

**UNIVERSIDADE DE LISBOA**  
**FACULDADE DE FARMÁCIA**



**Bioactive terpenoids from *Euphorbia pubescens*: isolation  
and derivatization**

João Paulo Magalhães Carréu

Dissertation supervised by Professor Noélia Maria da Silva Dias Duarte and co-supervised by Professor Maria José Umbelino Ferreira

**Master's in Medicinal and Biopharmaceutical Chemistry**

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## Abstract

*Euphorbia* species have been used in traditional medicine to treat cancer, tumors and warts. In the last decades, considerable attention has been focused on plants from this genus as a source of biologically active compounds (e.g. antitumor, cytotoxic, antibacterial and antiviral). These plants are characterized for biosynthesizing macrocyclic and polycyclic diterpenes, as well as triterpenes, steroids and phenols. In particular, the discovery of macrocyclic jatrophone- and lathyrane-type diterpenes as a new class of potent modulators of the transmembrane protein P-gp (P-glycoprotein), the main ABC transporter involved in multidrug resistance in cancer, has fostered an increasing interest in the research of this genus.

To find out new bioactive compounds, the phytochemical study of three fractions of *Euphorbia pubescens* methanol extract was performed using chromatographic techniques, such as column chromatography or preparative thin layer chromatography.

Overall, eleven compounds were isolated, including euphobubescenol, a jatrophone-type macrocyclic diterpene, three *ent*-abietane diterpenic lactones (helioscopinolides A, B and E) five pentacyclic triterpenes (taraxerone, taraxerol, taraxerol acetate; isomultiflorenol and isomultiflorenyl acetate), one steroid ( $\beta$ -sitosterol), and one phenolic compound (coniferaldehyde). Taraxerone was isolated in major amounts. Therefore, taking advantage of the ketone function at C-3, molecular derivatization reactions were attempted, namely condensation of taraxerone with hydrazine monohydrate, phenylhydrazine and 2,4-dinitrophenylhydrazine to obtain three new hydrazone derivatives. The structures of all compounds, including stereochemical features, were deduced from their physical and spectroscopic data, which included: infrared spectroscopy, mass spectrometry (MS), and extensive one- and two- dimensional nuclear magnetic resonance studies (COSY, HMQC and HMBC), and by comparison of these data with those reported in the literature.

Further studies will be performed in order access their ability as P-gp modulators.

**Keywords:** *Euphorbia pubescens*, multidrug resistance, pentacyclic triterpenes, diterpenes.

## Resumo

As espécies do género *Euphorbia* têm sido utilizadas na medicina tradicional, desde a Antiguidade, para o tratamento de verrugas, tumores, cancro e infeções intestinais, entre outras patologias.

Nas últimas décadas, o estudo fitoquímico destas espécies tem merecido uma especial atenção, devido ao isolamento de um enorme número de compostos biologicamente ativos, dos quais se destacam os diterpenos policíclicos e macrocíclicos, com esqueletos do tipo jatrofano e latirano, triterpenos tetracíclicos e pentacíclicos e esteroides. Das diversas atividades biológicas demonstradas por estes compostos, são de salientar os resultados obtidos com os diterpenos macrocíclicos, que revelaram uma importante atividade moduladora da glicoproteína-P, uma das principais proteínas transmembranares envolvidas na multirresistência das células tumorais aos fármacos anti-neoplásicos.

Esta dissertação teve como principal objetivo a pesquisa de novos compostos bioativos a partir de *Euphorbia pubescens*. Para tal, foram utilizadas duas abordagens: i) o isolamento de compostos a partir três frações do extrato metanólico; ii) derivatização molecular de compostos isolados em maior quantidade. O estudo fitoquímico foi realizado recorrendo a técnicas cromatográficas, nomeadamente cromatografia em coluna ou cromatografia preparativa em camada fina.

Foram isolados e identificados onze compostos, incluindo cinco triterpenos pentacíclicos (taraxerona, taraxerol, acetato de taraxerol, isomultiflorenol e acetato de isomultiflorenol), um diterpeno macrocíclico do tipo jatrofano (euphpubescenol), três lactonas diterpénicas do tipo *ent*-abietano (helioscopinolidos A, B e E), um esteroide ( $\beta$ -sitosterol) e um composto fenólico (coniferaldeído).

A taraxerona foi isolada em maiores quantidades, pelo que se procedeu à derivatização da função cetona em C-3 por condensação com a hidrazina, fenil-hidrazina e a 2,4-dinitrofenilhidrazina, obtendo-se três novos derivados.

As estruturas químicas dos compostos foram deduzidas com base nas suas características físicas e espectroscópicas (IV, espectrometria de massa, ressonância magnética nuclear unidimensional -  $^1\text{H}$ -RMN e  $^{13}\text{C}$ -RMN - e bidimensional -  $^1\text{H}$ - $^1\text{H}$ -COSY, HMQC, HMBC) e por comparação com dados descritos na literatura.

Futuramente, serão realizados estudos adicionais que permitirão avaliar a capacidade dos compostos obtidos como moduladores da glicoproteína-P em células tumorais resistentes.

**Palavras-chave:** *Euphorbia pubescens*, multiresistência, triterpenos pentacíclicos, diterpenos

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## Abbreviations and Symbols

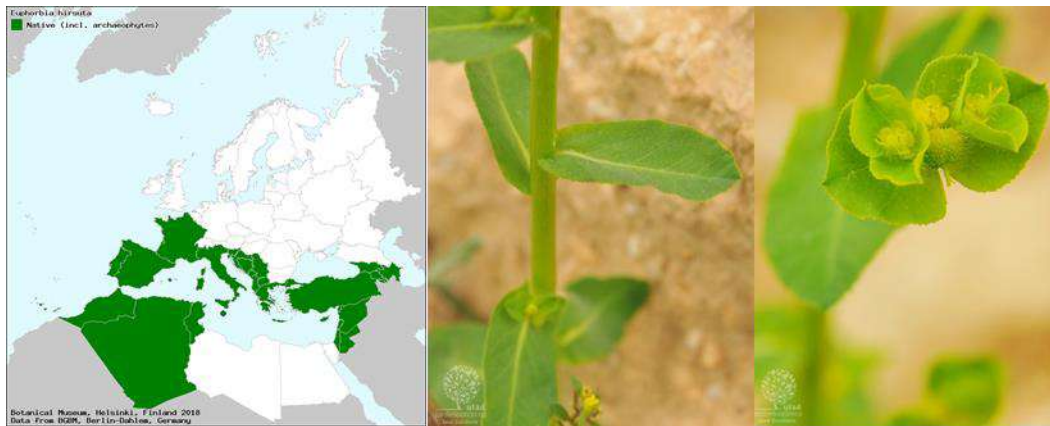
<b><sup>1</sup>H-NMR</b>	<sup>1</sup> H Nuclear Magnetic Resonance
<b><sup>13</sup>C-NMR</b>	<sup>13</sup> C Nuclear Magnetic Resonance
<b>AcAc-CoA</b>	Acetoacetyl-Coenzyme A
<b>Acetyl-CoA</b>	Acetyl-Coenzyme A
<b>ATP</b>	Adenosine Triphosphate
<b>ABC</b>	ATP Binding Cassette
<b>Bz</b>	Benzoyl
<b>δ</b>	Chemical Shift
<b>CC</b>	Column Chromatography
<b>COSY</b>	Correlation Spectroscopy
<b><i>J</i></b>	Coupling Constant
<b>d</b>	Doublet
<b>dd</b>	Doublet of Doublet
<b>ddd</b>	Doublet of Doublet of Doublet
<b>DEPT</b>	Distortionless Enhancement by Polarization Transfer
<b>DMAPP</b>	Dimethylallyl diphosphate
<b>dt</b>	Doublet of Triplets
<b>ESI-MS</b>	Electrospray Ionization Mass Spectrometry
<b>HMBC</b>	Heteronuclear Multiple Bond Correlation
<b>HMG-CoA</b>	S-3-hydroxy-3-methylglutaryl-CoA
<b>HMGR</b>	S-3-hydroxy-3-methylglutaryl reductase
<b>HMQC</b>	Heteronuclear Multiple-Quantum Correlation
<b>IR</b>	Infrared
<b>IPP</b>	Isopentenyl diphosphate
<b>ispC</b>	1-deoxy-D-xylulose-5-phosphate reductoisomerase
<b>ispD</b>	4-diphosphocytidyl-2-C-methyl-Derythritol synthase
<b>ispE</b>	4-diphosphocytidyl-2-C-methyl-Derythritol kinase
<b>ispF</b>	2-C-methyl-D-erythritol 2,4-cyclodiphosphate synthase
<b>ispG</b>	4-Hydroxy-3-methyl-but-2-enyl diphosphate synthase
<b>ispH</b>	4-Hydroxy-3-methyl-but-2-enyl diphosphate reductase
<b>m.p.</b>	Melting point

<b>MEP</b>	Methylerythritol Phosphate Pathway
<b>MVD</b>	Mevalonate diphosphate decarboxylase
<b>MK</b>	Mevalonate kinase
<b>MVA</b>	Mevalonate pathway
<b>MDR</b>	Multidrug resistance
<b>m</b>	Multiplet
<b>NMR</b>	Nuclear Magnetic Resonance
<b>P-gp</b>	P-glycoprotein
<b>ppm</b>	Parts per million
<b>PMK</b>	Phosphomevalonate kinase
<b>[M+H]<sup>+</sup></b>	Protonated Molecule Ion
<b>q</b>	Quadruplet
<b><i>m/z</i></b>	Ratio of mass to charge
<b>s</b>	Singlet
<b>t</b>	Triplet
<b>TLC</b>	Thin Layer Chromatography
<b>UV</b>	Ultraviolet

## Introduction

### 1.1 *Euphorbia pubescens*

*Euphorbia pubescens* Vahl, also known as *Euphorbia hirsuta* L., is an erect perennial herb, up to 1 meter in height and possessing yellowish green flowers (Figure 1). This plant is native to the Mediterranean Region and was introduced in Mexico, Canary Islands and South Africa.<sup>123</sup>



**Figure 1** - From left to right: Native distribution of *Euphorbia pubescens*, stem and leaves of the plant; and flowers and leaves of the plant.<sup>23</sup>

The Euphorbiaceae is one of the largest and more diverse families amongst angiosperms, ranging from large trees to small weeds, distributed all over the world. Consensus on the number of species in this family has not yet been achieved, with some authors reporting that it has more than 2000 species and others claiming that it has more than 8000 species in 300 genera (1,2).

*Euphorbia* species have been used as medicinal plants since ancient times for the treatment of various ailments, such as tumors and warts (3). For example, in Nigeria, *Euphorbia hirta* is used by the major tribes for the treatment of asthma, with the plant being squeezed into water and then the extract is drunk by the patient (4). *Euphorbia kansui* has been traditionally used in the Far East for the treatment of edema, asthma and various other diseases, and is now under studies for the prevention and treatment of psoriasis(5).

Another example of the therapeutical potential of this genus is *Euphorbia nerifolia*, which is used in traditional medicine as an effective treatment for rheumatism, sciatica, gout

<sup>1</sup> <http://ww2.bgbm.org/EuroPlusMed/PTaxonDetail.asp?NameCache=Euphorbia%20hirsuta>

<sup>2</sup> <http://www.worldfloraonline.org/taxon/wfo-0000962564>

<sup>3</sup> [https://jb.utad.pt/especie/Euphorbia\\_hirsuta](https://jb.utad.pt/especie/Euphorbia_hirsuta)

and bronchitis. It is under several studies, in which anti-inflammatory, analgesic, immunomodulation, hepato-protection, antioxidant, diuretic, and anti-viral/tumoral/bacterial activities have shown the most promising results (6). A biopolymeric fraction named BET, isolated from *Euphorbia tirucalli*, has also been shown to be an auspicious therapeutic agent in the treatment of arthritis through the modulation of the host's immune response (7).

More importantly, there is the example of Picato®, an ingenol mebutate (Figure 2) gel for the treatment of actinic keratosis, a skin lesion usually caused by sun exposure that can, sometimes, evolve into squamous cell carcinoma. In comparison to the market options at the time, this gel offered a treatment with shorter duration of use with a favorable side-effect profile in which the lesions healed faster. Ingenol mebutate is a diterpenic compound found in *Euphorbia peplus*, treating the lesions of this condition by inducing cell death and is also able of inhibiting tumor cell growth.

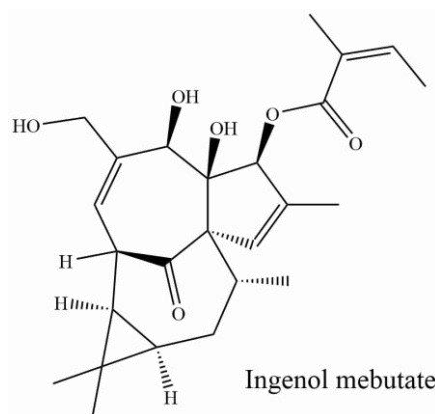


Figure 2 - Structure of ingenol mebutate (Picato®).

## 1.2 Main constituents of *Euphorbia* species: triterpenes and diterpenes

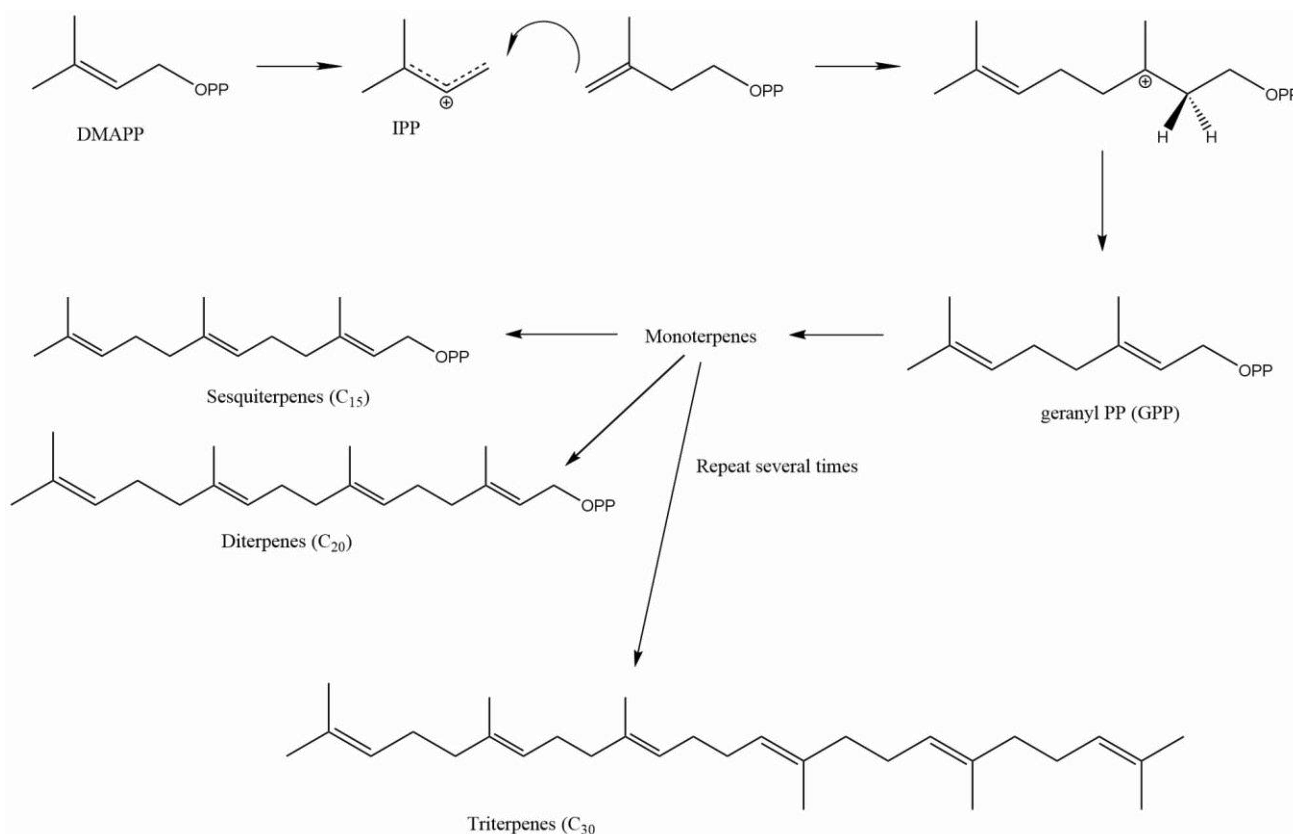
Triterpenes and diterpenes are major constituents found in *Euphorbia* genus. Terpenoids, also known as terpenes or isoprenoids, are a large group of structurally diverse natural products that comprises at least 40.000 compounds. The terpenic skeletons are derived from two isoprene units ( $C_5$  building blocks, Scheme 1), the isopentenyl diphosphate (IPP) and its isomer dimethylallyl diphosphate (DMAPP). As shown in Scheme 1, depending on the number of isoprene units ( $C_5$  building blocks) they can be classified as monoterpenes ( $C_{10}$ ), sesquiterpenes ( $C_{15}$ ), diterpenes ( $C_{20}$ ), sesterterpenes ( $C_{25}$ ), triterpenes ( $C_{30}$ ) and tetraterpenes ( $C_{40}$ )(8,9).

Terpenoids have fundamental functions, such as growth, development, reproduction and defense; as well as a more specialized role in the plants, originating from immense divergent



gene evolution, such as stress response, pollinator attraction or even repellent activity. In human society, this specialization has been profitable, as terpenoids are used as flavors, fragrances, insecticides, pharmaceuticals and as raw industry materials for adhesives or coatings.

The isoprene units (IPP and DMAPP) are biosynthesized in plants by two different pathways: the mevalonate (MVA) pathway and the methylerythritol phosphate (MEP) pathway (Scheme 2)(8).

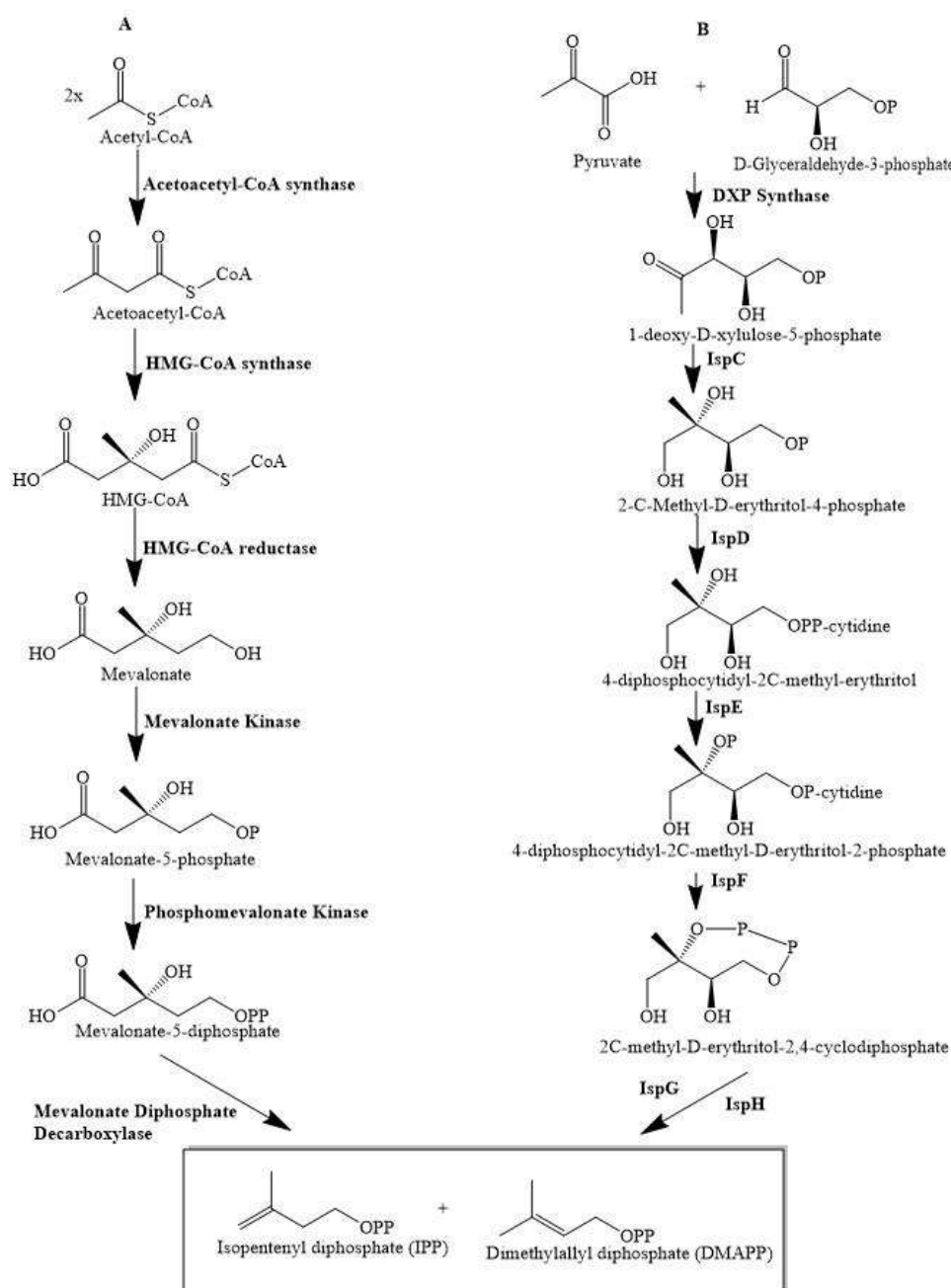


*Scheme 1 - Formation of terpenes from DMAPP and IPP.*

The mevalonate pathway (MVA, Scheme 2) starts with Claisen condensation of two molecules of acetyl-CoA to yield acetoacetyl-CoA (AcAc-CoA) that further reacts with another molecule of acetyl-CoA, forming S-3-hydroxy-3-methylglutaryl-CoA (HMG-CoA). S-HMG-CoA is then converted to R-mevalonate through the catalyst HMG-CoA reductase (HMGR), and sequentially being converted into IPP and DMAPP through phosphorylation catalyzed by mevalonate kinase (MK) and phosphomevalonate kinase (PMK) and an ATP-driven decarboxylative elimination catalyzed by mevalonate diphosphate decarboxylase (MVD)(8–10).

The methylerythritol phosphate (MEP, Scheme 2) pathway, also known as mevalonate-independent pathway and deoxyxylulose phosphate pathway, was one of the greatest

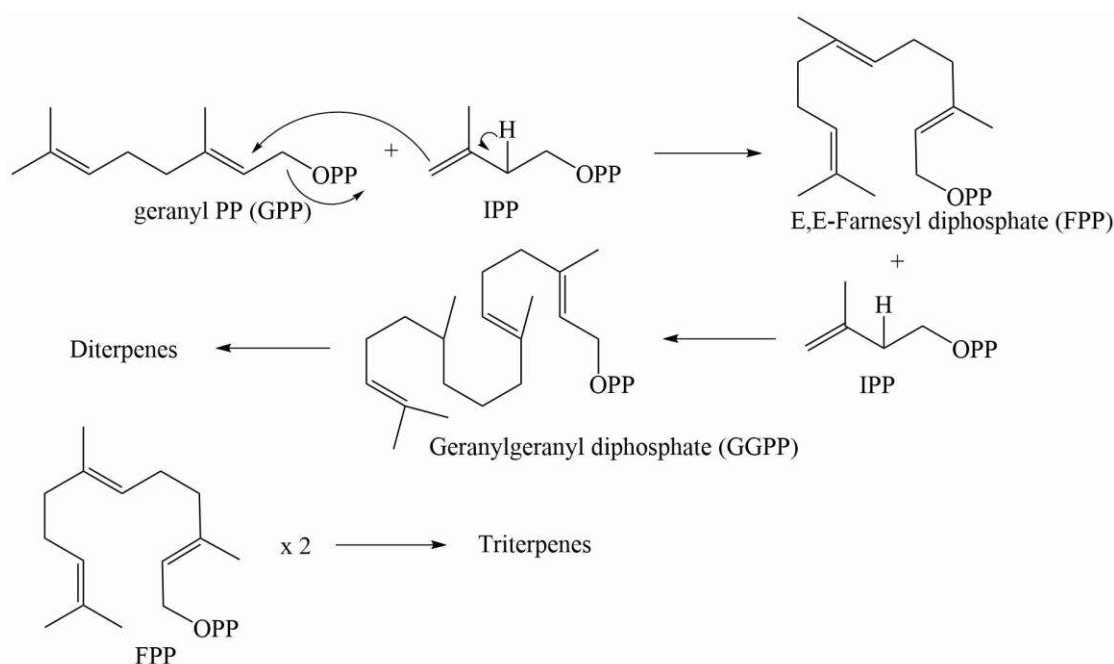
discoveries regarding plant terpenoid biosynthesis (9,11). This pathway begins with a condensation of pyruvate and D-Glyceraldehyde-3-phosphate (GAP), catalyzed by DXP synthase, forming 1-deoxy-D-xylulose-5-phosphate (DXP), which is then converted into 2-C-methyl-D-erythritol-4-phosphate (MEP) by IspC. MEP is then converted into 4-diphosphocytidyl-2C-methyl-erythritol (CDP-ME) by the enzyme IspD, and CDP-ME suffers phosphorylation by IspE, leading to the formation of 4-diphosphocytidyl-2C-methyl-D-erythritol-2-phosphate (CDP-ME2P). CDP-ME2P is cyclized by IspF into 2C-methyl-D-erythritol-2,4-cyclodiphosphate (MEcPP) which then suffers a two-electron reduction by IspG and is converted into a mixture of IPP and DMAPP by IspH. (8,12).



