities. Hexane sub-extract had moderate DPPH scavenging activities as compared to gallic acid. In SOD assay at 125  $\mu$ g/mL, water sub-extract displayed significant SOD radical scavenging activities with 81.07%. Moreover, water sub-extract showed higher absorbance than methanol extract in PRAP assay. In FRAP assay, the result of methanol extract was found as 44.32 mg BHAE/g extract while the result of water sub-extract was as 93.46 mg BHAE/g extract [25].

The antioxidant activities of various extracts of the aerial parts and roots of *A. acutiloba* were determined using the DPPH<sup>•</sup> and ABTS<sup>•+</sup> radical scavenging assays. It was found that at a dose of 50.0  $\mu$ g/mL, the DPPH<sup>•</sup> scavenging abilities were the highest for the ethyl acetate (94.85%) and butanol (87.31%) fraction of the aerial parts, and in the ABTS<sup>•+</sup> assay for butanol fraction (80.56%). Among studied extracts, butanol fraction of the aerial parts was also the most active ones interfering with the formation of iron and ferrozine complexes (IC<sub>50</sub> = 11.43  $\mu$ g/mL of DE) [40].

#### 5. Conclusions and Research Gaps/Future Investigations

The available data suggest that recent times have brought a fundamental change in classical medicine considering the treatment of diseases. The above-mentioned phenomenon is related to large-scale application of drug combinations, thus the multidrug way of treatment is currently of great significance. Simultaneously, mono-substance therapies are becoming less and less popular. Moreover, a gradual development of drugs that activate natural defense and protective as well as repair mechanisms instead of impairing disadvantageous agents (such as cancer cells and microorganisms) can be observed [60].

Taking into consideration the above-mentioned facts, we can assume that phytomedicine and natural products chemistry are of significant importance due to the fact that extract combinations with various bioactive compounds can protect the human body rather that disturb damaging factors.

The World Health Organization (WHO) indicates that application of 74% of the curatives with plant origin using in modern medicine correlates with their traditional usages in various traditional medicines [10].

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