*Scheme 2 - Terpenoid pathways, A being the MVA pathway and B being the MEP pathway (adapted from (5,6)).*

Isopentenyl diphosphate (IPP) is transformed into dimethylallyl diphosphate (DMAPP) by the enzyme isopentenyl-isomerase (IPP isomerase) (Scheme 1). A molecule of DMAPP is condensed with a molecule of IPP to obtain geranyl diphosphate (GPP) and the addition of another molecule of IPP originates farnesyl diphosphate (FPP). If one more IPP is added, geranylgeranyl diphosphate (GGPP) is obtained, while a sequential addition leads to geranyl farnesyl diphosphate (GFPP). GPP is the precursor of monoterpenes (C<sub>10</sub>), FPP of sesquiterpenes (C<sub>15</sub>), GGPP of diterpenes (C<sub>20</sub>) and GFPP of sesterterpenes (C<sub>25</sub>).

A reductive tail-to-tail dimerization of two molecules of FPP, catalyzed by a membrane-bound enzyme, leads to squalene, which is the precursor of triterpenes and steroids. All-*trans*-squalene is the immediate precursor of all cyclic triterpenoids.



*Scheme 3 - General scheme of the formation of diterpenes and triterpenes.*

In this section, a literature review of the new triterpenes and diterpenes isolated from *Euphorbia* sp. between 2015 and 2019 are presented, as well as their respective biological activities, focusing on cytotoxic and antiproliferative activities (Tables 1 – 2 and Figure 3). This timeframe was chosen due to the fact that a previous work about these compounds and this *Euphorbia* species already presented a literature review until 2015 (13). This review was conducted on Google Scholar and PubMed, using the keywords “*Euphorbia* terpenes”, “*Euphorbia* diterpenes”, “*Euphorbia* triterpenes” and “*Euphorbia pubescens*”, with the year filter from 2015 to 2019.

**Table 1** - Triterpenes isolated from *Euphorbia* sp. and their respective activity

<b><i>Euphorbia</i> Species</b>	<b>Triterpene (Structure number)</b>	<b>Biological Activity</b>	<b>Reference</b>
<i>Euphorbia helioscopia</i>	Euphorbatrine A ( <b>1</b> )	Compounds <b>1-7</b> increase cell death in HeLa and HT-29 cell lines.	(14)
	Euphorbatrine B ( <b>2</b> )		
	Euphorbatrine C ( <b>3</b> )		
	Euphorbatrine D ( <b>4</b> )		
	Euphorbatrine E ( <b>5</b> )		
	Euphorbatrine F ( <b>6</b> )		
	Euphorbatrine G ( <b>7</b> )		
<i>Euphorbia pedroi</i>	Spiropedroxodiol ( <b>8</b> )	Compound <b>8</b> demonstrated to be a very strong MDR reversal agent in both L5178Y-MDR and Colo320 cell lines.	(15)
<i>Euphorbia schimperi</i> C. Presl	Cycloschimperols A ( <b>9</b> )	Compounds <b>9</b> and <b>10</b> showed significant cytotoxic activity against MCF-7, HepG2, and HCT-116 cells lines.	(16)
	Cycloschimperols B ( <b>10</b> )		
<i>Euphorbia pterococca</i> Brot.	Cycloartenyl-2' <i>E</i> ,4' <i>E</i> -decadienoate ( <b>11</b> )	Compound <b>11</b> showed exhibited moderate but significant inhibitory effects on $\alpha/\beta$ -hydrolase 12.	(17)
	Cycloartenyl-2' <i>E</i> ,4' <i>Z</i> -decadienoate ( <b>12</b> )		
	24-methylenecycloartanyl-2' <i>E</i> ,4' <i>Z</i> -tetradecadienoate ( <b>13</b> )		
	24-oxo-29-norcycloartanyl-2' <i>E</i> ,4' <i>Z</i> -hexadecadienoate ( <b>14</b> )		
<i>Euphorbia alata</i> Boiss	Alatavolide ( <b>15</b> )	Compounds <b>15</b> and <b>16</b> showed weak and moderate cytotoxicity, respectively, against HeLa, MCF-7 and A549 cell lines.	(18)
	Alatavoic acid ( <b>16</b> )		

**Table 1** - Triterpenes isolated from *Euphorbia* sp. and their respective activity (cont.)

<i>Euphorbia boissierana</i> Prokh.	3-[(O-β-D-fucopyranosyl-1-(1→2)-β-D-glucopyranosyl)oxy]lup-20(29)-en-3β,28-diol ( <b>17</b> )		(19)
<i>Euphorbia resinifera</i> Berg	Euphorol A ( <b>18</b> )	Compounds <b>18</b> to <b>26</b> exhibited various degrees of cytotoxic effect against MCF-7, U937 and C6 cancer cell lines.	(20)
	Euphorol B ( <b>19</b> )		
	Euphorol C ( <b>20</b> )		
	Euphorol D ( <b>21</b> )		
	Euphorol E ( <b>22</b> )		
	Euphorol F ( <b>23</b> )		
	Euphorol G ( <b>24</b> )		
	Euphorol H ( <b>25</b> )		
	Euphorol I ( <b>26</b> )		
	Euphorol K ( <b>27</b> )		
Euphorol J ( <b>28</b> )			
<i>Euphorbia tirucalli</i> L.	Euphorol L ( <b>29</b> )		(22)
	Euphorol M ( <b>30</b> )		
	Euphorol N ( <b>31</b> )		

**Table 2** - Diterpenes isolated from *Euphorbia* sp. and their respective activity

<i>Euphorbia</i> Species	Diterpenes (Structure number)	Biological Activity	Reference
<i>Euphorbia stracheyi</i> Boiss	<i>nt</i> -Kaurane-16β,17,19-triol-3-one ( <b>32</b> )		(23)
<i>Euphorbia neriifolia</i>	Euphorantin S ( <b>33</b> )	Compound <b>38</b> demonstrated moderate anti-HIV-1 activity.	(24)
	Euphorantin T ( <b>34</b> )		
	Euphorneroid A ( <b>35</b> )		
	Euphorneroid B ( <b>36</b> )		

**Table 2** - Diterpenes isolated from *Euphorbia* sp. and their respective activity (cont.)

	Euphorneroid C (37)		
	Euphorneroid D (38)		
<i>Euphorbia lathyris</i>	<i>Euphorbia</i> factor L <sub>27</sub> (39)	Compound <b>40</b> showed antiproliferative activity against MCF-7, MDA-MB-468, 786-0 and HepG2 human cancer cell lines.	(25)
	<i>Euphorbia</i> factor L <sub>28</sub> (40)		
<i>Euphorbia laurifolia</i>	15 $\beta$ -Acetoxy-7 $\beta$ -nicotinoyloxy-3 $\beta$ ,8 $\alpha$ -di-(2-methylpropanoyloxy)-4 $\alpha$ H,9 $\alpha$ H,11 $\alpha$ H-lathyra-5 <i>E</i> ,12 <i>E</i> -dien-14-one (41)		(26)
	2- <i>Epi</i> -latazienone (42)		
<i>Euphorbia ebracteolata</i>	Euphoroid A (43)	Compounds <b>44</b> and <b>45</b> demonstrated cytotoxic activity against human cancer cell lines A549, MCF-7, Mewo, SH-SY5Y, Lovo, and HepG2.	(27)
	Euphoroid B (44)		
	Euphoroid C (45)		
<i>Euphorbia ebracteolata</i>	Ebractenoid O (46)	Compounds <b>46-48</b> exhibited anti-inflammatory activities, by inhibition of LPS-induced NO production <i>in vitro</i> in macrophages.	(28)
	Ebractenoid P (47)		
	Ebractenoid Q (48)		
<i>Euphorbia clementei</i> Boiss	4,20-dideoxy-4 $\alpha$ -phorbol-12 $\beta$ -acetate-13 $\alpha$ -isobutyrate (49)	Compound <b>49</b> exhibited a weak selective cytotoxic activity against HL60 cell line.	(29)
<i>Euphorbia fischeriana</i> Steud	Euphorin E (50)	Compounds <b>50-58</b> inhibited mammosphere (a clump of mammary gland cells) formation in MCF-7 cells.	(30)
	Euphorin H (51)		
	Yuexiandajisu E (52)		
	<i>ent</i> -3 $\beta$ -hydroxyatis-16-ene-2,14-dione (53)		
	19-O- $\beta$ -D-glucopyranosyl- <i>ent</i> -atis-16-ene-3,14-dione (54)		
	Euphorin C (55)		

*Table 2 - Diterpenes isolated from Euphorbia sp. and their respective activity (cont.)*

	Ebractenoid C ( <b>56</b> )		
	Ebractenoid F ( <b>57</b> )		
	Jolkinol A ( <b>58</b> )		
<i>Euphorbia kansui</i>	kansuinin P ( <b>59</b> )		(31)
	kansuinin Q ( <b>60</b> )		
<i>Euphorbia yinshanica</i>	<i>Ent</i> -3 $\alpha$ ,16-dihydroxyabda-8(17),12(E),14-triene ( <b>61</b> )		(32)
	<i>Ent</i> -14(S),15-dihydroxyabda-8(17)-12(E)-dien-18-oic acid ( <b>62</b> )		
<i>Euphorbia pithyusa</i>	3 $\beta$ ,7 $\beta$ ,13 $\beta$ ,17-O-Tetraacetyl-5 $\alpha$ -O-benzoyl-14-oxopremyrsinol ( <b>63</b> )	Compound <b>68</b> exhibited good inhibitory activity against CHIKV replication on Vero cells, while compounds <b>63-67</b> and <b>69</b> exhibited weak inhibitory activity.	(33)
	3 $\beta$ ,7 $\beta$ ,15 $\beta$ ,17-O-Tetraacetyl-5 $\alpha$ -O-benzoyl-14-oxopremyrsinol ( <b>64</b> )		
	3 $\beta$ ,7 $\beta$ ,13 $\beta$ ,17-O-Tetraacetyl-5 $\alpha$ -O-(2-methylbutyryl)-14-oxopremyrsinol ( <b>65</b> )		
	7 $\beta$ ,13 $\beta$ ,17-O-Triacetyl-5 $\alpha$ -O-(2-methylbutyryl)-3 $\beta$ -O-propanoyl-14-oxopremyrsinol ( <b>66</b> )		
	7 $\beta$ ,17-O-Diacetyl-5 $\alpha$ -O-benzoyl-13 $\beta$ -nicotinyl-		

*Table 2 - Diterpenes isolated from Euphorbia sp. and their respective activity (cont.)*

	3 $\beta$ -O-propanoyl-14-oxopremyrsinol ( <b>67</b> )		
	13 $\beta$ ,17-O-Diacetyl-5 $\alpha$ -O-benzoyl-7 $\beta$ -hydroxy-3 $\beta$ -O-propanoyl-14-oxopremyrsinol ( <b>68</b> )		
	5 $\alpha$ ,7 $\beta$ -O-Diacetyl-14 $\beta$ -O-benzoyl-3 $\beta$ -O-propanoylmyrsinol ( <b>69</b> )		
<i>Euphorbia welwitschii</i>	Euphowelwitschines A ( <b>70</b> )	Compounds <b>72</b> and <b>73</b> showed antiproliferative selectivity toward the resistant gastric cell line EPG85-257RDB, with <b>73</b> having an MDR-selective antiproliferative effect on pancreatic (EPP-181) human cell lines.	(34)
	Euphowelwitschines B ( <b>71</b> )		
	Welwitschene ( <b>72</b> )		
	Epoxywelwitschene ( <b>73</b> )		
<i>Euphorbia resinifera</i> Berg	Euphoresin A ( <b>74</b> )	Compounds <b>74</b> and <b>75</b> exhibited small cytotoxic activities against MCF-7, U937, and C6 cancer cell lines.	(35)
	Euphoresin B ( <b>75</b> )		
<i>Euphorbia thymifolia</i>	(2 <i>R</i> , 3 <i>S</i> , 5 <i>S</i> , 9 <i>S</i> , 10 <i>R</i> )-2,3-dihydroxy- <i>ent</i> -abieta-8(14),12(13)-dien-7-one ( <b>76</b> )		(36)
<i>Euphorbia dracunculoides</i> Lam	4-deoxy-4( $\beta$ )H-8-hydroperoxyphorbol-12-benzoate-13-isobutyrate ( <b>77</b> )		(37)
<i>Euphorbia guyoniana</i>	Guyonianin G ( <b>78</b> )	Compounds <b>78</b> and <b>79</b> showed strong activity as inhibitors of GIRK current, showing promise in the development of antiarrhythmic drugs.	(38)
	Guyonianin H ( <b>79</b> )		



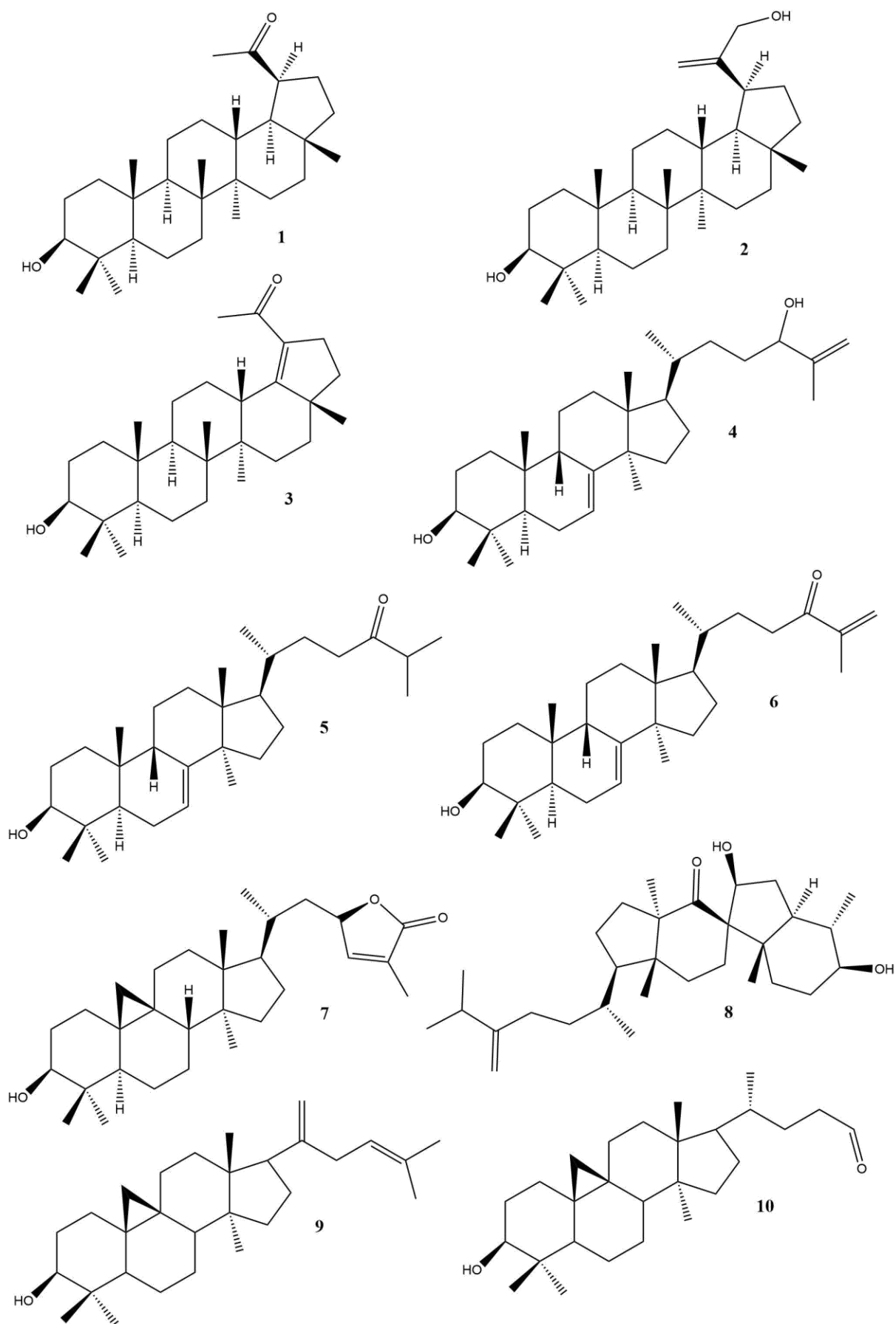


Figure 3 - Structures of compounds from Tables 1 and 2.

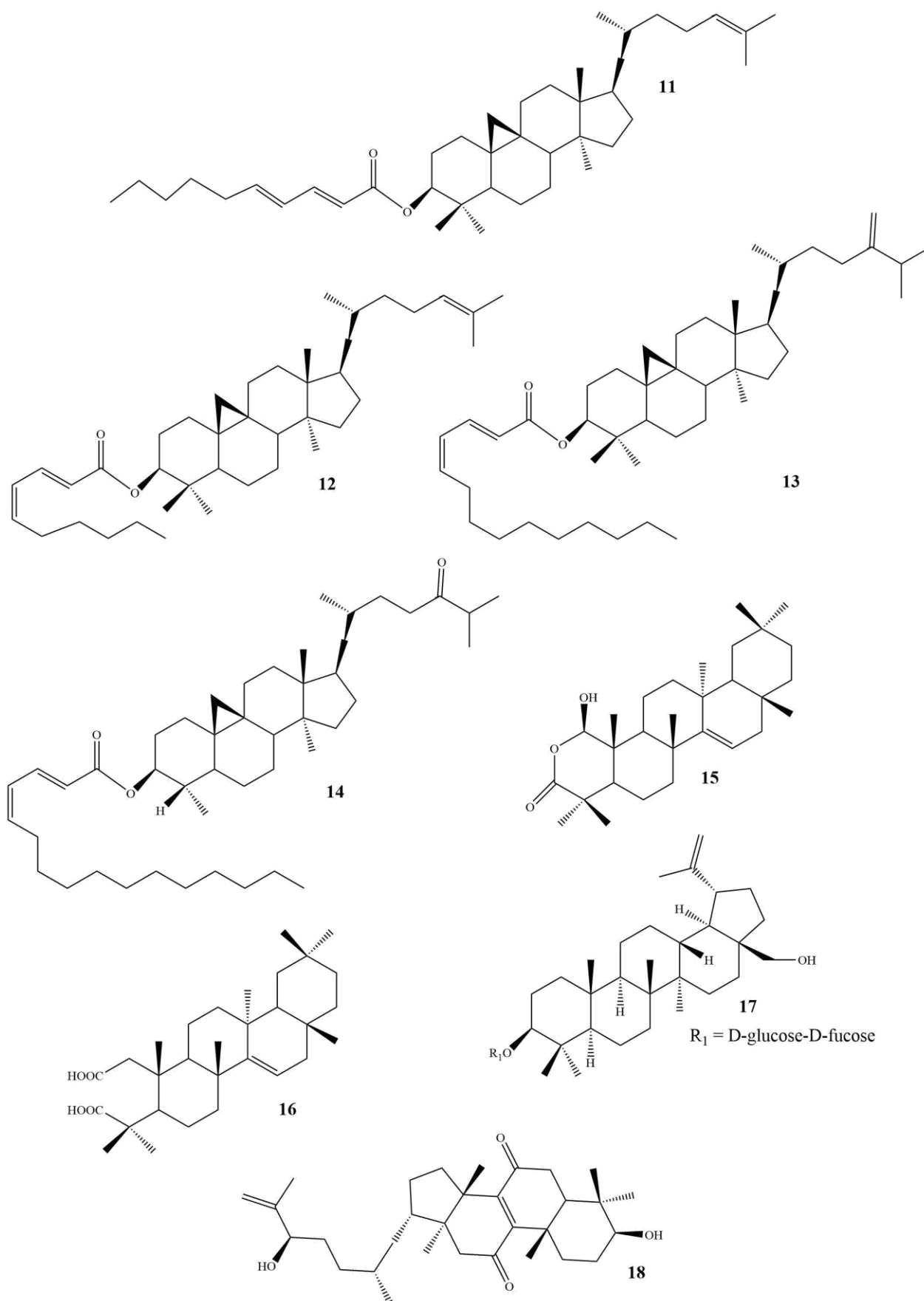


Figure 3 - Structures of compounds from Tables 1 and 2 (cont.)

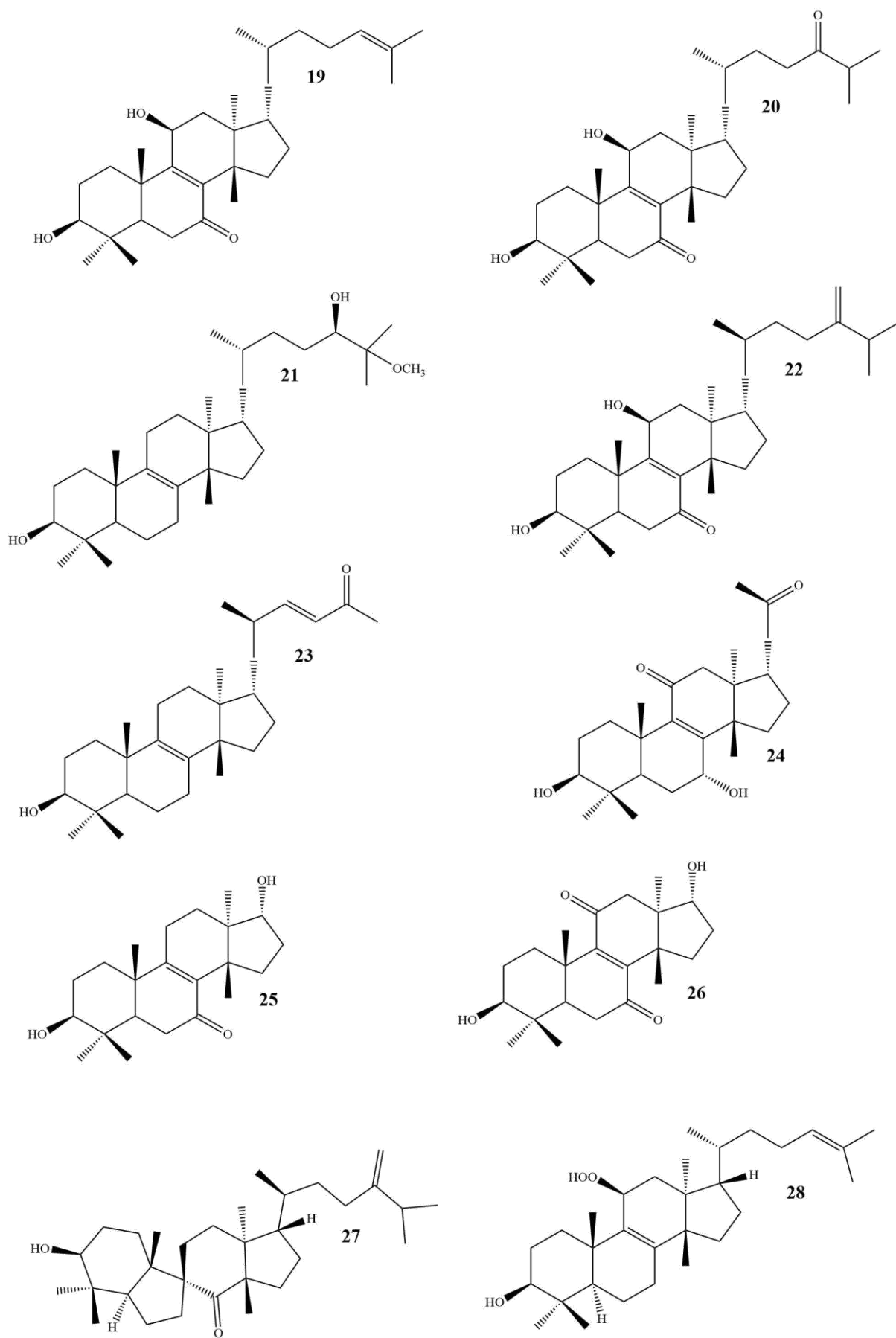


Figure 3 - Structures of compounds from Tables 1 and 2 (cont.)

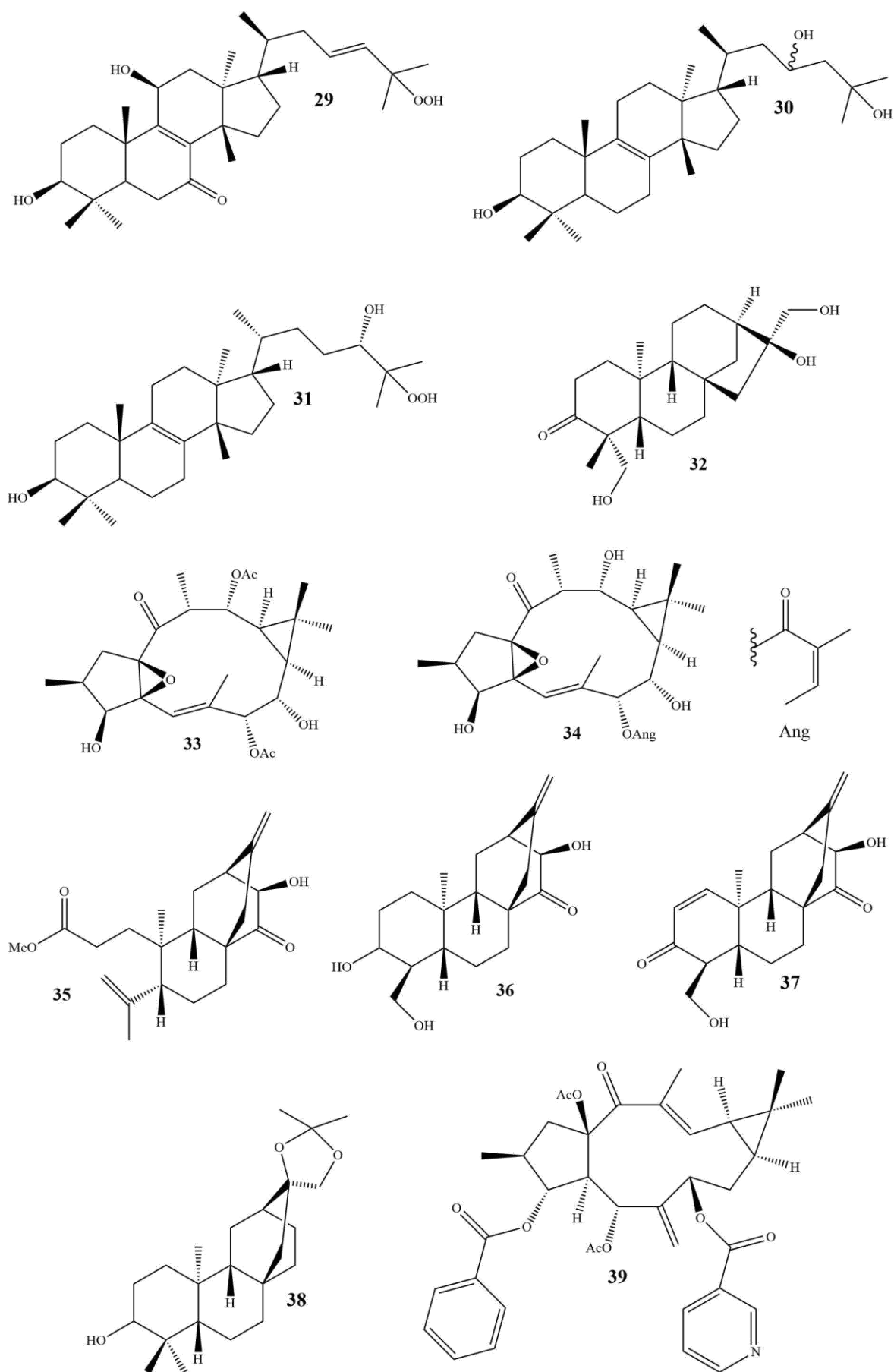


Figure 3 - Structures of compounds from Tables 1 and 2 (cont.)

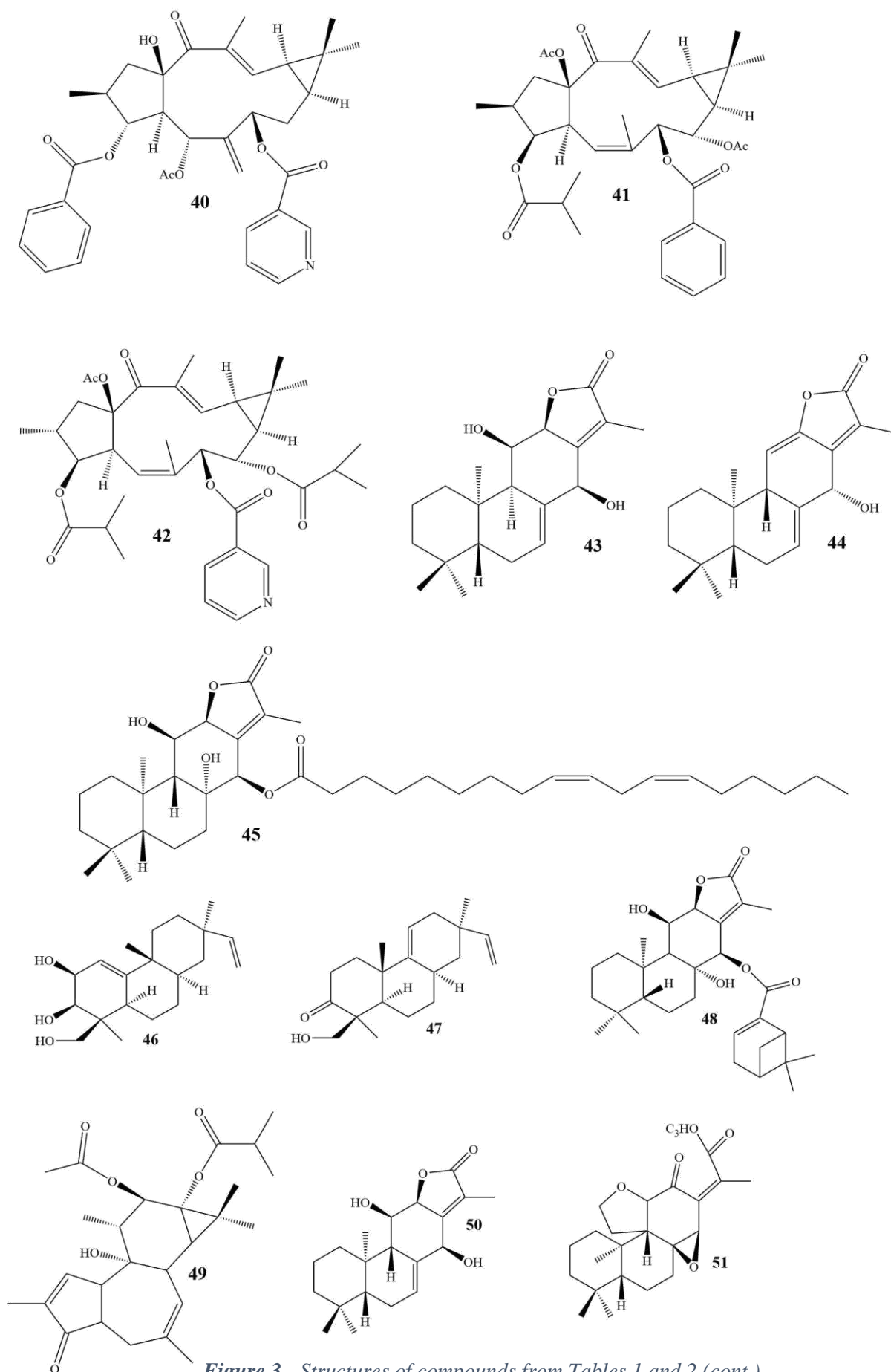


Figure 3 - Structures of compounds from Tables 1 and 2 (cont.)

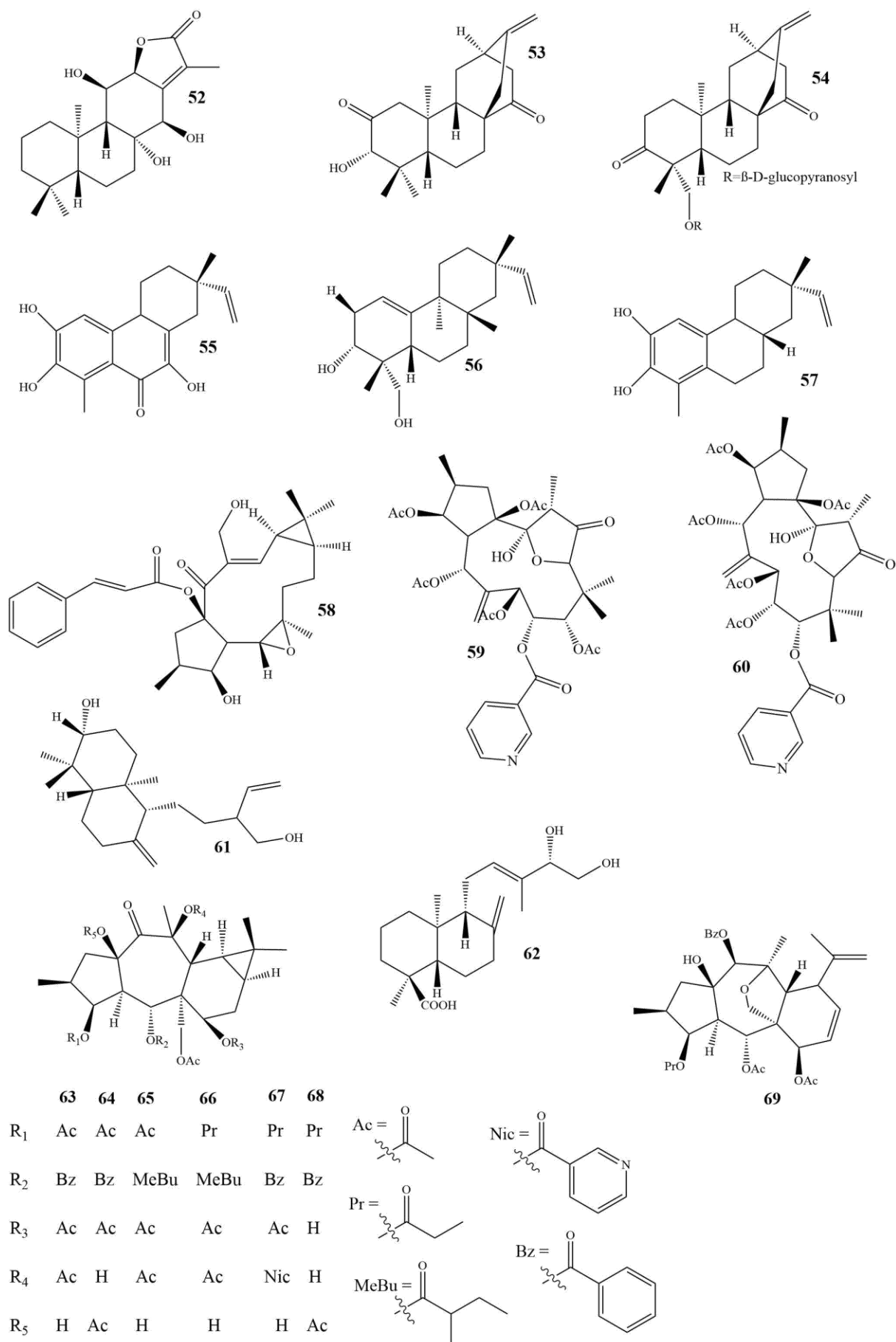


Figure 3 - Structures of compounds from Tables 1 and 2 (cont.)

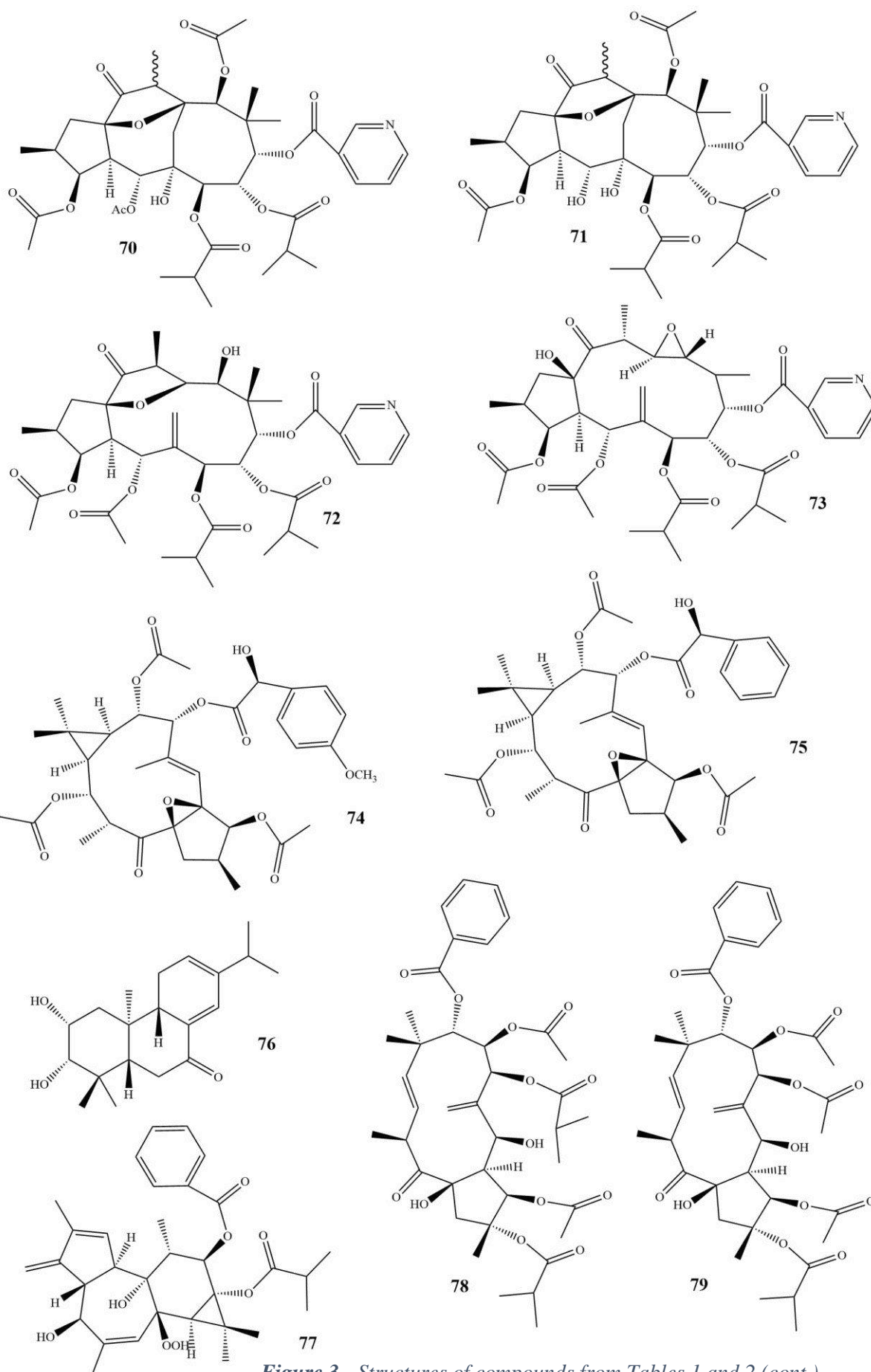


Figure 3 - Structures of compounds from Tables 1 and 2 (cont.)

### 1.3 Multidrug resistance in cancer

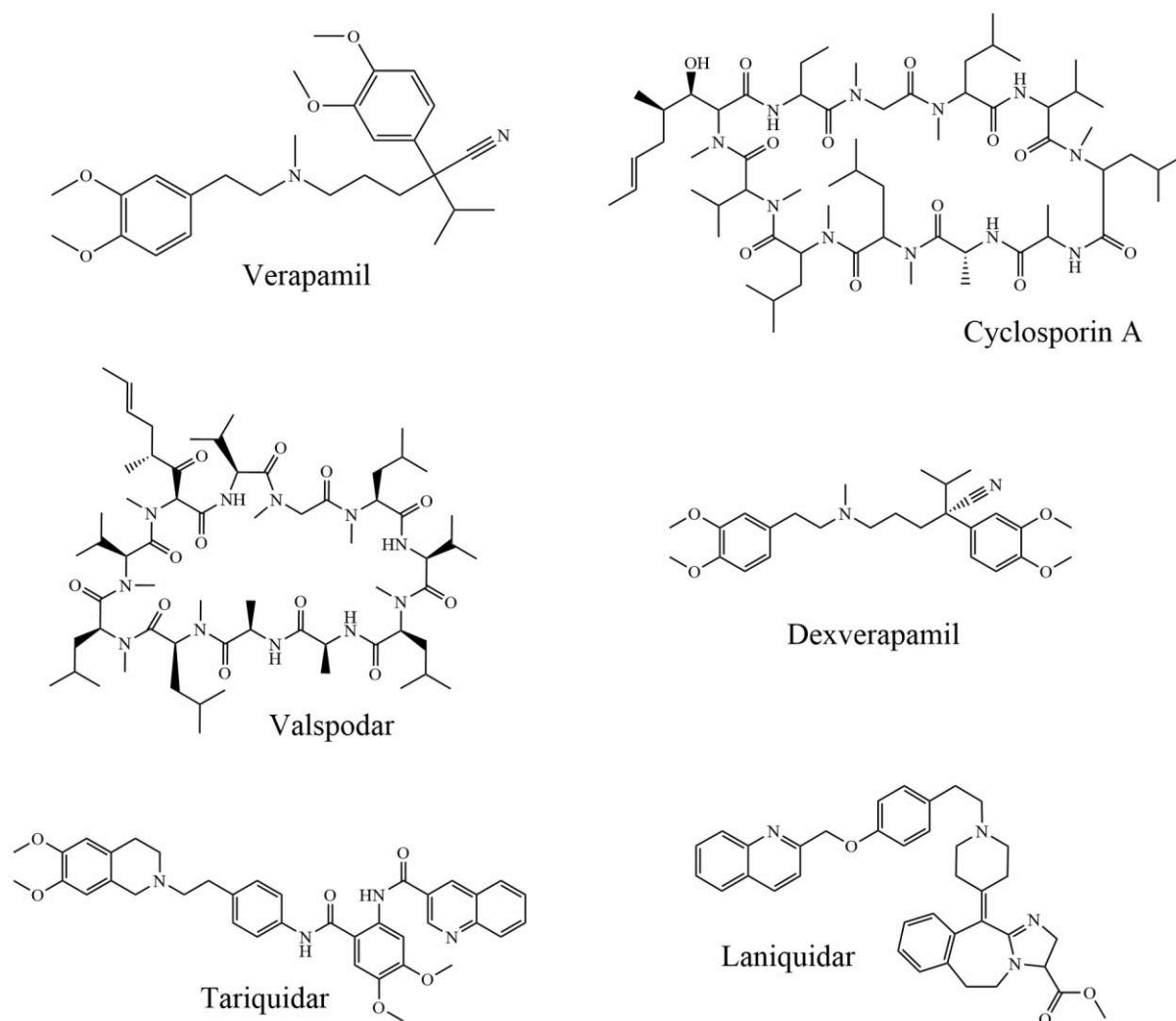
Cancer is one of the most concerning health threats, being responsible for 21% of all mortalities worldwide and 9.1% in developing countries in 2013 (38). In 2018 it is estimated that 18.1 million new cases of cancer appeared and caused 9.6 million deaths, with statistics saying that 1 in 5 men and 1 in 6 women worldwide developing cancer in their lifetime (39).

One of the major obstacles in cancer treatment is multidrug resistance (MDR).

MDR happens when cancer cells present a phenotype that is cross resistant against multiple unrelated drugs that are different, both structurally and functionally, posing a serious health issue (40). MDR can be divided into two types: intrinsic, occurring due to the various genetic characteristics of cells, possibly being present at diagnosis and with cancer cells not responding to standard chemotherapy drugs from the beginning; and acquired, that occurs due to genetic alterations that follow chemotherapy (38,41). Successful treatment of patients is impaired due to the widespread of this phenomenon and, since MDR can include various drugs, from hormones and antibiotics to immunosuppressive agents, MDR inhibitors have become compounds of interest (42).

A common cause of MDR is the increased levels of certain ATP-binding cassette (ABC) transporters in the cell membranes (43). One of these transporters, P-gp (P-glycoprotein), can be blocked by a group of compounds called MDR modulators. Several studies showed that they may enhance the pharmacological treatment when combined with an anticancer drug, by competing with the cytotoxic agent for binding to the active site of membrane transporters and thus reducing drug efflux (44–46). Some of these compounds, like verapamil or cyclosporin A, failed in clinical trials (Figure 3). They belong to the 1st-generation P-gp modulators, which normally are associated with toxicity, lack of specificity and require high doses to be effective. The 2nd-generation modulators, such as valspodar and dexverapamil, which, despite being better than the 1st generations in every point described before, suffer from unwanted interactions with other transport proteins. The 3rd-generation modulators, of which tariquidar and laniquidar are examples, showed high specificity, low toxicity and high potency, and yet have no notable drug interactions with common chemotherapy compounds, however they also failed in clinical trials (Figure 4) (42).





**Figure 4** - Examples of 1st (Verapamil and Cyclosporin A), 2nd (Valspodar and Dexverapamil) and 3rd generation (Tariquidar and Laniquidar) MDR modulators.

#### 1.4 The future of 4<sup>th</sup> generation modulators

Despite several issues, natural products have a great potential for bringing new drugs to the market (47). In the last decade, over 70% of ABC transporter inhibitors reported were either natural products or their respective secondary metabolites, therefore these natural-derived compounds were designated as fourth generation, due to the relevance of these natural sources in finding increasingly ideal ABC inhibitors (46). Natural products are investigated for their low toxicity and selective, but potent, behavior, so plants, marine organisms and fungi have been the main sources for compounds such as flavonoids, coumarins, diterpenes, amongst others (43).

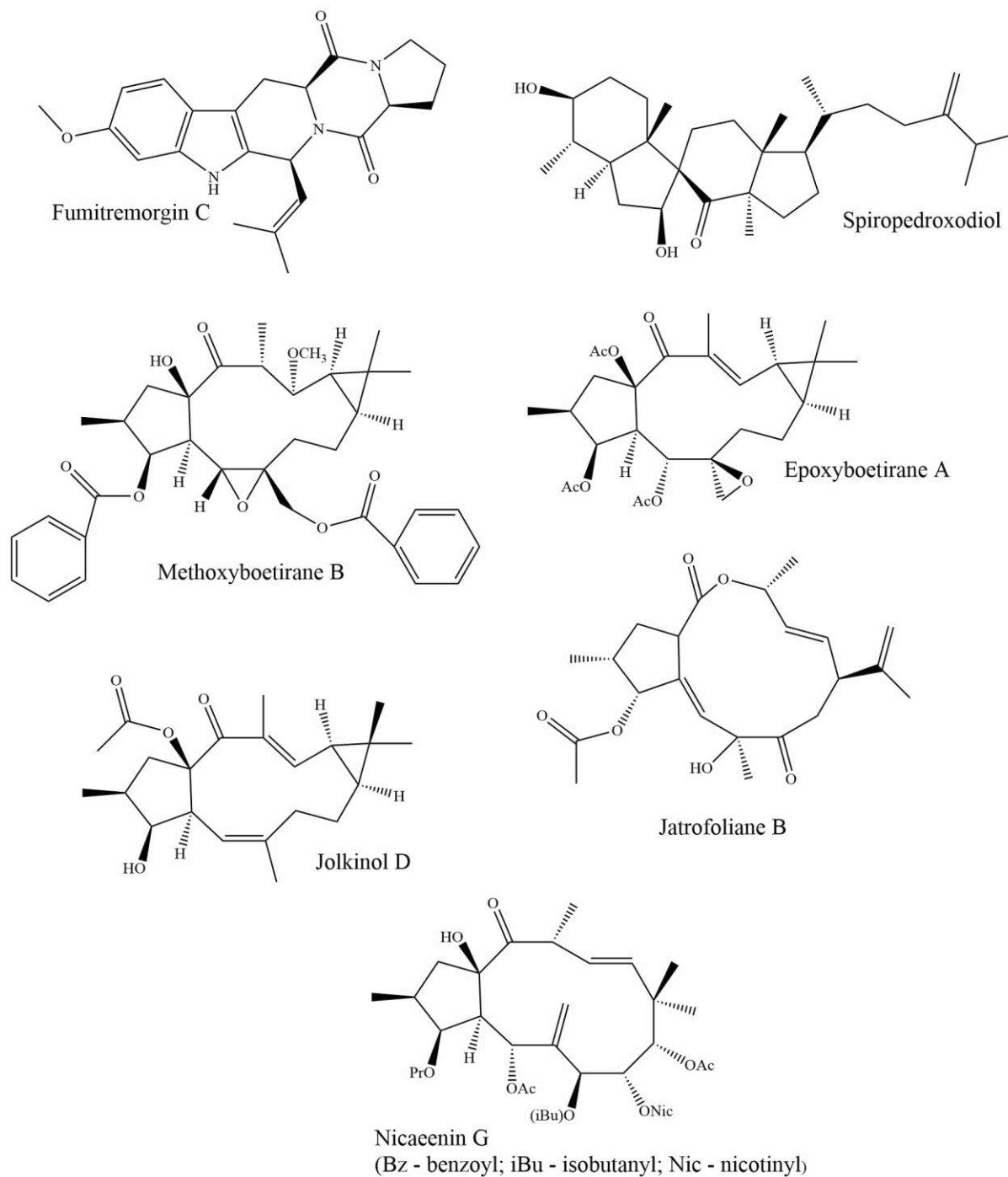
Yet, sometimes analogues of the original compound are needed, as can be seen in the example of fumitremorgin C (Figure 5), a fungal toxin that is a very efficient ABCG2 inhibitor, but displays high neurotoxicity, whereas the analogue generated displays higher potency and selectivity, but less toxicity (45). Another example is found in *Desmodium styracifolium* (Fabaceae), a plant that may have an effect in drug metabolism, particularly P-gp. This activity can be attributed to its contents, like flavonoids/isoflavonoids, alkaloids and triterpenoids, that possess one or more phenolic hydroxyl groups, that, under physiological conditions, can dissociate partly, forming phenolate ions that could interact unselectively with enzymes, possibly leading to protein inactivation (48).

Spiropedroxodiol (Figure 5), a novel compound isolated from *Euphorbia pedroi*, was evaluated for its MDR reversal activity on resistant human colon adenocarcinoma cells (Colo320) and human ABCB1 gene transfected mouse lymphoma cells (L5178YMDR), demonstrating to be a strong MDR reversal agent in both (14). Methoxyboetirane B (Figure 5), a derivative of the macrocyclic diterpene epoxyboetirane A (Figure 5), isolated from *Euphorbia boetica* Boiss., was evaluated for its anti-MDR potential in the same cell lines, being the one that, among other 22 derivatives, demonstrated the most promising activity for the development of MDR-reversing agents (49).

A small library of jolkinol D (Figure 5) derivatives, a macrocyclic diterpene isolated from *Euphorbia piscatoria*, was synthesized. The derivatives were evaluated for their ability to reverse ABCB1-mediated MDR. The results showed that addition of electron-donating groups to the aromatic moieties improved the activity. Some derivatives were evaluated for their collateral sensitivity effect against the human cancer cells EPG85–257 (gastric) and EPP85–181 (pancreatic), and the corresponding drug-selected cells EPG85-257RDB, EPG85-257RNOV, EPP85-181RDB, and EPP85-181RNOV. Some of the compounds were more cytotoxic against the resistant sublines. They were found to induce apoptosis through caspase-3 activation, with noticeable differences between parental and resistant cells (50). Other examples of the anti-MDR potential of macrocyclic diterpenes of the jatrophone and lathyrane-type isolated from *Euphorbia* species, are reported, being the key factor for their inhibition potential their hydrophobicity (51).

The lathyrane jatrolfoliane B (Figure 5), isolated from *Jatropha gossypifolia* Linn (Euphorbiaceae), has shown effects comparable to the 1st generation inhibitor verapamil, in regards to HepG2/ADR and HCT-15/5-FU cell lines, but with less toxicity (52). Nicaeenin G (Figure 5), a jatrophone isolated from *Euphorbia nicaeensis*, was tested as P-gp inhibitor in two human MDR-resistant cell lines: non-small cell lung carcinoma NCI-H460/R and colorectal

carcinoma DLD1-TxR. From all the compounds tested in this study, Nicaeenin G was the most potent inhibitor and showed a prolonged effect comparable do dexverapamil, a known 2nd-generation inhibitor (53).



*Figure 5 - Structure of a few 4th generation modulators.*

## Aim of this work

The aim of this project was to find out new bioactive compounds from *Euphorbia pubescens*. To achieve this goal, the work was divided in two parts. Firstly, the phytochemical study of three fractions of *E. pubescens* methanol extract was carried out, using chromatographic techniques. Then, the molecular manipulation of taraxerone, a pentacyclic triterpene isolated in large amounts, was also performed, taking advantage of the ketone function at C-3. Overall, fourteen compounds were obtained and characterized through spectroscopic techniques.

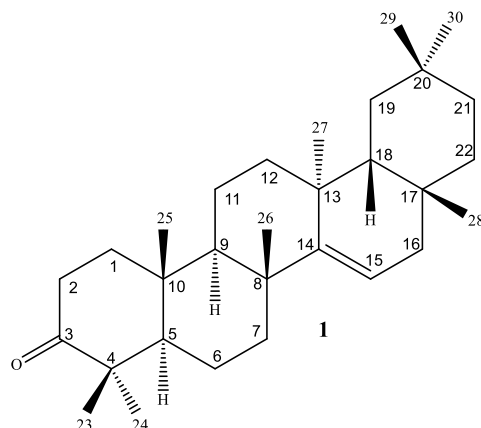
## Results and Discussion

### 1.5 Phytochemical Study of *Euphorbia pubescens*

The phytochemical study of *E. pubescens* aerial parts led to the isolation of ten terpenic compounds, including five pentacyclic triterpenes (**1-3**, **5** and **6**), one steroid (**4**), one jatrophane-type diterpene (**7**), and three *ent*-abietane diterpenic lactones (**8-10**). Moreover, one phenolic compound was also isolated (**11**).

This section presents the description and discussion of physical and spectroscopic data that allowed the structural identification of all the isolated compounds.

#### 1.5.1 Taraxerone



Compound **1** was isolated as white crystals, with a melting point of 141-143 °C and  $[\alpha]_D^{25} +50.0$ .

The IR spectrum presented a characteristic absorption band, for the carbonyl group ( $1708\text{ cm}^{-1}$ ). The ESI-MS spectrum showed a protonated molecular ion at  $m/z$  425  $[M+H]^+$ , that together with the analysis of the NMR data were indicative of the molecular formula  $C_{30}H_{48}O$  (seven degrees of unsaturation), pointing out to the existence of a pentacyclic triterpenic scaffold.

The  $^1\text{H-NMR}$  spectrum (Table 3) showed the presence of eight tertiary methyl groups displayed as singlets at  $\delta_H$  1.14, 1.08 (2 x  $\text{CH}_3$ ) 1.07, 0.95, 0.91 (2x  $\text{CH}_3$ ) and 0.83, and one olefinic proton displayed as a double doublet at  $\delta_H$  5.56 ( $J = 8.2$  and  $3.3$  Hz).

The  $^{13}\text{C-NMR}$  spectrum (Table 4) showed the presence of thirty signals that were discriminated by a DEPT experience as eight quaternary carbons (one olefinic at  $\delta_C$  157.77 and

one carbonyl at  $\delta_C$  217.70), four methines (one  $sp^2$  at  $\delta_C$  117.35), ten methylenes and eight methyl groups.

From this information, and by comparison with the literature(39), compound **1** was identified as taraxerone.

*Table 3 -  $^1H$ -NMR data of taraxerone (1), taraxerol (2) and taraxerol acetate (3) ( $CDCl_3$ , 300 MHz,  $\delta$  in ppm, J in Hz).*

<b>Position</b>	<b>1</b>	<b>2</b>	<b>3</b>
<b>3</b>		3.19 <i>dd</i> ( $J = 15.6, 5.4$ )	4.46 ( <i>dd</i> , $J = 9.8, 6.2$ )
<b>15</b>	5.56 <i>dd</i> ( $J = 8.2, 3.3$ )	5.53 <i>dd</i> ( $J = 8.2, 3.3$ )	5.53 ( <i>dd</i> , $J = 8.1, 3.2$ )
<b>23</b>	1.08 <i>s</i>	0.98 <i>s</i>	0.95 <i>s</i> *
<b>24</b>	1.07 <i>s</i>	0.93 <i>s</i>	0.82 <i>s</i>
<b>25</b>	1.08 <i>s</i>	0.80 <i>s</i>	0.90 <i>s</i> *
<b>26</b>	0.91 <i>s</i> *	0.91 <i>s</i> *	1.09 <i>s</i>
<b>27</b>	1.14 <i>s</i>	1.09 <i>s</i>	0.87 <i>s</i>
<b>28</b>	0.83 <i>s</i>	0.82 <i>s</i>	0.86 <i>s</i>
<b>29</b>	0.95 <i>s</i>	0.95 <i>s</i>	0.95 <i>s</i> *
<b>30</b>	0.91 <i>s</i> *	0.91 <i>s</i> *	0.90 <i>s</i> *
<b>COCH<sub>3</sub></b>			2.04 <i>s</i>

\*Overlapped signals.

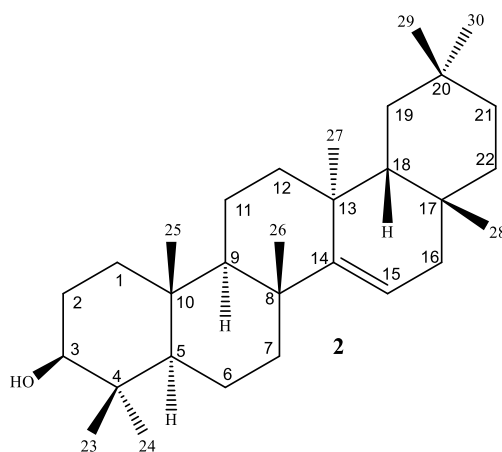
**Table 4** -  $^{13}\text{C}$ -NMR data of taraxerone (**1**), taraxerol (**2**) and taraxerol acetate (**3**) ( $\text{CDCl}_3$ , 75 MHz,  $\delta$  in ppm).

Position	DEPT	<b>1</b>	<b>2</b>	<b>3</b>
<b>1</b>	CH <sub>2</sub>	38.51	38.16	37.84*
<b>2</b>	CH <sub>2</sub>	34.29	27.31	28.95
<b>3</b>	C ( <b>1</b> ) / CH ( <b>2, 3</b> )	217.70	79.23	81.17
<b>4</b>	C	47.73	38.92	39.14
<b>5</b>	CH	55.94	55.69	55.80
<b>6</b>	CH <sub>2</sub>	20.12	18.95	18.84
<b>7</b>	CH <sub>2</sub>	35.26	35.28	35.27
<b>8</b>	C	39.03	39.14	38.05
<b>9</b>	CH	48.86	48.92	48.93
<b>10</b>	C	35.93	37.90	37.84*
<b>11</b>	CH <sub>2</sub>	17.60	17.66	17.66
<b>12</b>	CH <sub>2</sub>	37.90	35.94	35.93
<b>13</b>	C	37.85	37.87	37.55
<b>14</b>	C	157.77	158.25	158.14
<b>15</b>	CH	117.35	117.03	117.09
<b>16</b>	CH <sub>2</sub>	36.83	36.84	36.82
<b>17</b>	C	37.70	37.73	37.71
<b>18</b>	CH	48.96	49.44	49.35
<b>19</b>	CH <sub>2</sub>	40.80	41.49	41.38
<b>20</b>	C	28.95	28.95	29.98
<b>21</b>	CH <sub>2</sub>	33.73	33.86	33.83
<b>22</b>	CH <sub>2</sub>	33.24	33.26	33.25
<b>23</b>	CH <sub>3</sub>	26.27	28.15	28.13
<b>24</b>	CH <sub>3</sub>	21.50	15.58	15.64
<b>25</b>	CH <sub>3</sub>	14.96	15.6	16.73
<b>26</b>	CH <sub>3</sub>	30.01	30.08	31.06
<b>27</b>	CH <sub>3</sub>	25.72	26.06	26.07
<b>28</b>	CH <sub>3</sub>	30.07	29.98	30.07
<b>29</b>	CH <sub>3</sub>	33.51	33.51	33.49
<b>30</b>	CH <sub>3</sub>	21.63	21.47	23.61
<b>COCH<sub>3</sub></b>	C			171.12
<b>COCH<sub>3</sub></b>	CH <sub>3</sub>			21.43

\*Overlapped signals.

Taraxerone has shown to possess anti-tumoral (K562 leukemic cell line and A-549 lung cancer cells) (40,41) and anti-leishmanial [*Leishmania donovani* (strain AG 83)] (40) activity.

## 1.5.2 Taraxerol



Compound **2** was isolated as an amorphous white powder and  $[\alpha]_D^{25} +80.0$ .

The IR spectrum presented an absorption band at  $3487\text{ cm}^{-1}$ , indicating the presence of a hydroxyl group, which was corroborated by the ESI-MS spectrum that revealed an ion at  $m/z$  409  $[M+H-H_2O]^+$ .

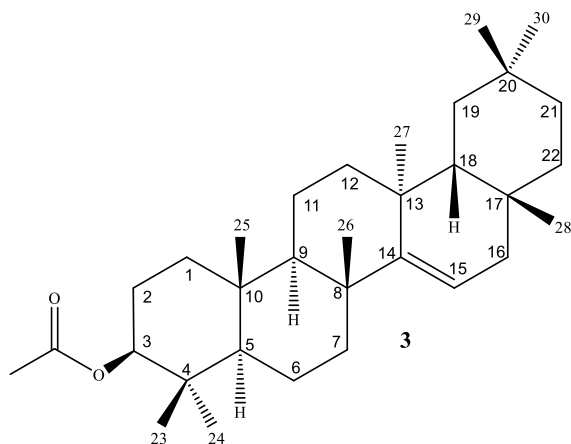
A comparison of the NMR data of compound **2** with those of taraxerone (**1**) showed that the  $^1\text{H}$  and  $^{13}\text{C}$  chemical shifts are almost identical, except those for ring A (Tables 3 and 4). In fact, the major differences in the NMR spectra were the absence of the carbonyl signal at  $\delta_C$  217.70 and the appearance of an oxymethine signal at  $\delta_H$  3.19 (*dd*,  $J = 15.6$  and  $5.4$  Hz) and  $\delta_C$  79.23.

The identity of compound **2** as taraxerol was confirmed by comparison of all spectroscopic data to those reported in the literature (42).

Taraxerol has shown to be able to reduce HeLa cells survival rate by suppressing the PI3K/Akt signaling pathway, thus promoting apoptosis (43). It has also shown inhibitory effects against sarcoma 180 cell line in mice, and is known to be beneficial against diabetes, Parkinson and Alzheimer's (44).



## 1.5.3 Taraxerol Acetate



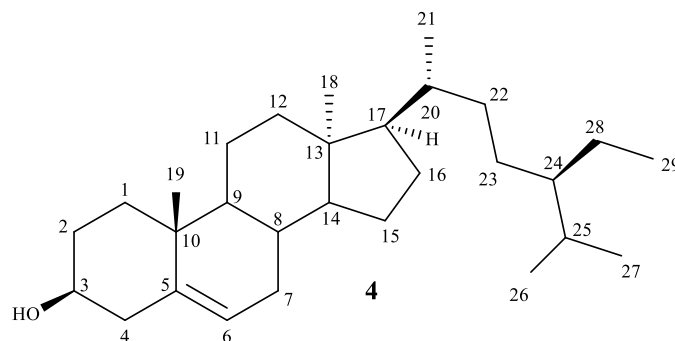
Compound **3** was isolated as an amorphous white powder and  $[\alpha]_{\text{D}}^{25} +55.5$ .

Its IR spectrum displayed an absorption band for an ester group ( $1724$  and  $1252\text{ cm}^{-1}$ ) and the ESI-MS spectrum showed an ion at  $m/z$  409  $[\text{M}+\text{H}-\text{C}_2\text{H}_4\text{O}_2]^+$ .

A comparison of the NMR data of compound **3** with those of taraxerol (**2**) showed that the  $^1\text{H}$  and  $^{13}\text{C}$  chemical shifts were almost identical, except those for ring A (Tables 3 and 4), with the major differences being the appearance of the acetyl group signals at ( $\delta_{\text{H}}$  2.04;  $\delta_{\text{C}}$  171.12 and 21.43) and the downfield shift of the oxymethine signal that appeared at  $\delta_{\text{H}}$  4.46 ( $dd$ ,  $J = 9.8$  and  $6.2$  Hz).

Due to similarities of the spectroscopic data with those of taraxerone (**1**) and taraxerol (**2**), and by comparing the obtained data with the literature(45), compound **3** was identified as taraxerol acetate.

Taraxerol acetate has demonstrated anti-cancer activity against U87 cells, mediated through apoptosis, autophagy, cell cycle arrest and inhibition of cell migration (46). It also showed anti-pyretic (47) and anti-inflammatory activities (48).

1.5.4  $\beta$ -Sitosterol

Compound **4** was isolated as white crystals, m.p. 130-131 °C and  $[\alpha]_D^{25}$  -15.8.

In the IR spectrum a characteristic absorption band for hydroxyl group ( $3421\text{ cm}^{-1}$ ) was observed. The ESI-MS spectrum showed an ion at  $m/z$  397  $[\text{M}+\text{H}-\text{H}_2\text{O}]^+$ .

The  $^1\text{H-NMR}$  spectrum showed signals for six methyl groups, namely two tertiary ( $\delta_{\text{H}}$  0.68 and 1.01), three secondary displayed as doublets at  $\delta_{\text{H}}$  0.80 ( $d$ ,  $J = 6.6$  Hz), 0.83 ( $d$ ,  $J = 6.6$  Hz) and 0.92, and one primary at  $\delta_{\text{H}}$  0.86 ( $t$ ,  $J = 7.2$  Hz). Moreover, this spectrum also showed the presence of a multiplet at  $\delta_{\text{H}}$  3.51 and a doublet at  $\delta_{\text{H}}$  5.34, signals that were assigned to an oxymethine and olefinic protons, respectively (Table 5).

The analysis of the  $^{13}\text{C-NMR}$  spectrum revealed twenty-nine signals, which were discriminated by a DEPT experiment as three quaternary carbons (one olefinic at  $\delta_{\text{C}}$  140.93), nine methines (one oxygenated at  $\delta_{\text{C}}$  71.77 and one olefinic at  $\delta_{\text{C}}$  121.67), eleven methylenes and six methyl carbons (Table 6). From this data it was clear that compound **4** was a  $\Delta^5$ - $3\beta$ -hydroxy sterol.

By comparison of all the spectroscopic data with those described in the literature(49), compound **4** was unambiguously identified as  $\beta$ -sitosterol.

Table 5 -  $^1\text{H-NMR}$  data of  $\beta$ -sitosterol (**4**) ( $\text{CDCl}_3$ , 300 MHz,  $\delta$  in ppm,  $J$  in Hz).

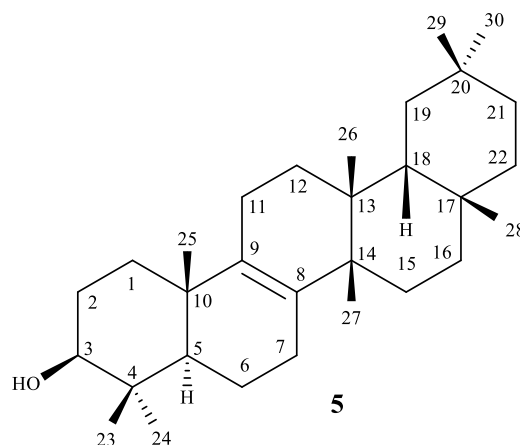
Position	<b>4</b>
<b>3</b>	3.52 <i>m</i>
<b>6</b>	5.35 <i>d</i> (5.4)
<b>18</b>	0.68 <i>s</i>
<b>19</b>	1.01 <i>s</i>
<b>21</b>	0.92 <i>d</i> (6.5)
<b>26</b>	0.80 <i>d</i> ( $J = 6.6$ )
<b>27</b>	0.83 <i>d</i> ( $J = 6.5$ )
<b>29</b>	0.86 <i>t</i> ( $J = 7.2$ )

*Table 6* -  $^{13}\text{C}$ -NMR data of  $\beta$ -sitosterol (**4**) ( $\text{CDCl}_3$ , 75 MHz,  $\delta$  in ppm).

Position	DEPT	<b>4</b>	Position	DEPT	<b>4</b>
<b>1</b>	CH <sub>2</sub>	37.42	<b>15</b>	CH <sub>2</sub>	26.26
<b>2</b>	CH <sub>2</sub>	31.85	<b>16</b>	CH <sub>2</sub>	28.20
<b>3</b>	CH	71.77	<b>17</b>	CH	56.23
<b>4</b>	CH <sub>2</sub>	42.27	<b>18</b>	CH <sub>3</sub>	36.30
<b>5</b>	C	140.93	<b>19</b>	CH <sub>3</sub>	18.99
<b>6</b>	CH	121.67	<b>20</b>	CH	34.12
<b>7</b>	CH <sub>2</sub>	32.08	<b>21</b>	CH <sub>3</sub>	26.06
<b>8</b>	CH	31.64	<b>22</b>	CH <sub>2</sub>	46.02
<b>9</b>	CH	50.31	<b>23</b>	CH <sub>2</sub>	23.24
<b>10</b>	C	36.67	<b>24</b>	CH	11.93
<b>11</b>	CH <sub>2</sub>	21.04	<b>25</b>	CH	29.13
<b>12</b>	CH <sub>2</sub>	39.79	<b>26</b>	CH <sub>3</sub>	19.76
<b>13</b>	C	42.49	<b>27</b>	CH <sub>3</sub>	19.35
<b>14</b>	CH	56.73	<b>28</b>	CH <sub>2</sub>	18.73
			<b>29</b>	CH <sub>3</sub>	11.81

$\beta$ -sitosterol has shown to have anti-cancer activity against TPA-induced EBV-EA activation in Raji cells, as PI3K inhibitor (50); MCF-7 tumor cell lines and MDA-MB-231 human breast cancer cell line (51). It also displayed good analgesic and anti-inflammatory activities, working through central mechanism by opioid receptors (52).

## 1.5.5 Isomultiflorenol



Compound **5** was isolated as an amorphous white powder and  $[\alpha]_D^{25} +88.2$ .

The IR spectrum exhibited an absorption band at  $3410\text{ cm}^{-1}$  that together with the ion at  $m/z\ 409\ [M+H-H_2O]^+$  in the ESI-MS suggested the presence of a hydroxyl group.

The  $^1\text{H-NMR}$  spectrum (Table 7) showed a double doublet signal corresponding to an axial oxymethine proton at  $\delta_{\text{H}}\ 3.23$  ( $J = 11.4$  and  $4.8$  Hz) and signals for eight tertiary methyl groups ( $\delta_{\text{H}}\ 1.07, 1.06, 1.00, 0.99, 0.97, 0.96$  ( $2 \times \text{CH}_3$ ) and  $0.80$ ).

The  $^{13}\text{C-NMR}$  spectrum presented thirty signals that were discriminated by a DEPT experiment as eight quaternary carbons (two olefinic at  $\delta_{\text{C}}\ 133.67$  and  $135.24$ ), three methines (one oxygenated at  $\delta_{\text{C}}\ 79.18$ ), eleven methylenes and eight methyl carbons (Table 8).

By comparison of all spectroscopic data with those reported in the literature(53), compound **5** was undoubtedly identified as isomultiflorenol.

*Table 7 -  $^1\text{H-NMR}$  data of isomultiflorenol (**5**) and isomultiflorenyl acetate (**6**) ( $\text{CDCl}_3$ , 300 MHz,  $\delta$  in ppm,  $J$  in Hz).*

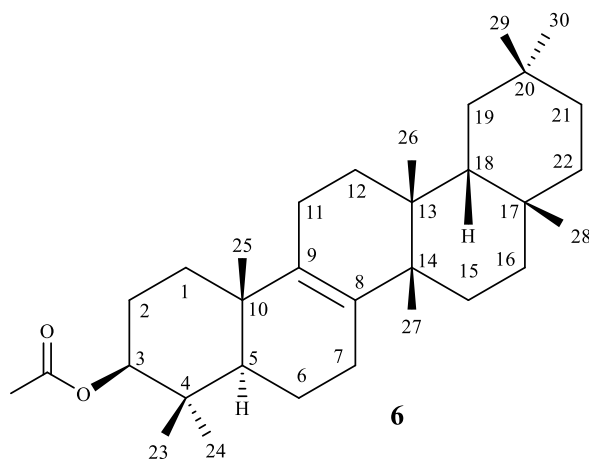
Position	<b>5</b>	<b>6</b>	Position	<b>5</b>	<b>6</b>
<b>3</b>	3.23 dd ( $J = 11.4, 4.8$ )	4.48 dd ( $J = 11.3, 6.5$ )	<b>27</b>	0.97 s	1.05 s
<b>23</b>	0.80 s	0.87 s	<b>28</b>	1.07 s	1.07 s
<b>24</b>	0.96 s*	0.88 s	<b>29</b>	0.99 s	0.95 s
<b>25</b>	0.96 s*	0.97 s*	<b>30</b>	1.00 s	0.97 s*
<b>26</b>	1.06 s	0.97 s*	<b>COCH<sub>3</sub></b>		2.05 s

\*Overlapped signals

Table 8 -  $^{13}\text{C}$ -NMR data of compounds **5** and **6** ( $\text{CDCl}_3$ , 75 MHz,  $\delta$  in ppm).

Position	DEPT	5	6	Position	DEPT	5	6
1	CH <sub>2</sub>	34.34	34.76	28	CH <sub>3</sub>	31.65	31.63
2	CH <sub>2</sub>	24.84	24.32	29	CH <sub>3</sub>	34.66	34.45
3	CH	79.18	81.10	30	CH <sub>3</sub>	33.14	33.07
4	C	38.96	37.84	COCH <sub>3</sub>	C		171.25
5	CH	50.86	50.87	COCH <sub>3</sub>	CH <sub>3</sub>		21.49
6	CH <sub>2</sub>	19.39	19.20				
7	CH <sub>2</sub>	28.08	27.46				
8	C	135.24	135.23				
9	C	133.67	133.45				
10	C	37.77	37.56				
11	CH <sub>2</sub>	20.96	20.94				
12	CH <sub>2</sub>	30.84	30.74				
13	C	37.44	37.35				
14	C	41.14	41.11				
15	CH <sub>2</sub>	27.65	26.30				
16	CH <sub>2</sub>	36.85	36.83				
17	C	30.95	30.90				
18	CH	44.27	44.23				
19	CH <sub>2</sub>	34.55	34.31				
20	C	28.48	28.45				
21	CH <sub>2</sub>	35.20	43.11				
22	CH <sub>2</sub>	36.96	36.81				
23	CH <sub>3</sub>	28.20	16.83				
24	CH <sub>3</sub>	15.76	28.11				
25	CH <sub>3</sub>	19.97	20.00				
26	CH <sub>3</sub>	19.12	19.04				
27	CH <sub>3</sub>	26.41	24.74				

## 1.5.6 Isomultiflorenyl Acetate



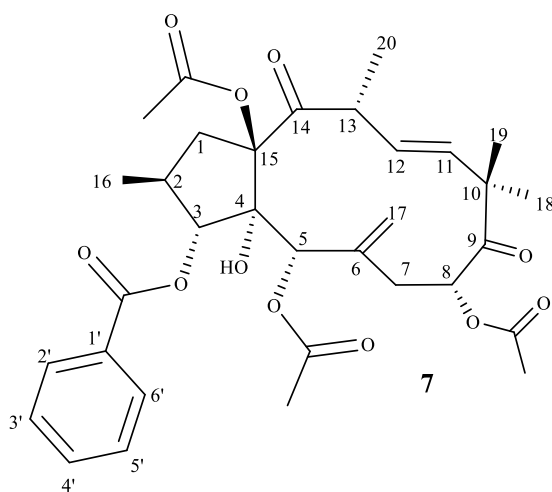
Compound **6** was isolated as an amorphous white powder and  $[\alpha]_{\text{D}}^{25} +98.8$ .

The IR spectrum was similar to compound **5**, except for the absence of the hydroxyl absorption band and the presence of additional absorption bands for the ester carbonyl group at  $1726\text{ cm}^{-1}$  and  $1250\text{ cm}^{-1}$ .

A comparison of the NMR data of compound **6** with those of isomultiflorenol (**5**) showed that the  $^1\text{H}$  and  $^{13}\text{C}$  chemical shifts were almost identical, except those for ring A (Table 7 and 8). The major differences were the presence of an acetyl group at C-3 ( $\delta_{\text{C}}$  81.10), indicated by the downfield shift of oxymethine signal ( $\delta_{\text{H}}$  4.48 *dd* ( $J = 11.3$  and  $6.5$  Hz)) and by the existence of the acetyl signals at  $\delta_{\text{H}}$  2.05 (*s*) and  $\delta_{\text{C}}$  171.25 and 21.49. The presence of the acetyl group was also confirmed by the ESI-MS spectrum that showed an ion at  $m/z$  409 [ $\text{M}+\text{H}-\text{C}_2\text{H}_4\text{O}_2$ ] $^+$ .

With all this information, and through comparison with isomultiflorenol and the data reported from literature(54), compound **6** was unambiguously identified as isomultiflorenyl acetate.

### 1.5.7 Euphopubescenol



Compound **7** was isolated as an amorphous white powder and with a  $[\alpha]_{\text{D}}^{25} +74.6$ .

The IR spectrum revealed a band at  $3520\text{ cm}^{-1}$  for a hydroxyl group and bands at  $1748$  and  $1715\text{ cm}^{-1}$  for carbonyl groups (carbonyl ester and ketone, respectively). A band at  $1633\text{ cm}^{-1}$  supports the presence of at least double bond. The ESI-MS spectrum showed an ion at  $m/z$  553 [ $\text{M}+\text{H}-\text{CH}_3\text{COOH}$ ] $^+$ .

The  $^1\text{H}$ -NMR spectrum showed two signals for tertiary methyls at  $\delta_{\text{H}}$  1.46 and 1.22, two secondary methyls as doublet signals at  $\delta_{\text{H}}$  1.38 ( $J = 6.5$  Hz) and 1.24 ( $J = 6.9$  Hz), two doublet signals and one singlet for protons geminal to ester groups at  $\delta_{\text{H}}$  5.67 ( $J = 8.9$  Hz), 5.61 ( $J =$

10.5 Hz) and  $\delta_{\text{H}}$  4.86. Moreover, signals corresponding to one exocyclic double bond indicated by two singlets at  $\delta_{\text{H}}$  4.86 and 5.12 and a *trans*-disubstituted one ( $\delta_{\text{H}}$  5.89 *d*,  $J = 9.0$  Hz, and 5.92 *dd*,  $J = 16.0, 9.0$  Hz) were also observed. In addition, one benzoate group ( $\delta_{\text{H}}$  7.42, 7.57 and 7.97) and three acetyl groups (at  $\delta_{\text{H}}$  1.91, 2.08 and 2.16) were also identified. Also noteworthy is the presence of a signal at  $\delta_{\text{H}}$  3.46 without correlation in the HMQC spectrum, which suggested the presence of a tertiary hydroxyl group in the molecule (Table 9).

In the  $^{13}\text{C}$ -NMR, thirty-one carbon signals were present, but as eleven of these signals belonged to the ester functions, the remaining twenty carbon signals were assigned to the main scaffold and were, through a DEPT experience, sorted as six quaternary carbons (two carbonyl groups at  $\delta_{\text{C}}$  208.20 and 204.61, two oxygenated carbons at  $\delta_{\text{C}}$  82.94 and 94.62, and one  $\text{sp}^2$  carbon at  $\delta_{\text{C}}$  140.17), seven methines (two  $\text{sp}^2$  at  $\delta_{\text{C}}$  136.91 and 132.84, and three oxygenated at  $\delta_{\text{C}}$  77.89, 73.28 and 72.41), three methylenes (one  $\text{sp}^2$  at  $\delta_{\text{C}}$  116.89) and four methyl groups (Table 10).

On the basis of all these data, a jatrophane-type skeleton was proposed for compound **7**. The unambiguous assignment of all proton and carbon resonances was possible through the combined analysis of 2D-NMR spectra, namely COSY, HMQC and HMBC (Figure 6). In particular,  $^1\text{H}$ - $^1\text{H}$ -COSY and HMQC spectra defined four structural fragments (A-D) with correlated protons (Figure 6), which are separated by quaternary carbons. The linkage of fragments A and D was possible through the  $^2J_{\text{C-H}}$  and  $^3J_{\text{C-H}}$  correlations displayed in the HMBC spectrum, namely, those between C-15 and H-1 and those between C-14 and H-1 and H-20. Moreover, heteronuclear  $^2J_{\text{C-H}}$  and  $^3J_{\text{C-H}}$  correlations allowed the assignment of the ketone function at C-9, which was connected with the fragment D due to the observed cross peaks between C-9 with H-18 and H-19, and C-10 with H-11 and H-12.

Analysis of the HMBC spectrum also allowed the connection between fragments B and C, sustained by the correlations found between C-6 with H-7 and H-8 together with those between C-8 and H-17 $\alpha$  and H-17 $\beta$ . The linkage of fragments B and A was possible due to the correlations found between C-6 and H-5, and those found between C-15 and C-17 with H-5, together with the information found in the literature. Analysis of HMBC spectrum additionally permitted to establish the position of the benzoyl group at C-3 and acetate groups at C-5 and C-8. All the spectroscopic data were in agreement to those described in the literature for euphobubescenol. (55)

Table 9 -  $^1\text{H-NMR}$  data of euphopubescenol (7) ( $\text{CDCl}_3$ , 300 MHz,  $\delta$  in ppm,  $J$  in Hz).

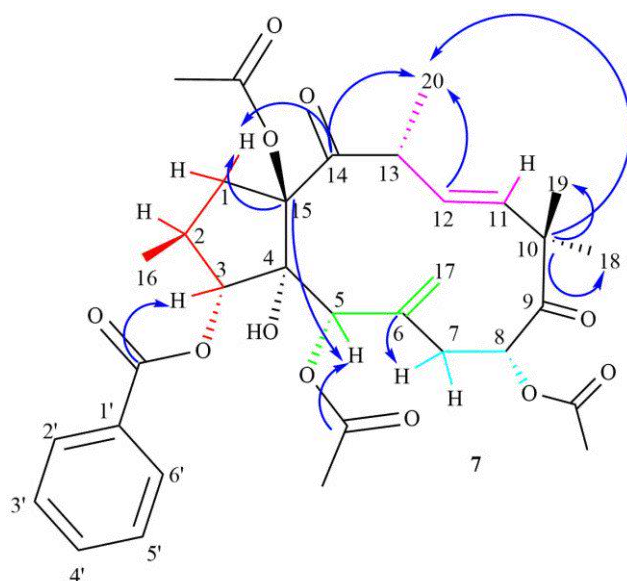
Position	7	Position	7
1	2.43 $m^*$	17	4.85 s (H- $\alpha$ ) / 5.12 s (H- $\beta$ )
2	2.91 $tdd$ ( $J = 10.4, 7.0, 3.5$ )	19	1.46 s
3	5.61 $d$ ( $J = 10.5$ )	20	1.38 $d$ ( $J = 6.5$ )
5	4.86 s	Bz (2',6')	7.97 $dd$ ( $J = 8.0; 1.5$ )
7	1.71 $dd$ ( $J = 14.9, 9.2$ , H- $\alpha$ ) / 2.00 s (H- $\beta$ )	Bz (3',5')	7.42 $t$ ( $J = 7.8$ )
8	5.67 $d$ ( $J = 8.9$ )	Bz (4')	7.57 $tt$ ( $J = 7.5, 1.5$ )
11	5.89 $d$ ( $J = 9.0$ )	5-O(COCH <sub>3</sub> )	2.08 s
12	5.92 $dd$ ( $J = 16.0, 9.0$ )	8-O(COCH <sub>3</sub> )	1.91 s
13	3.85 $m$	15-O(COCH <sub>3</sub> )	2.16 s
16	1.24 $d$ ( $J = 6.9$ )	18	1.22 s

\* This value corresponds to either protons 1 $\alpha$  and 1 $\beta$ .

Table 10 -  $^{13}\text{C-NMR}$  data of euphopubescenol (7) ( $\text{CDCl}_3$ , 75 MHz,  $\delta$  in ppm).

Position	DEPT	7	Position	DEPT	7
1	CH <sub>2</sub>	38.82	17	CH <sub>2</sub>	116.89
2	CH	36.90	18	CH <sub>3</sub>	24.57
3	CH	77.89	19	CH <sub>3</sub>	24.57
4	C	82.94	20	CH <sub>3</sub>	20.16
5	CH	73.28	3-OBz	C	166.60
6	C	140.17	Bz (1')	C	129.09
7	CH <sub>2</sub>	35.97	Bz (2',6')	CH	130.19
8	CH	72.41	Bz (3',5')	CH	128.53
9	C	208.20	Bz (4')	CH	133.70
10	C	50.66	5-O(COCH <sub>3</sub> )	C	168.42
11	CH	136.91		CH <sub>3</sub>	20.71
12	CH	132.84	8-O(COCH <sub>3</sub> )	C	169.87
13	CH	46.91		CH <sub>3</sub>	20.42
14	C	204.61	15-O(COCH <sub>3</sub> )	C	168.86
15	C	94.62		CH <sub>3</sub>	21.14
16	CH <sub>3</sub>	17.79			

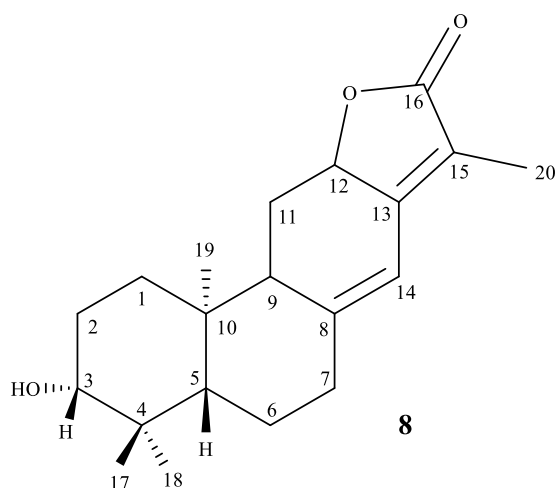




**Figure 6** - Euphobubescenol, with the colored lines indicating the different  $^1\text{H}$ -spin systems, assigned through HMQC and  $^1\text{H}$ - $^1\text{H}$  COSY, and respective correlations displayed in the HMBC spectrum (indicated by arrows).

Euphobubescenol (**7**) has demonstrated activity as inhibitor of multidrug transporters in *Saccharomyces cerevisiae* cells overexpressing the CaCdr1p ABC-transporter (AD-CDR1) (56), and also showed anti-tumoral activity as it inhibited the growth of MCF-7 and NCI-H460 cancer cell lines (55).

### 1.5.8 Helioscopinolide A



Compound **8** was isolated as white crystals, with m.p. of 204-206 °C and  $[\alpha]_{\text{D}}^{25} +587.4$ .

The IR spectrum indicated the presence of a hydroxyl function ( $3478\text{ cm}^{-1}$ ), and an unsaturated lactone ( $1734\text{ cm}^{-1}$ ). The ESI-MS spectrum showed a protonated molecular ion at  $m/z\ 317\ [M+H]^+$ , that together with the analysis of the NMR data were indicative of the molecular formula  $C_{20}H_{28}O_3$  (seven degrees of unsaturation).

The  $^1\text{H-NMR}$  spectrum (Table 11) showed three singlets for tertiary methyl groups at  $\delta_{\text{H}}\ 0.81, 0.92, 1.02$ , and one vinylic methyl group at  $\delta_{\text{H}}\ 1.81$  ( $d, J = 1.7\text{ Hz}$ ). Signals corresponding to two oxymethine protons at  $\delta_{\text{H}}\ 3.27$  ( $dd, J = 11.6$  and  $4.4\text{ Hz}$ ) and  $4.85$  ( $ddd, J = 13.6, 6.2$  and  $1.8$ ), and one olefinic proton at  $\delta_{\text{H}}\ 6.27$  ( $s$ ), were also observed.

The  $^{13}\text{C-NMR}$  spectrum (Table 12) exhibited twenty carbons resonances, that were discriminated by a DEPT experiment as four methyl groups (one vinylic at  $\delta_{\text{C}}\ 8.37$ ), five methylenes, five methines (one olefinic at  $\delta_{\text{C}}\ 114.31$  and two oxygenated at  $\delta_{\text{C}}\ 76.0$  and  $78.66$ ), and six quaternary carbons (three  $\text{sp}^2$  carbons at  $\delta_{\text{C}}\ 116.61, 151.58$  and  $156.17$  and one carbonyl group at  $\delta_{\text{C}}\ 175.39$ ).

On the basis of all these data and by the comparison with the literature (57), compound (**8**) was identified as the *ent*-abietane diterpenic lactone helioscopinolide A.

*Table 11* -  $^1\text{H-NMR}$  data of helioscopinolide A (**8**), helioscopinolide B (**9**) and helioscopinolide E (**10**) ( $\text{CDCl}_3, 300\text{ MHz}, \delta$  in ppm,  $J$  in Hz).

Position	<b>8</b>	<b>9</b>	<b>10</b>
<b>1</b>	1.94 $dt$ ( $J = 13.1, 3.6$ )	1.60-1.70 $m^*$	2.22 $m^*$
<b>2</b>	1.64 $m$	1.60-1.70 $m^*$	2.56 $m$
<b>3</b>	3.27 $dd$ ( $J = 11.6, 4.4$ )	3.46 $t$ ( $J = 2.7, 2.7$ )	
<b>5</b>	1.14 $dd$ ( $J = 12.5, 2.5$ )	1.64 $m$	1.59 $m^{**}$
<b>6</b>		1.41 $m$	1.59 $m^{**}$
<b>7</b>		2.47 $dd$ ( $J = 4.2, 2.3$ )	2.22 $m^*$
<b>9</b>	2.18 $dd$ ( $J = 15.1, 9.0$ )	2.29 $d$ ( $J = 8.7$ )	
<b>11</b>	2.53 $m$	2.56 $m$	1.59 $m^{**}$
<b>12</b>	4.85 $ddd$ ( $J = 13.6, 6.2, 1.8$ )	4.86 $ddd$ ( $J = 13.3, 6.1, 1.2$ )	4.87 $ddd$ ( $J = 13.4, 6.1, 1.8$ )
<b>14</b>	6.27 $s$	6.25 $s$	6.33 $s$
<b>17</b>	1.02 $s$	0.98 $s$	1.13 $s$
<b>18</b>	0.81 $s$	0.86 $s$	1.05 $s$
<b>19</b>	0.92 $s$	0.93 $s$	1.08 $s$
<b>20</b>	1.81 $d$ ( $J = 1.7$ )	1.81 $d$ ( $J = 1.3$ )	1.84 $d$ ( $J = 1.6$ )

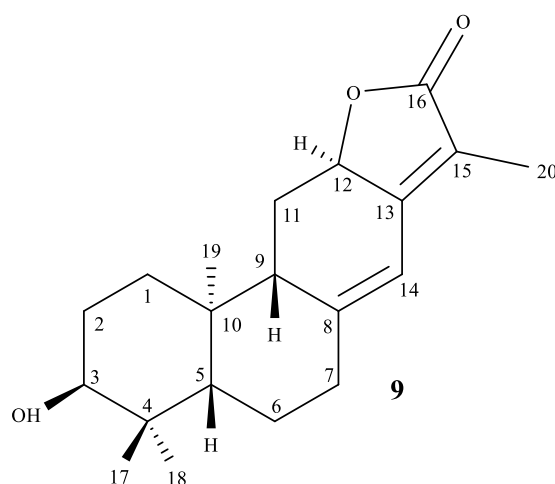
\*/\*\* Overlapped signals

**Table 12** -  $^{13}\text{C}$ -NMR data of helioscopinolide A (**8**), helioscopinolide B (**9**) and helioscopinolide E (**10**)  
( $\text{CDCl}_3$ , 75 MHz,  $\delta$  in ppm).

Position	DEPT	<b>8</b>	<b>9</b>	<b>10</b>
<b>1</b>	$\text{CH}_2$	37.53	32.19	37.53
<b>2</b>	$\text{CH}_2$	27.70	25.81	34.53
<b>3</b>	$\text{CH}$ ( <b>8,9</b> ) / $\text{C}$ ( <b>10</b> )	78.66	75.62	215.63
<b>4</b>	$\text{C}$	39.21	37.85	47.67
<b>5</b>	$\text{CH}$	54.48	48.41	45.90
<b>6</b>	$\text{CH}_2$	23.56	23.49	24.74
<b>7</b>	$\text{CH}_2$	37.04	37.14	36.73
<b>8</b>	$\text{C}$	151.58	152.23	150.30
<b>9</b>	$\text{CH}$	51.67	51.64	50.80
<b>10</b>	$\text{C}$	41.34	41.39	41.06
<b>11</b>	$\text{CH}_2$	27.66	27.56	27.94
<b>12</b>	$\text{CH}$	76.00	76.12	75.75
<b>13</b>	$\text{C}$	156.17	156.30	155.65
<b>14</b>	$\text{CH}$	114.31	114.12	114.87
<b>15</b>	$\text{C}$	116.61	116.37	117.87
<b>16</b>	$\text{C}$	175.39	175.45	175.17
<b>17</b>	$\text{CH}_3$	28.79	28.83	26.66
<b>18</b>	$\text{CH}_3$	15.71	22.33	21.93
<b>19</b>	$\text{CH}_3$	16.82	16.81	16.38
<b>20</b>	$\text{CH}_3$	8.37	8.33	8.45

Helioscopinolide A has shown activity against *Staphylococcus aureus* (58), and as an inhibitor of ROS and proinflammatory cytokines in LPS-stimulated BV2 microglial cells by suppressing the expression of COX-2 and iNOS and inhibiting the enzyme activity (59), showing to be promising as therapeutic agent in treatment of neurodegenerative and neuroinflammatory diseases.

### 1.5.9 Helioscopinolide B



Compound **9** was isolated as a colorless oil and  $[\alpha]_D^{25} +417.6$ .

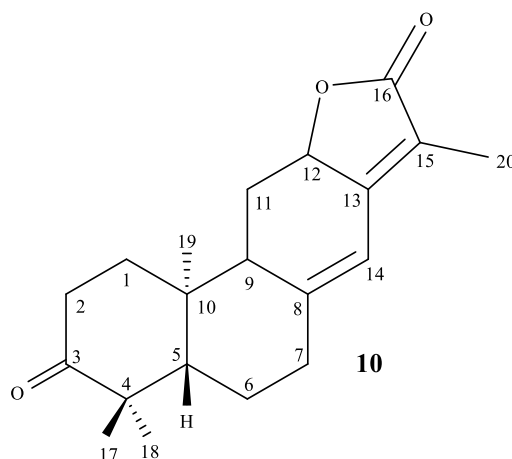
The IR spectrum displayed absorption bands for hydroxyl  $3467\text{ cm}^{-1}$  and for a lactone carbonyl group at  $1730\text{ cm}^{-1}$ . The ESI-MS spectrum showed a protonated molecular ion at  $m/z$  317  $[M+H]^+$ , that together with the analysis of the NMR data were indicative of the molecular formula  $C_{20}H_{28}O_3$  (seven degrees of unsaturation), pointing to the same structure as helioscopinolide A (**8**).

The NMR data for both compounds are very similar (Tables 11 and 12), with the exception of the signal of H-3. In compound **8** this signal appeared as a double doublet at  $\delta_H$  3.26 ( $J = 11.6$  and  $4.4$  Hz). On the other hand, in compound **9**, H-3 signal is a triplet located downfield at  $\delta_H$  3.46 ( $J = 2.7, 2.7$  Hz). This difference could be justified by the different configuration at C-3.

Through comparison with the literature (60), and by analysis of the spectra obtained, compound **9** was identified as helioscopinolide B.

Helioscopinolide B, like helioscopinolide A, has shown activity against *Staphylococcus aureus* (58), and antineoplastic and MDR modulatory activities against gastric (EPG85-257), pancreatic (EPP85-181) and colon (HT-29) carcinomas and their resistant phenotypes (EPG85-257/RDB), being particularly effective against the drug-resistance sublines (61).

#### 1.5.10 Helioscopinolide E



Compound **10** was isolated as a colorless oil and with a  $[\alpha]_D^{25} +395.9$ .

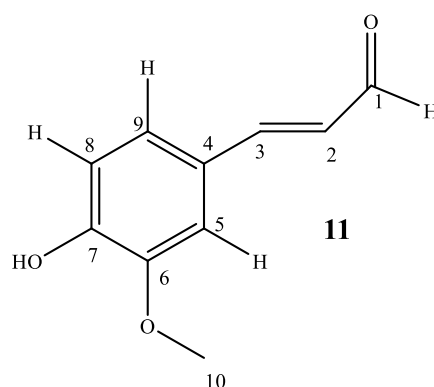
The IR spectrum displayed strong absorption bands for carbonyl groups at  $1706\text{ cm}^{-1}$  and  $1751\text{ cm}^{-1}$ , and the ESI-MS spectrum showed a protonated molecular ion at  $m/z$  317  $[M+H]^+$ , that together with the analysis of the NMR data were indicative of the molecular

formula  $C_{20}H_{26}O_3$  (eight degrees of unsaturation), suggesting a similar structure as helioscopinolide A (**8**) and B (**9**). Comparing the NMR spectra (Tables 11 and 12) of compound **10** with those of helioscopinolide B (**9**), the most important difference was the absence of the C-3 oxymethine signal ( $\delta_H$  3.46 and  $\delta_C$  75.62) that was replaced by ketone carbonyl group ( $\delta_C$  215.63).

Through analysis of the spectra, and comparison with helioscopinolide A, B and the literature (62), it was determined that compound **10** is helioscopinolide E.

Helioscopinolide E, like helioscopinolide B, has also shown promising antineoplastic and MDR modulatory activity against gastric (EPG85-257), pancreatic (EPP85-181) and colon (HT-29) carcinomas and their resistant phenotypes (EPG85-257/RDB) (61).

#### 1.5.11 Coniferaldehyde



Compound **11** was isolated as an amorphous light-brown powder.

The IR spectrum showed an absorption band at  $1691\text{ cm}^{-1}$  that is characteristic of unsaturated carbonyl groups. Bands at  $3357\text{ cm}^{-1}$  and  $1618\text{ cm}^{-1}$  suggested the presence of a hydroxyl group and an alkene, respectively. The protonated molecular ion at  $m/z$  179  $[M+H]^+$  showed to be compatible with the molecular formula  $C_{10}H_{10}O_3$  (six degrees of unsaturation), suggesting the presence of an aromatic ring and two double bonds.

The  $^1\text{H-NMR}$  spectrum presented a methyl singlet at  $\delta_H$  3.94, and a signal belonging to the proton of an aldehyde function at  $\delta_H$  9.65 ( $d$ ,  $J = 7.7$  Hz). Also shown were the signals for protons of the benzene ring at  $\delta_H$  6.96 ( $d$ ,  $J = 8.1$  Hz), 7.07 ( $d$ ,  $J = 1.9$  Hz) and 7.12 ( $dd$ ,  $J = 8.3$ , 2.0 Hz), and two signals for olefinic protons at  $\delta_H$  6.59 ( $dd$ ,  $J = 15.8$ , 7.7 Hz) and 7.42 ( $d$ ,  $J = 15.7$  Hz) (Table 13).

The  $^{13}\text{C-NMR}$  showed ten signals that were differentiated through DEPT, into three quaternary carbons ( $\delta_C$  126.80, 147.12 and 149.12), six methynes (the signal at  $\delta_C$  193.77 belonging to a carbon of an aldehyde group) and one methyl carbon ( $\delta_C$  56.15) (Table 14).

Taking into account the NMR data and comparison with the literature (63), compound **11** was identified as coniferaldehyde. All proton and carbon signals were unambiguously assigned by two-dimensional NMR experiments (Figure 7).

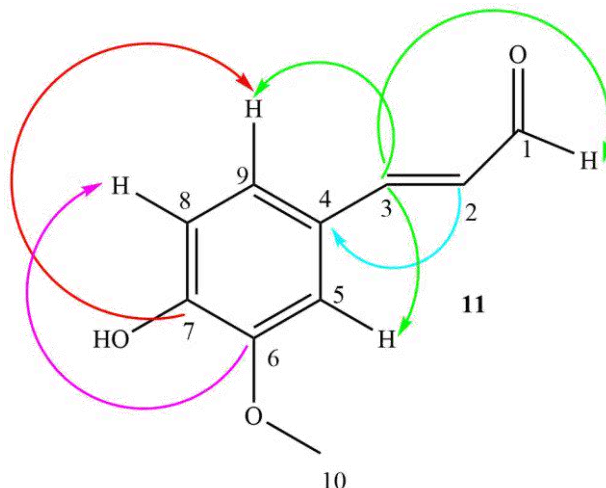


Figure 7 -  $^3J_{C-H}$  correlations displayed in the HMBC spectrum of compound **11**.

Table 13-  $^1H$ -NMR data of coniferaldehyde (**11**) ( $CDCl_3$ , 300 MHz,  $\delta$  in ppm,  $J$  in Hz).

Position	<b>11</b>
<b>1</b>	9.65 <i>d</i> ( $J = 7.7$ )
<b>2</b>	6.59 <i>dd</i> ( $J = 15.8, 7.7$ )
<b>3</b>	7.42 <i>d</i> ( $J = 15.7$ )
<b>5</b>	7.07 <i>d</i> ( $J = 1.9$ )
<b>8</b>	6.96 <i>d</i> ( $J = 8.1$ )
<b>9</b>	7.12 <i>dd</i> ( $J = 8.3, 2.0$ )
<b>10</b>	3.94 <i>s</i>

Table 14 -  $^{13}C$ -NMR data of coniferaldehyde (**11**) ( $CDCl_3$ , 300 MHz,  $\delta$  in ppm).

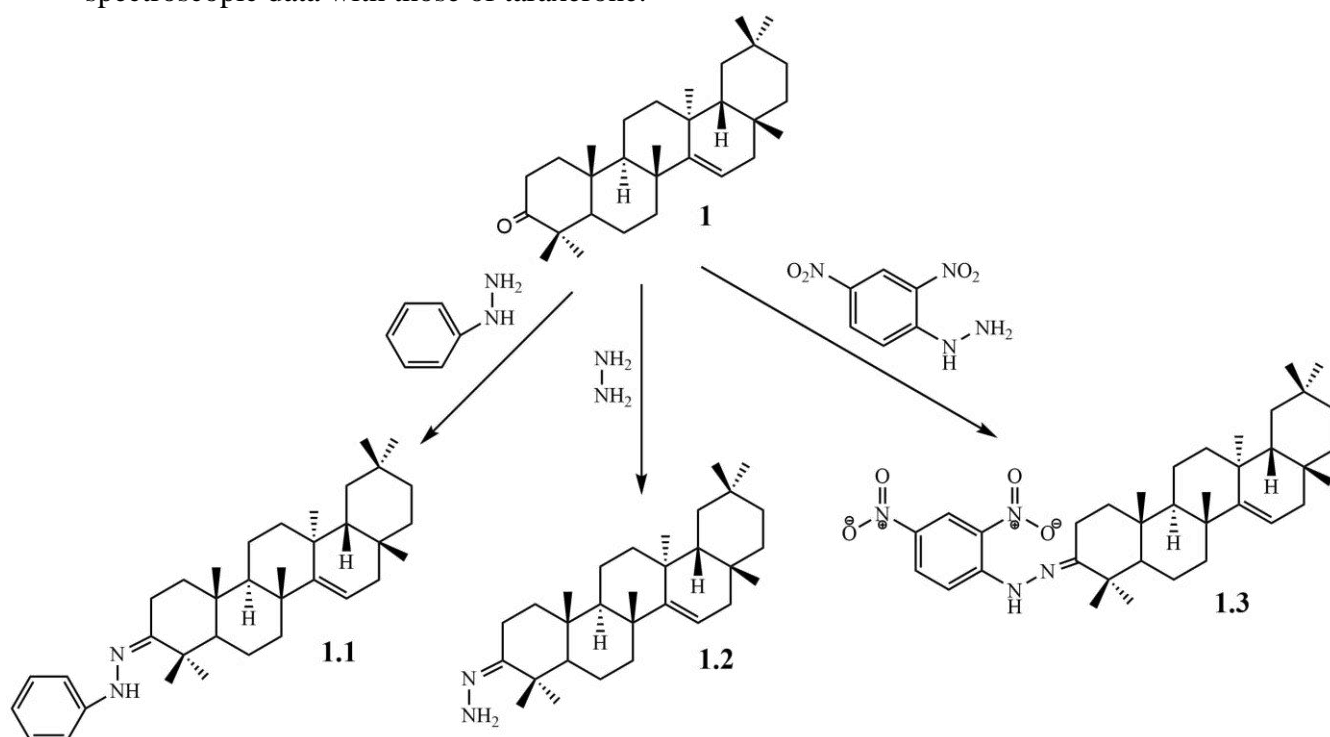
Position	DEPT	<b>11</b>
<b>1</b>	CH	193.77
<b>2</b>	CH	126.57
<b>3</b>	CH	153.24
<b>4</b>	C	126.80
<b>5</b>	CH	109.64
<b>6</b>	C	147.12
<b>7</b>	C	149.12
<b>8</b>	CH	115.10
<b>9</b>	CH	124.20
<b>10</b>	OCH <sub>3</sub>	56.15

Coniferaldehyde has demonstrated to possess anti-microbial activity against zoospores, chlamydospores and mycelia of *Phytophthora ramorum* (64); anti-inflammatory activity as an effective inducer of HO-1 (an intrinsic cell-protective enzyme) and exerting anti-inflammatory properties in response to LPS (endotoxin components of gram-negative bacterial cell walls)(65). It also showed anti-oxidant and anti-radical activities (66).

It also demonstrated promising results as promotor of neuronal differentiation, as it facilitated neurite growth by ERK activation in Neuro2a cells with neurite growth blocked by an ERK inhibitor (66).

### 1.6 Derivatization of taraxerone (1)

Taraxerone (1) was isolated in higher quantity. Therefore, taking advantage of the ketone group, reactions with hydrazine, phenylhydrazine and 2,4-dinitrophenylhydrazine were successfully performed, yielding imines **1.1-1.3** (Scheme 4). The structures of the taraxerone hydrazones (**1.1-1.3**) were established through NMR experiments and by comparison of their spectroscopic data with those of taraxerone.



**Scheme 4** - Preparation of taraxerone derivatives (1.1 – 1.3). Conditions: hydrazines (5 eq), MeOH:CH<sub>2</sub>Cl<sub>2</sub> (3:2), 200  $\mu$ L of acetic acid 10%. This reaction occurred overnight under reflux.

### 1.6.1 Structural identification of taraxerone derivatives

Compound **1.1** was obtained as a yellow amorphous solid with  $[\alpha]_{\text{D}}^{25} +225.9$ . The ESI-MS spectrum showed a protonated molecular ion at  $m/z$  513  $[\text{M}+\text{H}]^+$ , which together with the analysis of the NMR data were indicative of the molecular formula  $\text{C}_{36}\text{H}_{54}\text{N}_2$  (eleven degrees of unsaturation).

Compound **1.2** was obtained as a white amorphous solid with  $[\alpha]_{\text{D}}^{25} +19.1$ . The ESI-MS spectrum revealed a protonated molecular ion at  $m/z$  439  $[\text{M}+\text{H}]^+$ , substantiating the molecular formula  $\text{C}_{30}\text{H}_{50}\text{N}_2$  (six degrees of unsaturation).

Compound **1.3** was obtained as an orange amorphous solid with  $[\alpha]_{\text{D}}^{25} +13.3$ , and the ESI-MS spectrum showed a protonated molecular ion at  $m/z$  605  $[\text{M}+\text{H}]^+$ , which together with the analysis of the NMR data were indicative of the molecular formula  $\text{C}_{36}\text{H}_{52}\text{N}_4\text{O}_4$  (twelve degrees of unsaturation).

Through analysis of the IR spectrum, the presence of a secondary amine was suggested by a weak absorption band at  $3300\text{-}3700\text{ cm}^{-1}$ . The disappearance of the band characteristic of carbonyl groups was noticeable. The presence of an aromatic ring was also shown through bands at  $684/743\text{ cm}^{-1}$  in compounds **1.1** and **1.3**.

Besides the signals due to the different hydrazone moieties (**1.1** and **1.3**), the analysis of  $^1\text{H-NMR}$  and  $^{13}\text{C-NMR}$  spectra showed, as expected, very similar data regarding the triterpenic core (Tables 15 and 16). In fact, concerning the  $^{13}\text{C-NMR}$  spectra, the most remarkable difference was the absence of the carbonyl signal of taraxerone at  $\delta_{\text{C}} 217.70$  that was replaced by a carbon signal at higher field ( $\delta_{\text{C}} 148.31$  for **1.1**,  $\delta_{\text{C}} 151.11$  for **1.2** and  $\delta_{\text{C}} 167.31$  for **1.3**), which corroborated the substitution of the carbonyl moiety by the corresponding imine.



*Table 15* -  $^1\text{H-NMR}$  data of taraxerone derivatives **1.1**, **1.2** and **1.3** ( $\text{CDCl}_3$ , 300 MHz,  $\delta$  in ppm,  $J$  in Hz).

Position	<b>1.1</b>	<b>1.2</b>	<b>1.3</b>
<b>15</b>	5.54 dd ( $J = 8.1, 3.2$ )	5.55 dd ( $J = 8.3, 3.4$ )	5.56 dd ( $J = 8.2, 3.2$ )
<b>24</b>	1.09 m	1.04 s	1.07 s
<b>25</b>	1.07 s	1.11 s	1.18 s
<b>26</b>	0.95 s	0.91 s	0.90 s
<b>27</b>	1.14 s	1.21 s	1.15 s
<b>28</b>	0.84 s	0.83 s	0.83 s
<b>29</b>	0.96 s	0.95 s	0.95 s
<b>30</b>	0.92 s	0.89 s	0.92 s
<b>2'</b>	7.49 dd ( $J = 5.1, 2.0$ )		7.95 d ( $J = 9.6$ )
<b>3'</b>	7.49 dd ( $J = 5.1, 2.0$ )		8.29 dd ( $J = 9.6, 2.6$ )
<b>4'</b>	7.72 m		
<b>5'</b>	7.49 dd, ( $J = 5.1, 2.0$ )		9.12 d ( $J = 2.5$ )
<b>6'</b>	7.49 dd ( $J = 5.1, 2.0$ )		

*Table 16* -  $^{13}\text{C-NMR}$  data of taraxerone derivatives **1.1**, **1.2** and **1.3** ( $\text{CDCl}_3$ , 75 MHz,  $\delta$  in ppm).

Position	DEPT	<b>1.1</b>	<b>1.2</b>	<b>1.3</b>
<b>1</b>	CH <sub>2</sub>	38.27	38.25	38.00
<b>2</b>	CH <sub>2</sub>	33.93	27.80	28.54
<b>3</b>	C	151.11	148.31	167.31
<b>4</b>	C	42.50	41.80	42.49
<b>5</b>	CH	51.74	56.35	56.12
<b>6</b>	CH <sub>2</sub>	18.78	20.00	20,17
<b>7</b>	CH <sub>2</sub>	34.00	35.27	35.26
<b>8</b>	C	39.23	39.12	39.06
<b>9</b>	CH	48.86	48.95	48.83
<b>10</b>	C	35.30	35.93	35.94
<b>11</b>	CH <sub>2</sub>	17.61	17.60	17,63
<b>12</b>	CH <sub>2</sub>	37.88	38.08	37.85
<b>13</b>	C	37.71	37.86	37.68
<b>14</b>	C	158.15	158.08	157.77
<b>15</b>	CH	117.02	117.13	117.36
<b>16</b>	CH <sub>2</sub>	35.95	36.83	36.83
<b>17</b>	C	36.86	37.70	37.56
<b>18</b>	CH	48.95	49.07	48.95
<b>19</b>	CH <sub>2</sub>	41.25	41.11	40.91
<b>20</b>	C	26.22	28.95	28.94
<b>21</b>	CH <sub>2</sub>	33.28	25.84	33.71
<b>22</b>	CH <sub>2</sub>	30.10	33.25	33.23
<b>23</b>	CH <sub>3</sub>	23.16	23.67	25.78
<b>24</b>	CH <sub>3</sub>	21.43	21.47	20.65

**Table 16** -  $^{13}\text{C}$ -NMR data of taraxerone derivatives 1.1, 1.2 and 1.3 ( $\text{CDCl}_3$ , 75 MHz,  $\delta$  in ppm). (Cont.)

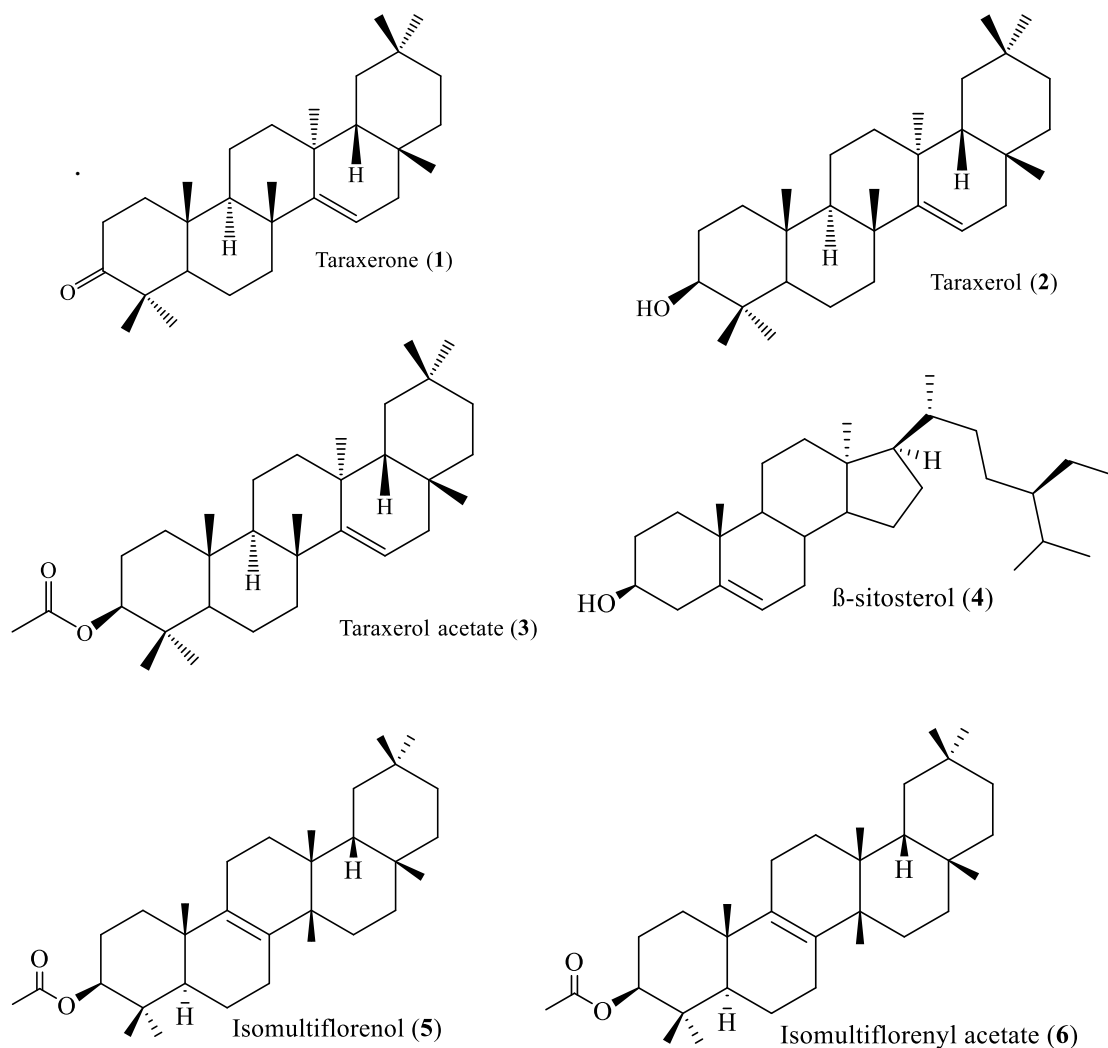
<b>25</b>	$\text{CH}_3$	15.70	15.06	14.79
<b>26</b>	$\text{CH}_3$	28.97	30.01	30.01
<b>27</b>	$\text{CH}_3$	22.42	17.91	24.25
<b>28</b>	$\text{CH}_3$	30.01	30.08	30.07
<b>29</b>	$\text{CH}_3$	33.51	33.51	33.50
<b>30</b>	$\text{CH}_3$	21.49	23.43	21.49
<b><math>\text{COCH}_3</math> (1) / 1'</b>	C	131.32		137.65
<b><math>\text{COCH}_3</math> (1) / 2'</b>	$\text{CH}_3$ (1) / CH (1.1, 1.3)	107.29		123.75
<b>3'</b>	CH	129.31		129.11
<b>4'</b>	CH (1.1) / C (1.3)	122.76		145.77
<b>5'</b>	CH	129.31		116.64
<b>6'</b>	CH (1.1) / C (1.3)	107.29		130.10

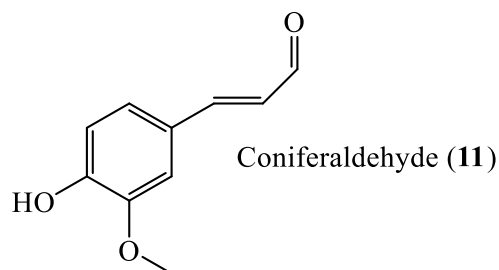
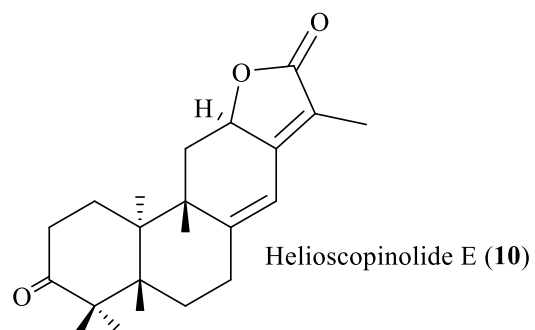
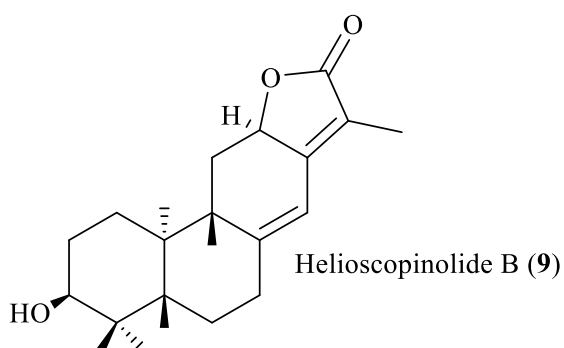
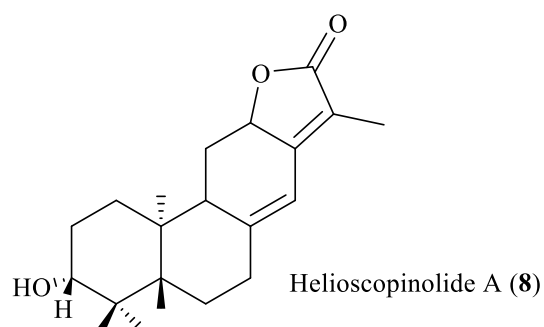
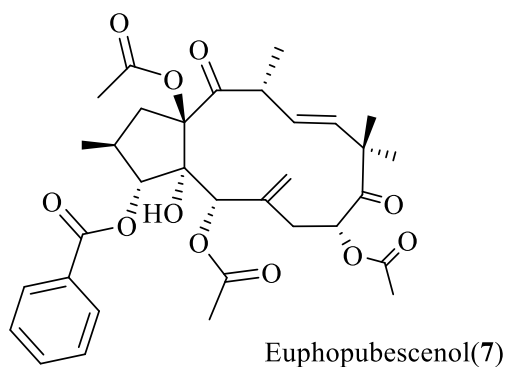
Several other reactions were attempted using different reagents (2-furoic hydrazide, 2-thiophenecarboxylic acid hydrazide, (4-aminobenzoyl) hydrazide, 4-chlorobenzhydrazide, 4-(trifluoromethyl) benzhydrazide and 9-indole-3-acetic hydrazide). However, due to the low yield obtained, it was not possible to perform reliable spectra in order to characterize the compounds, though perhaps in a larger scale better results could be obtained.

## Conclusion

The purpose of this study was to find out bioactive terpenes through isolation and molecular derivatization of compounds isolated in large amounts. After the phytochemical study of three fractions of *Euphorbia pubescens* methanol extract, eleven compounds were isolated through the use of chromatographic procedures, such as column and preparative thin layer chromatography.

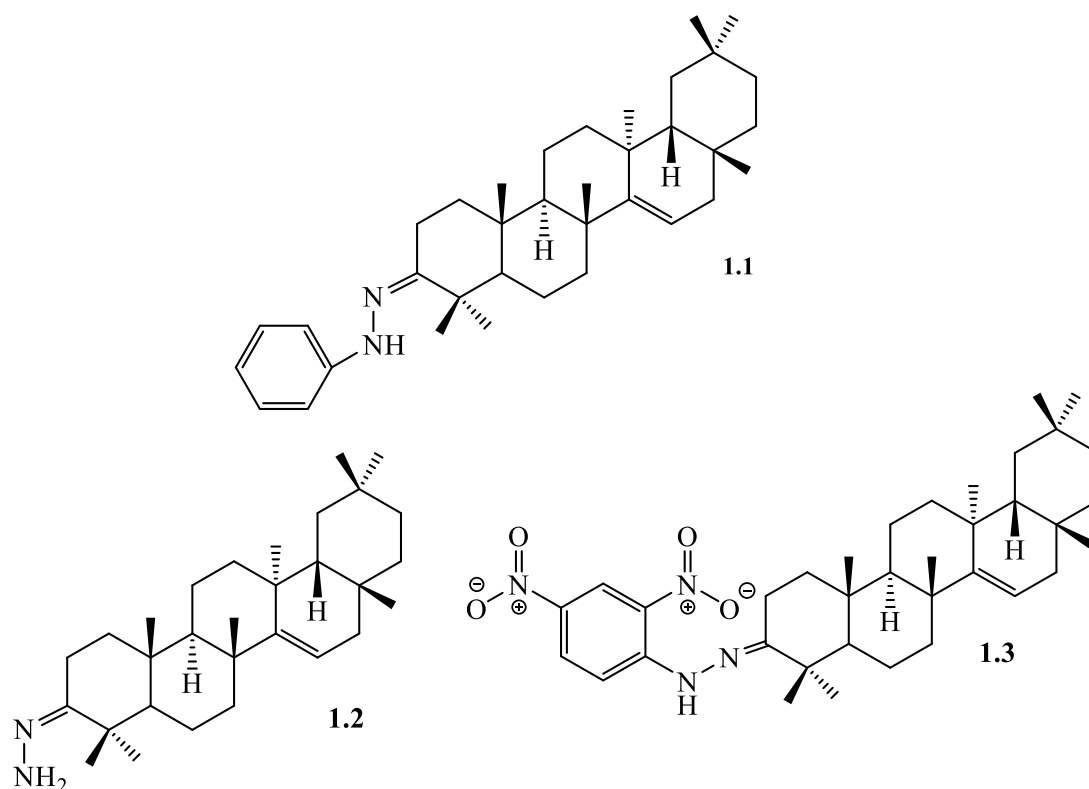
From the non-polar fractions A and B, five pentacyclic triterpenes were isolated, namely, taraxerone (1), taraxerol (2), taraxerol acetate (3), isomultiflorenol (5) and isomultiflorenol acetate (6). Moreover,  $\beta$ -sitosterol (4) was also isolated from these fractions. A macrocyclic diterpene polyester, euphobubescenol (7), three diterpenic lactones, helioscopinolide A, B and E (8, 9 and 10), and a phenylpropanoid derivative, coniferaldehyde (11) were isolated from the more polar fraction J.





The chemical structures of all compounds were ascertained through spectroscopic methods, including IR, MS,  $^1\text{H}$  and  $^{13}\text{C}$  NMR and 2D NMR (COSY, HMQC and HMBC experiments).

Taraxerone (1) was isolated in major amounts. Therefore, taking advantage of its carbonyl function, three hydrazine derivatives (1.1 – 1.3) were prepared through condensation with hydrazine, phenylhydrazine and 2,4-dinitrophenylhydrazine.



*Euphorbia* species have been used worldwide in traditional medicine in the treatment of ulcers, warts, skin affections and asthma. Therefore, in the last decades have been seen as an important source of bioactive compounds. Some of the compounds isolated in this work have shown promising results as anti-tumoral agents, including some that have demonstrated effectiveness as MDR modulators, as reported by other authors.

This work stands as a contribution for the phytochemical study of *Euphorbia pubescens*, corroborating the importance of natural products in the search of new drugs. Continuing research of this species should follow, as just some parts were studied, and more compounds could be isolated and derivatized, leading to promising bioactive compounds.

The isolated compounds and derivatives will be further tested as P-gp modulators on multidrug resistant cancer cells overexpressing this transporter.

## Experimental Section

### 1.7 General Procedures

All solvents were dried according to published methods and distilled prior to use. All the other reagents were obtained from commercial suppliers and were used without further purification.

Analytical thin layer chromatography (TLC) was performed on precoated silica-gel F254 plates (Merck 5554), visualized under UV light (254 and 366 nm) followed by spraying with H<sub>2</sub>SO<sub>4</sub>/MeOH (1:1) and finally heating the plate. Preparative TLC was carried out on 20 x 20 cm silica plates (Merck 5774). Column chromatography (CC) was performed over SiO<sub>2</sub> (Merck 9385) and over reverse-phase C18 silica (YMC-GEL ODS-A 12nm S-50µm AA12S50). CombiFlash® Rf 200 (Teledyne Isco) with SiO<sub>2</sub> or reverse-phase C18 prepacked columns were also used.

The NMR spectra were collected on a Bruker Fourier 300 spectrometer (<sup>1</sup>H 300 MHz, <sup>13</sup>C 75 MHz) using CDCl<sub>3</sub> as solvent. <sup>1</sup>H and <sup>13</sup>C chemical shifts are expressed in δ (ppm) and the proton coupling constants (*J*) in hertz (Hz).

Melting points were determined on a Köpffler apparatus. Specific optical rotations were measured using a Jasco P-2000 digital polarimeter with a 1 dm long quartz cell and the samples were prepared in CHCl<sub>3</sub>.

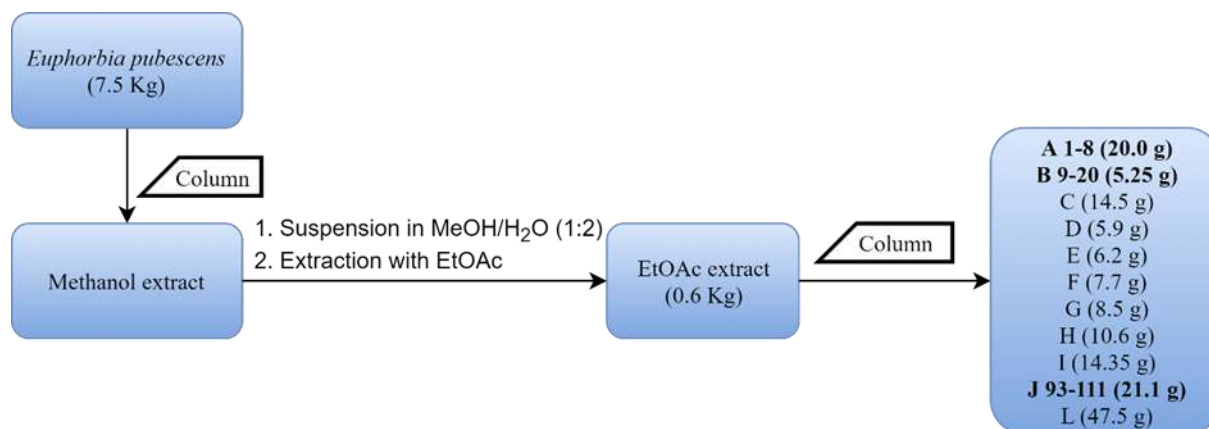
Infrared (IR) spectra were collected on an Affinity-1 (Shimadzu) FTIR spectrophotometer in KBr pellets with only the most relevant absorption bands being reported.

Low resolution mass spectrometry (ESI-MS) was performed on a triple quadrupole (QT) mass spectrometer (Micromass Quattro Micro API, Waters) with ionization electrospray (ESI), operating in positive mode.

### 1.8 Phytochemical Study of *Euphorbia pubescens*

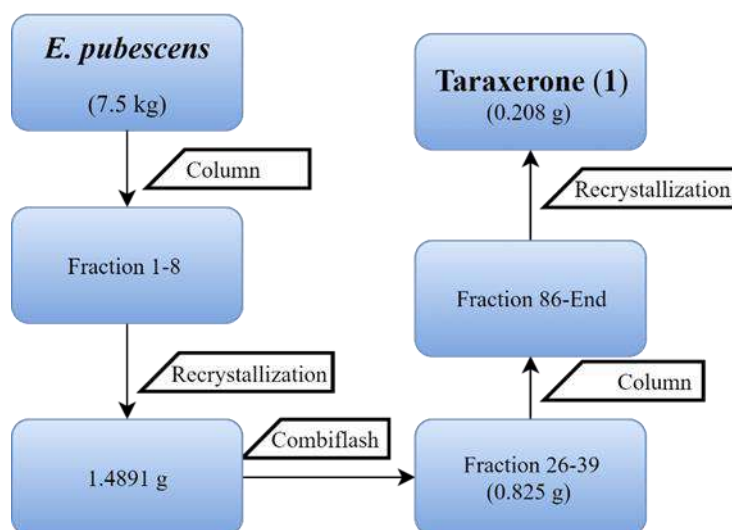
*Euphorbia pubescens* Spreng. (Euphorbiaceae) was harvested in July 2015, in Sesimbra, Portugal, and identified by Dr. Teresa Vasconcelos of Instituto Superior de Agronomia, Universidade de Lisboa.

In Scheme 5 is depicted the process of extraction and fractionation of the plant, both of which were conducted by João Gomes (13). From this process eleven crude fractions were obtained (A to L) in accordance to the corresponding TLC that showed different compositions. In this work, just fractions A (1-8), B (9-20) and J (93-111) were studied.



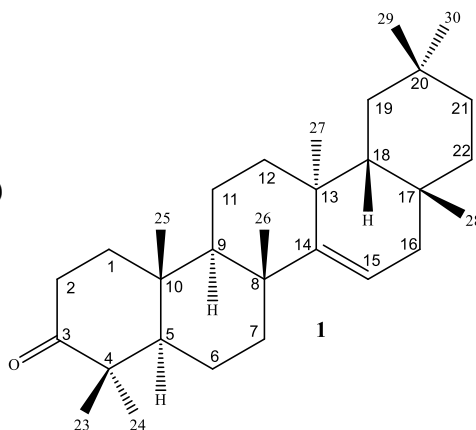
Scheme 5 - Extraction and fractionation of *E. pubescens*.

### 1.8.1 Study of fraction A (1-8) and B (9-20)



Scheme 6 - Study of fraction A (1-8).

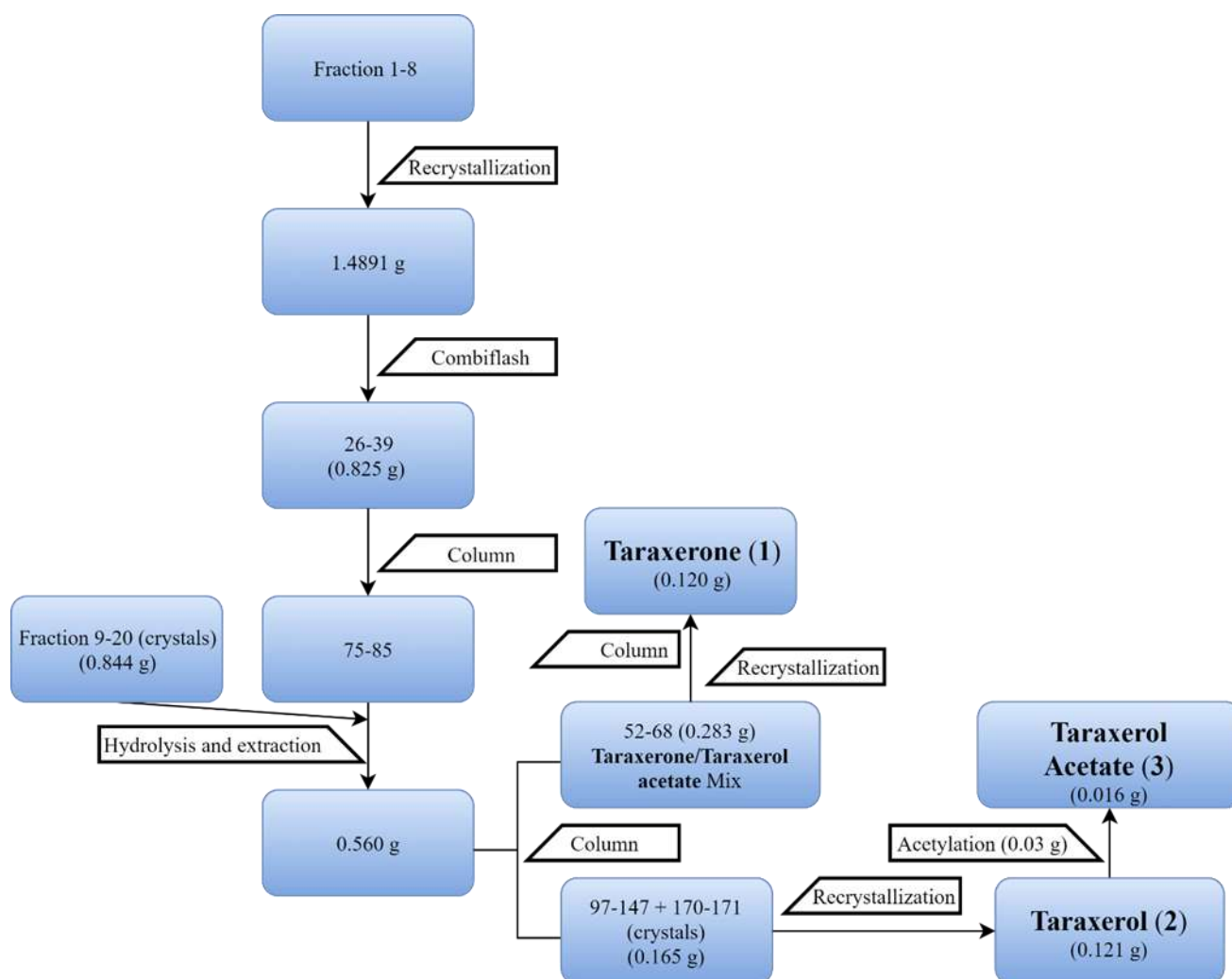
Fraction A (1-8) (20.0 g, eluted with *n*-hexane/EtOAc 1:0 to *n*-hexane/EtOAc 8:2) was recrystallized from acetone to give 1.25 g of white crystals. These crystals were then submitted to a combiflash chromatography, with the most interesting fraction being 26-39 (0.285 g) that was further separated through column chromatography (*n*-hexane/EtOAc 1:0 to 0:1), obtaining fraction 86-End, the more interesting sub-fraction. By recrystallization of this sub-fraction, 0.208 g of compound **1** (taraxerone) was obtained (Scheme 6).

**Taraxerone (1)****White crystals****m.p.** 141-143 °C (Lit. 141 – 142°C)(39) $[\alpha]_D^{25} +50.0$  (CHCl<sub>3</sub>) **$\nu_{\max}$ cm<sup>-1</sup> (KBr):** 2958 cm<sup>-1</sup>, 2885 cm<sup>-1</sup>,  
2368 cm<sup>-1</sup>, 1708 cm<sup>-1</sup> **$m/z$**  425 [M+H]<sup>+</sup>

**<sup>1</sup>H NMR ( 300 MHz, CDCl<sub>3</sub>):**  $\delta$  5.56 (1H, *dd*,  $J = 8.2, 3.3$  Hz, H-15), 1.14 (3H, *s*, H-27), 1.08 (3H, *s*, H-23), 1.08 (3H, *s*, H-25), 1.07 (3H, *s*, H-24), 0.95 (3H, *s*, H-29), 0.91 (6H, *s*, H-26, H-30), 0.83 (3H, *s*, H-28).

**<sup>13</sup>C NMR ( 75 MHz, CDCl<sub>3</sub>):**  $\delta$  217.70 (C-3), 157.77 (C-14), 117.35 (C-15), 55.94 (C-5), 48.96 (C-18), 48.86 (C-9), 47.73 (C-4), 40.80 (C-19), 39.03 (C-8), 38.51 (C-1), 37.90 (C-12), 37.85 (C-13), 37.70 (C-17), 36.83 (C-16), 35.93 (C-10), 35.26 (C-7), 34.29 (C-2), 33.73 (C-21), 33.51 (C-29), 33.24 (C-22), 30.07 (C-28), 30.01 (C-26), 30.01 (C-26), 28.95 (C-20), 26.27 (C-23), 25.72 (C-27), 21.63 (C-30), 21.50 (C-24), 20.12 (C-6), 17.60 (C-11), 14.96 (C-25).





*Scheme 7 - Partial workup of fraction 1-8 and 9-20(crystals).*

Another sub-fraction of fraction A (1-8), sub-fraction 26-39 (0.82 g), was submitted to column chromatography (*n*-hexane/EtOAc 1:0 to 0:1), with the most interesting fraction of this column being 75-85. To fraction 75-85 were added the crystals of fraction B (9-20) (0.84 g – obtained by recrystallization with acetone). The mixture was submitted to hydrolysis (5% KOH in methanol, 25 mL, stirring at room temperature for 6 hours), after which it was diluted with water and exhaustively extracted with EtOAc. The organic layers were dried with anhydrous sodium sulphate, evaporated and the corresponding fraction (0.560g) was submitted to column chromatography eluted with mixtures of *n*-hexane/EtOAc of increasing polarity. Several fractions were obtained, the most important being 52-68 (0.263 g), from which 0.120 g of compound **1** was isolated, and fractions 97-147 that was added to the crystals of 170-171, and recrystallized, yielding 0.121 g of compound **2** (Taraxerol).

Compound **2** (0.03 g) was acetylated to yield 0.016 g of compound **3** (Taraxerol Acetate), which was already present in some fractions, usually with taraxerone, and showed itself difficult to separate (Scheme 7).

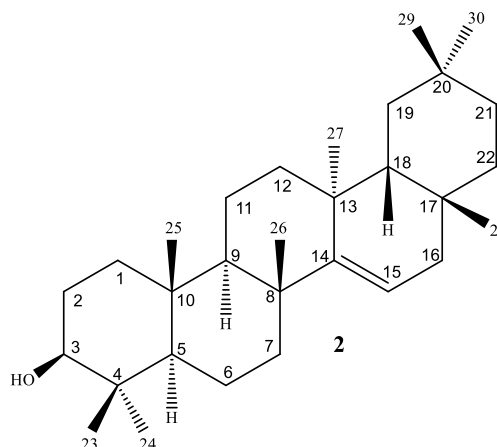
### Taraxerol (**2**)

Amorphous white powder

$$[\alpha]_D^{25} +79.8 \text{ (CHCl}_3\text{)}$$

$$\nu_{\max} \text{cm}^{-1} \text{ (KBr): } 3487 \text{ cm}^{-1}$$

$$m/z \text{ 409 [M+H-H}_2\text{O]}^+$$



$^1\text{H NMR}$  ( 300 MHz,  $\text{CDCl}_3$ ):  $\delta$  5.53 (1H, *dd*,  $J = 8.2, 3.3$  Hz, H-15), 3.19 (3H, *dd*, ( $J = 15.6, 5.4$ , H-3), 1.09 (3H, *s*, H-27), 0.98 (3H, *s*, H-23), 0.95 (3H, *s*, H-29), 0.93 (3H, *s*, H-24), 0.91 (6H, *s*, H-26, H-30), 0.82 (3H, *s*, H-28), 0.80 (3H, *s*, H-25).

$^{13}\text{C NMR}$  ( 75 MHz,  $\text{CDCl}_3$ ):  $\delta$  158.25 (C-14), 117.03 (C-15), 79.23 (C-3), 55.69 (C-5), 49.44 (C-18), 48.92 (C-9), 41.49 (C-19), 39.14 (C-8), 38.92 (C-4), 38.16 (C-1), 37.90 (C-10), 37.87 (C-13), 37.73 (C-17), 36.84 (C-16), 35.94 (C-12), 35.28 (C-7), 33.86 (C-21), 33.51 (C-29), 33.26 (C-22), 30.08 (C-26), 29.98 (C-28), 28.95 (C-20), 28.15 (C-23), 27.31 (C-2), 26.06 (C-27), 21.47 (C-30), 18.95 (C-6), 17.66 (C-11), 15.60 (C-25), 15.58 (C-24).

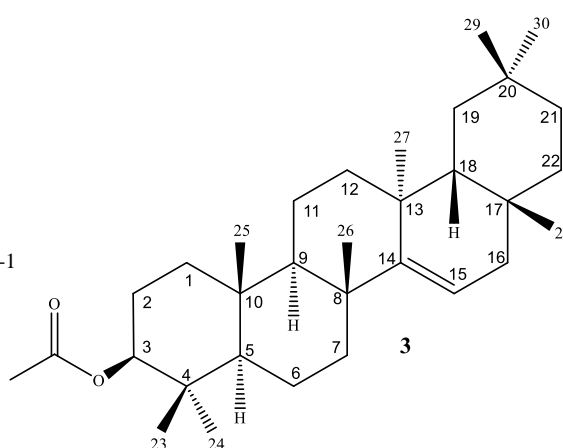
### Taraxerol acetate (**3**)

Amorphous white powder

$$[\alpha]_D^{25} +55.5 \text{ (CHCl}_3\text{)}$$

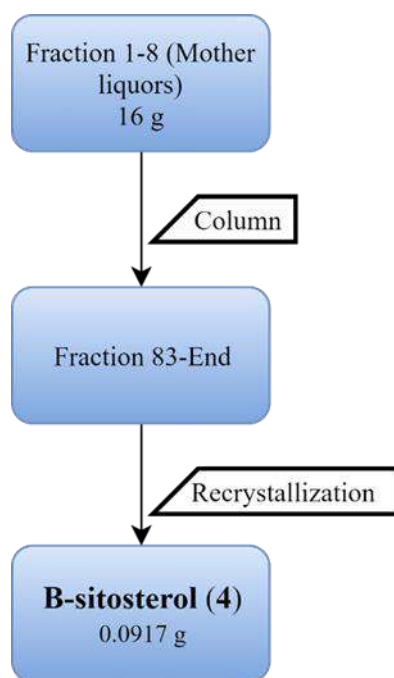
$$\nu_{\max} \text{cm}^{-1} \text{ (KBr): } 1724 \text{ cm}^{-1}, 1252 \text{ cm}^{-1}$$

$$m/z \text{ 409 [M+H-C}_2\text{H}_4\text{O}_2]^+$$



$^1\text{H NMR}$  ( 300 MHz,  $\text{CDCl}_3$ ):  $\delta$  5.53 (1H, *dd*,  $J = 8.1, 3.2$  Hz, H-15), 4.46 (3H, *dd*,  $J = 9.8, 6.2$ , H-3), 2.04 (3H, *s*,  $\text{COCH}_3$ ), 1.09 (3H, *s*, H-26), 0.95 (6H, *s*, H-23, H-29), 0.90 (6H, *s*, H-25, H-30), 0.87 (3H, *s*, H-27), 0.86 (3H, *s*, H-28), 0.82 (3H, *s*, H-24).

$^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  171.12 ( $\text{COCH}_3$ ), 158.14 (C-14), 117.09 (C-15), 81.17 (C-3), 55.80 (C-5), 49.35 (C-18), 48.93 (C-9), 41.38 (C-19), 39.14 (C-4), 38.05 (C-8), 37.84 (C-1/C-10), 37.71 (C-17), 37.55 (C-13), 36.82 (C-16), 35.93 (C-12), 35.27 (C-7), 33.83 (C-21), 33.49 (C-29), 33.25 (C-22), 31.06 (C-26), 30.07 (C-28), 29.98 (C-20), 28.95 (C-2), 28.13 (C-23), 26.07 (C-27), 23.61 (C-30), 21.43 ( $\text{COCH}_3$ ), 18.84 (C-6), 17.66 (C-11), 16.73 (C-25), 15.64 (C-24).



*Scheme 8 - Workup of fraction 83-End of fraction 1-8 (Mother liquors).*

After the recrystallization of fraction 1-8, the mother liquors were also studied and submitted to column chromatography (*n*-hexane/EtOAc 1:0 to 0:1), with the fractions 21 and 83-End being of interest and recrystallized. This led to the isolation of 0.0917 g of compound **4** ( $\beta$ -sitosterol) from sub-fraction 83-End (Scheme 8).

### $\beta$ -sitosterol (**4**)

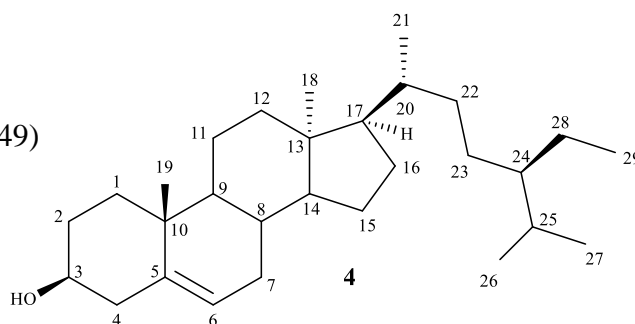
White powder

m.p. 130-131 °C (Lit. 128 – 130°C)(49)

$[\alpha]_{\text{D}}^{25}$  -15.9 ( $\text{CHCl}_3$ )

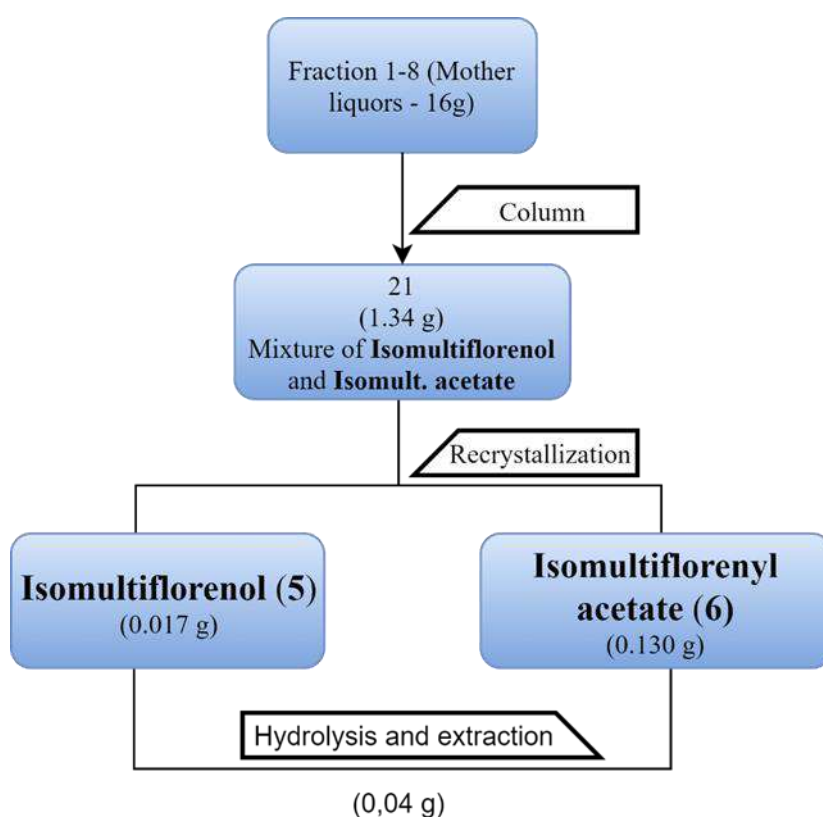
$\nu_{\text{max}}\text{cm}^{-1}$  (KBr): 3425  $\text{cm}^{-1}$

$m/z$  397  $[\text{M}+\text{H}-\text{H}_2\text{O}]^+$



$^1\text{H NMR}$  ( 300 MHz,  $\text{CDCl}_3$ ): $\delta$  5.35 (1H, *d*,  $J = 5.4$  Hz, H-6), 3.52 (1H, *m*, H-3), 1.01 (3H, *s*, H-19), 0.92 (3H, *d*,  $J = 6.5$  Hz, H-21), 0.86 (3H, *t*,  $J = 13.6, 7.2$  Hz, H-29), 0.83 (3H, *d*,  $J = 6.5$  Hz, H-27), 0.80 (3H, *d*,  $J = 6.6$  Hz, H-26), 0.68 (3H, *s*, H-18).

$^{13}\text{C NMR}$  ( 75 MHz,  $\text{CDCl}_3$ ): $\delta$  149.93 (C-5), 121.67 (C-6), 71.77 (C-3), 56.73 (C-14), 56.23 (C-17), 50.31 (C-9), 46.02 (C-22), 42.49 (C-13), 42.27 (C-4), 39.79 (C-12), 37.42 (C-1), 36.67 (C-10), 36.30 (C-18), 34.12 (C-20), 32.08 (C-7), 31.85 (C-2), 31.64 (C-8), 29.13 (C-25), 28.20 (C-16), 26.26 (C-15), 26.06 (C-21), 23.24 (C-23), 21.04 (C-11), 19.76 (C-26), 19.35 (C-27), 18.99 (C-19), 18.73 (C-28), 11.93 (C-24), 11.81 (C-29).



*Scheme 9 - Workup of fraction 21 of fraction 1-8 (Mother liquors).*

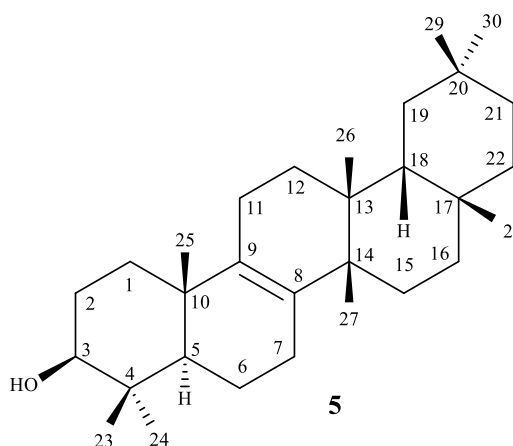
Fraction 21 (1.34 g) was also recrystallized (EtOAc/*n*-hexane) to obtain 130 mg of compound **5** and a mixture of compounds **5** (isomultiflorenol) and **6** (isomultiflorenyl acetate). As compound **6** was not able to be isolated for a correct characterization, 40 mg of compound **5** were submitted to hydrolysis (25mL MeOH /KOH 5%, stirring at room temperature for 6 hours). The reaction mixture was suspended in  $\text{H}_2\text{O}$  and extracted several times with EtOAc, yielding 0.017 g of isolated compound **6** (Scheme 9).

**Isomultiflorenol (5)****Amorphous white powder**

$$[\alpha]_{\text{D}}^{25} +88.3 \text{ (CHCl}_3\text{)}$$

$$\nu_{\text{max}}\text{cm}^{-1} \text{ (KBr): } 3410 \text{ cm}^{-1}, \\ 1458 \text{ cm}^{-1}, 1379 \text{ cm}^{-1}$$

$$m/z \text{ 409 [M+H-H}_2\text{O]}^+$$



$^1\text{H NMR}$  ( 300 MHz,  $\text{CDCl}_3$ ):  $\delta$  3.23 (1H, *dd*,  $J = 11.4, 4.8$  Hz, H-3), 1.07 (3H, *s*, H-28), 1.06 (3H, *s*, H-26), 1.00 (3H, *s*, H-30), 0.99 (3H, *s*, H-29), 0.97 (3H, *s*, H-27), 0.96 (6H, *s*, H-25, H-24), 0.80 (3H, *s*, H-23).

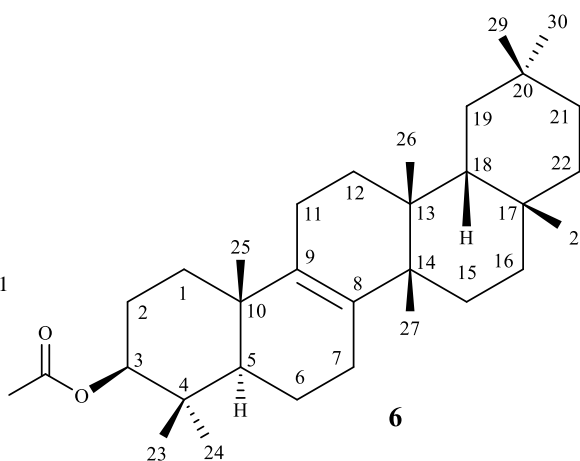
$^{13}\text{C NMR}$  ( 75 MHz,  $\text{CDCl}_3$ ):  $\delta$  135.24 (C-8), 133.67 (C-9), 79.18 (C-3), 50.86 (C-5), 44.27 (C-18), 41.14 (C-14), 38.96 (C-4), 37.77 (C-10), 37.44 (C-13), 36.96 (C-22), 36.85 (C-16), 35.20 (C-21), 34.66 (C-29), 34.55 (C-19), 34.34 (C-1), 33.14 (C-30), 31.65 (C-28), 30.95 (C-17), 30.84 (C-12), 28.48 (C-20), 28.20 (C-23), 28.08 (C-7), 27.65 (C-15), 26.41 (C-27), 24.84 (C-2), 20.96 (C-11), 19.97 (C-25), 19.39 (C-6), 19.12 (C-26), 15.76 (C-24).

**Isomultiflorenyl acetate (6)****Amorphous white powder**

$$[\alpha]_{\text{D}}^{25} +98.8 \text{ (CHCl}_3\text{)}$$

$$\nu_{\text{max}}\text{cm}^{-1} \text{ (KBr): } 1726 \text{ cm}^{-1}, 1250 \text{ cm}^{-1}$$

$$m/z \text{ 409 [M+H-C}_2\text{H}_4\text{O}_2]^+$$

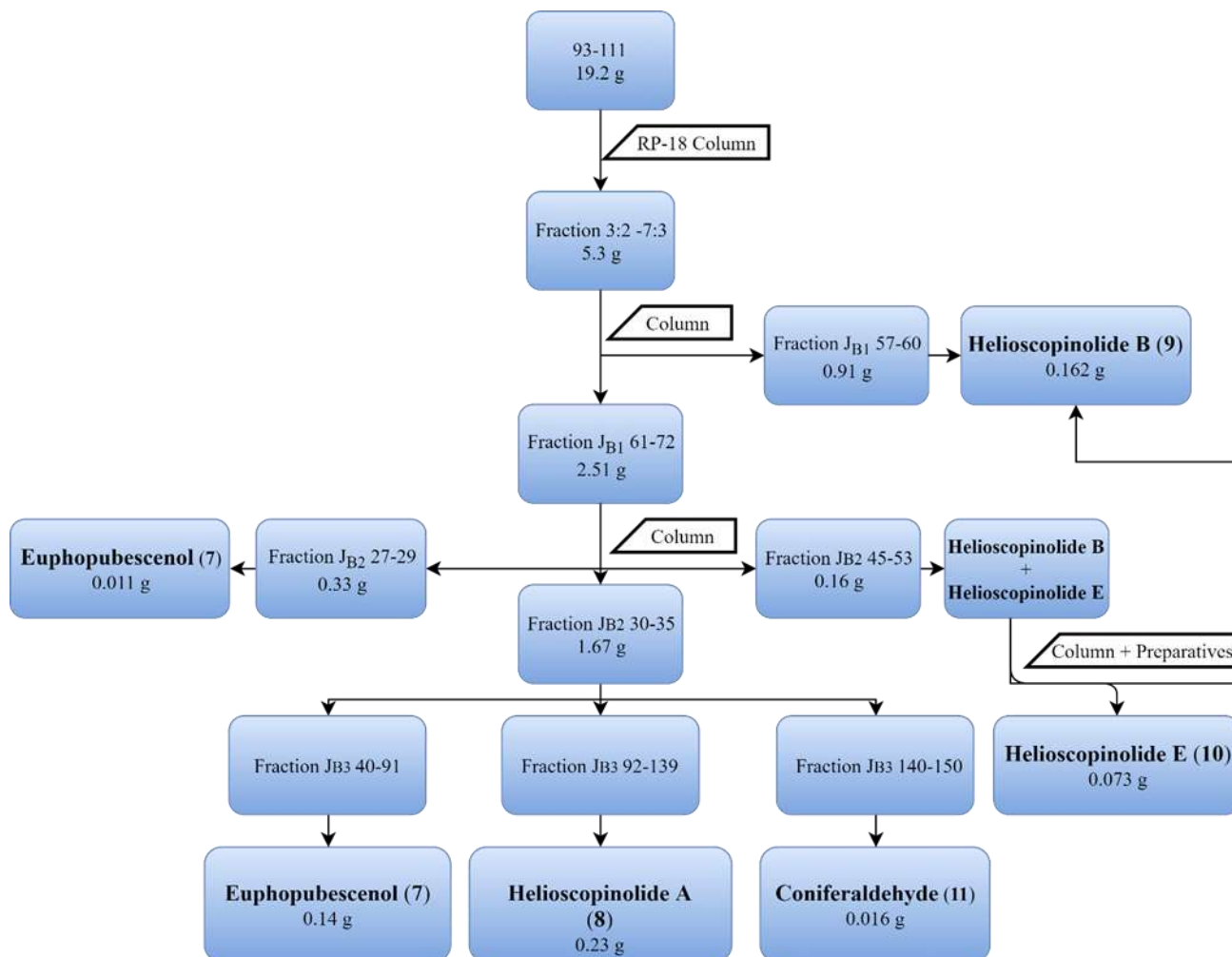


$^1\text{H NMR}$  ( 300 MHz,  $\text{CDCl}_3$ ):  $\delta$  4.48 (1H, *dd*,  $J = 11.3, 6.5$  Hz, H-3), 2.05 (3H, *s*,  $\text{COCH}_3$ ), 1.07 (3H, *s*, H-28), 1.05 (3H, *s*, H-27), 0.97 (9H, *s*, H-25, H-26, H-30), 0.95 (3H, *s*, H-29), 0.88 (3H, *s*, H-24), 0.87 (3H, *s*, H-23).

$^{13}\text{C NMR}$  ( 75 MHz,  $\text{CDCl}_3$ ):  $\delta$  171.25 ( $\text{COCH}_3$ ), 135.23 (C-8), 133.45 (C-9), 81.10 (C-3), 50.87 (C-5), 44.23 (C-18), 43.11 (C-21), 41.11 (C-14), 37.84 (C-4), 37.56 (C-10), 37.35 (C-13), 36.83 (C-16), 36.81 (C-22), 34.76 (C-1), 34.45 (C-29), 34.31 (C-19), 33.07 (C-30), 31.63 (C-28), 30.90 (C-17), 30.74 (C-12), 28.45 (C-20), 28.11 (C-24), 27.46 (C-7), 26.30 (C-

15), 24.74 (C-27), 24.32 (C-2), 21.49 ( $\text{COCH}_3$ ), 20.94 (C-11), 20.00 (C-25), 19.20 (C-6), 19.04 (C-26), 16.83 (C-23).

### 1.8.2 Study of fraction J (93-111)



*Scheme 10 - Workup of fraction J (93-111).*

Fraction J (93-111) (19.2g, eluted with *n*-hexane/EtOAc 7:3) went through reverse-phase chromatography (MeOH/H<sub>2</sub>O 1:1 to MeOH/H<sub>2</sub>O 1:0 in increasing gradients of 10%). After TLC monitoring, the fractions with similar profile were combined (Scheme 10).

From this, fractions eluted with mixtures of MeOH/H<sub>2</sub>O 3:2 – 7:3 were chosen to be studied and associated.

The referred fraction (5.3 g) was submitted to CC using *n*-hexane/EtOAc (1:0 up to 4:1). According to differences in composition as indicated by TLC, seven crude fractions were obtained (Fractions J<sub>B</sub>).

Fraction J<sub>B1</sub> (57-60) (0.91 g) was separated on a CombiFlash® Rf System (24 g, flash column, RediSepRf) with *n*-hexane-EtOAc (1:0 to 0:1, used in increasing gradients of 5%; 5 ml/min). To obtain helioscopinolide B (**9**), a subfraction was recrystallized with EtOAc/*n*-hexane yielding 115 mg of pure compound.

Fraction J<sub>B1</sub> (61-72) (2.51 g, *n*-hexane/EtOAc 3:2) was chromatographed on a Combiflash system (120 g, flash column, RediSepRf) with *n*-hexane-EtOAc (1:0 to 0:1, used in increasing gradients of 5%; 7 ml/min) and yielded three interesting fractions: J<sub>B2</sub> 27-29 (0.33 g), J<sub>B2</sub> 30-35 (1.67 g) and J<sub>B2</sub> 45-53 (0.16 g).

Fraction J<sub>B2</sub> 27-29 was submitted to a chromatographic column using 100g of SiO<sub>2</sub> and eluted with mixtures of CH<sub>2</sub>Cl<sub>2</sub>-MeOH of increasing polarity to give, 11 mg of compound **7** (euphobubescenol). Fraction J<sub>B2</sub> 30-35 was submitted to column chromatography (CH<sub>2</sub>Cl<sub>2</sub>/acetone 1:0 to 4:1), yielding three interesting fractions, J<sub>B3</sub> 40-91, J<sub>B3</sub> 92-139 and J<sub>B3</sub> 140-150. Fraction J<sub>B3</sub> 40-91 was again submitted to CC with similar eluent, yielding 0.14 g of the already isolated compound **7** (euphobubescenol). Fraction J<sub>B3</sub> 92-139 was recrystallized to give 0.23 g of helioscopinolide A (**8**). Fraction J<sub>B3</sub> 140-150 was further purified by preparative chromatography yielding 16 mg of coniferaldehyde (**11**).

Fraction J<sub>B2</sub> 45-53(0.16 g) was submitted to column chromatography (SiO<sub>2</sub>), using as eluents mixtures of hexane/EtOAc (1:0 to 0:1), and a mixture of helioscopinolide B (**9**) and E (**10**) was collected. This mixture was then submitted to preparative chromatography with the successful isolation of 73 mg of helioscopinolide E and 47 mg of helioscopinolide B.

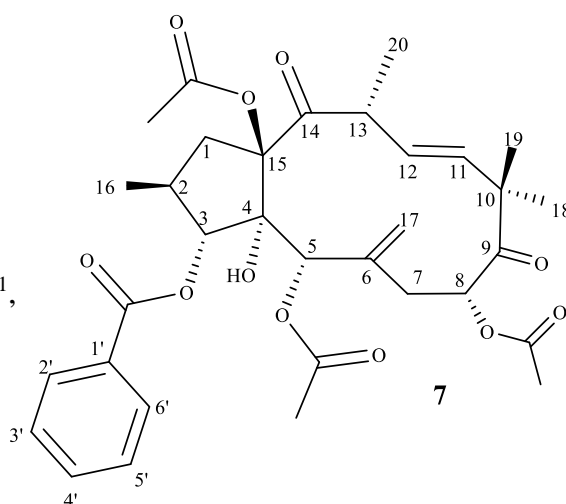
### Euphobubescenol (**7**)

#### Amorphous white powder

$$[\alpha]_D^{25} +74.6 \text{ (CHCl}_3\text{)}$$

$$\nu_{\max} \text{cm}^{-1} \text{ (KBr): } 3520 \text{ cm}^{-1}, 1748 \text{ cm}^{-1}, \\ 1715 \text{ cm}^{-1}, 1633 \text{ cm}^{-1}$$

$$m/z \text{ 553 [M+H-CH}_3\text{COOH]}^+$$



<sup>1</sup>H NMR ( 300 MHz, CDCl<sub>3</sub>): δ 7.97 (2H, *dd*, *J* = 8.0, 1.5 Hz, Bz (H-2',H-6')), 7.57 (1H, *tt*, *J* = 7.5, 1.5 Hz, Bz (H-4')), 7.42 (2H, *t*, *J* = 7.8 Hz, Bz (H-3',H-5')), 5.92 (1H, *dd*, *J* = 16.0, 9.0 Hz, H-12), 5.89 (1H, *d*, *J* = 9.0 Hz, H-11) 5.67 (1H, *d*, *J* = 8.9 Hz, H-8), 5.61 (1H, *d*,

$J = 10.5$  Hz, H-3), 5.12 (2H, *s*, H-7), 4.86 (1H, *s*, H-5), 3.85 (1H, *m*, H-13), 2.91 (1H, *tdd*,  $J = 10.4, 7.0, 3.5$  Hz, H-2), 2.43 (2H, *m*, H-1 $\alpha$  /1 $\beta$ ), 2.16 (3H, *s*, 15-O(COCH<sub>3</sub>)), 2.08 (3H, *s*, 5-O(COCH<sub>3</sub>)), 2.00 (1H, *s*, H-7 $\beta$ ), 1.91 (3H, *s*, 8-O(COCH<sub>3</sub>)), 1.71 (1H, *dd*,  $J = 14.9; 9.2$ , H7- $\alpha$ ), 1.46 (3H, *s*, H-19), 1.38 (3H, *d*,  $J = 6.5$  Hz, H-20), 1.24 (3H, *d*,  $J = 6.9$  Hz, H-16), 1.22 (3H, *s*, H-18)).

<sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  208.20 (C-9), 204.61 (C-14), 169.87 (8-O (COCH<sub>3</sub>)), 168.86 (15-O (COCH<sub>3</sub>)), 168.42 (5-O (COCH<sub>3</sub>)), 166.60 (3-OBz), 140.17 (C-6), 136.91 (C-11), 133.70 (Bz (4')), 132.84 (C-12), 130.19 (Bz (2', 6')), 129.09 (Bz (1')), 128.53 (Bz (3', 5')), 116.89 (C-17), 94.62 (C-15), 82.94 (C-4), 77.89 (C-3), 73.28 (C-5), 72.41 (C-8), 50.66 (C-10), 46.91 (C-13), 38.82 (C-1), 36.90 (C-2), 35.97 (C-7), 24.57 (C-18, C-19), 21.14 (15-O (COCH<sub>3</sub>)), 20.71 (5-O (COCH<sub>3</sub>)), 20.42 (8-O (COCH<sub>3</sub>)), 20.16 (C-20), 17.79 (C-16).

### Helioscopinolide A (8)

White crystals

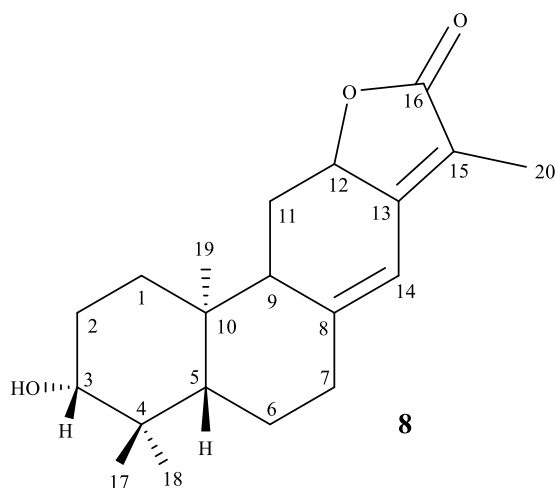
**m.p.** 204-206 °C (Lit. 205 °C) (13)

$[\alpha]_D^{25} +587.4$  (CHCl<sub>3</sub>)

$\nu_{\max}$  cm<sup>-1</sup> (KBr): 3478 cm<sup>-1</sup>, 1734 cm<sup>-1</sup>,

1398 cm<sup>-1</sup>

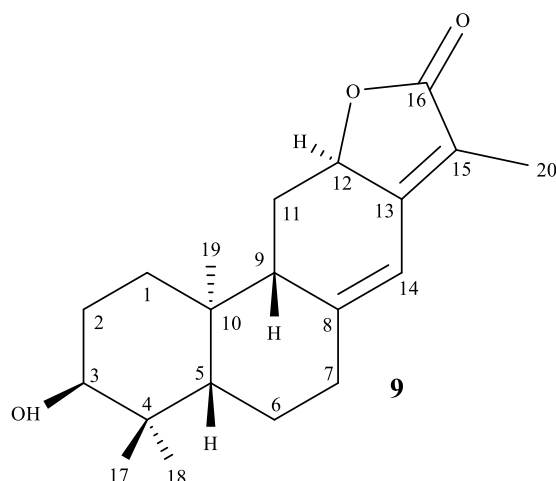
$m/z$  317 [M+H]<sup>+</sup>



<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  6.27 (1H, *s*, H-14), 4.85 (1H, *ddd*,  $J = 13.6, 6.2, 1.8$  Hz, H-12), 3.27 (1H, *dd*,  $J = 11.6, 4.4$  Hz, H-3), 2.53 (2H, *m*, H-11), 2.18 (1H, *dd*,  $J = 15.1, 9.0$  Hz, H-9), 1.94 (2H, *dt*,  $J = 13.1, 3.6$  Hz, H-1), 1.81 (3H, *d*,  $J = 1.7$  Hz, H-20), 1.64 (2H, *m*, H-2), 1.14 (1H, *dd*,  $J = 12.5, 2.5$  Hz, H-5), 1.02 (3H, *s*, H-17), 0.92 (3H, *s*, H-19), 0.81 (3H, *s*, H-18).

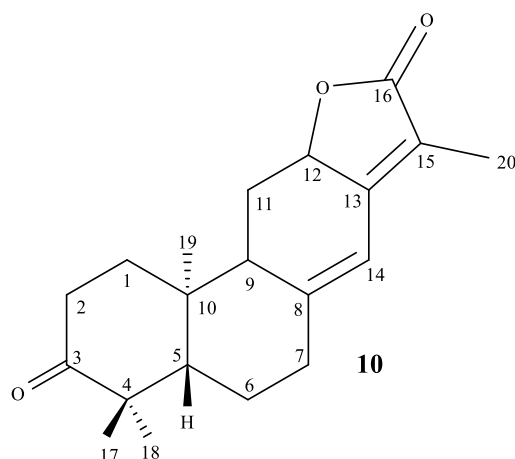
<sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  175.39 (C-16), 156.17 (C-13), 151.58 (C-8), 116.61 (C-15), 114.31 (C-14), 78.66 (C-3), 76.00 (C-12), 54.48 (C-5), 51.67 (C-9), 41.34 (C-10), 39.21 (C-4), 37.53 (C-1), 37.04 (C-7), 28.79 (C-17), 27.70 (C-2), 27.66 (C-11), 23.56 (C-6), 16.82 (C-19), 15.71 (C-18), 8.37 (C-20).



**Helioscopinolide B (9)****Colorless oil** $[\alpha]_{\text{D}}^{25} +417.6$  (CHCl<sub>3</sub>) $\nu_{\text{max}}\text{cm}^{-1}$  (KBr): 3467 cm<sup>-1</sup>, 1730 cm<sup>-1</sup>,  
1393 cm<sup>-1</sup> $m/z$  317 [M+H]<sup>+</sup>

<sup>1</sup>H NMR ( 300 MHz, CDCl<sub>3</sub>): 6.25 (1H, *s*, H-14), 4.86 (1H, *ddd*, *J* = 13.3, 6.1, 1.2, H-12), 3.46 (1H, *d*, *J* = 3.2, H-3), 2.56 (1H, *m*, H-11), 2.47 (1H, *dd*, *J* = 4.2, 2.3, H-7), 2.29 (1H, *d*, *J* = 8.7, H-9), 1.81 (3H, *d*, *J* = 1.3, H-20), 1.60-1.70 (4H, *m*, H-1, H-2), 1.64 (1H, *m*, H-5), 1.41 (2H, *m*, H-6), 0.98 (3H, *s*, H-17), 0.93 (3H, *s*, H-19), 0.86 (3H, *s*, H-18).

<sup>13</sup>C NMR ( 75 MHz, CDCl<sub>3</sub>): 175.45 (C-16), 156.30 (C-13), 152.23 (C-8), 116.37 (C-15), 114.12 (C-14), 76.12 (C-12), 75.62 (C-3), 51.64 (C-9), 48.41 (C-5), 41.39 (C-10), 37.85 (C-4), 37.14 (C-7), 32.19 (C-1), 28.83 (C-17), 27.56 (C-11), 25.81 (C-2), 23.49 (C-6), 22.33 (C-18), 16.81 (C-19), 8.33 (C-20).

**Helioscopinolide E (10)****Colorless oil** $[\alpha]_{\text{D}}^{25} +395.9$  (CHCl<sub>3</sub>) $\nu_{\text{max}}\text{cm}^{-1}$  (KBr): 1706 cm<sup>-1</sup>, 1751 cm<sup>-1</sup> $m/z$  315 [M+H]<sup>+</sup>

<sup>1</sup>H NMR ( 300 MHz, CDCl<sub>3</sub>): 6.33 (1H, *s*, H-14), 4.87 (1H, *ddd*, *J* = 13.3, 6.1, 1.8, H-12), 2.56 (1H, *m*, H-2), 2.22 (4H, *m*, H-1, H-7), 1.84 (3H, *d*, *J* = 1.6, H-20), 1.59 (5H, *m*, H-5, H-6, H-11), 1.13 (3H, *s*, H-17), 1.08 (3H, *s*, H-19), 1.05 (3H, *s*, H-18).

<sup>13</sup>C NMR ( 75 MHz, CDCl<sub>3</sub>): 215.63 (C-3), 175.17 (C-16), 155.65 (C-13), 155.65 (C-13), 150.30 (C-8), 117.87 (C-15), 114.87 (C-14), 75.75 (C-12), 50.80 (C-9), 47.67 (C-4), 45.90

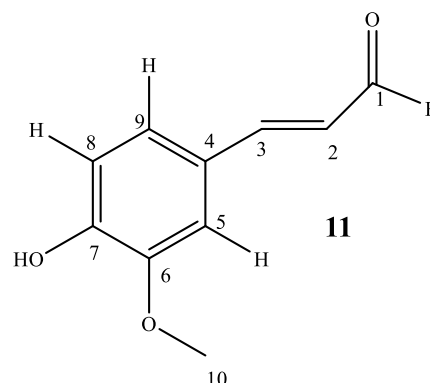
(C-5), 41.06 (C-10), 37.53 (C-1), 36.73 (C-7), 34.53 (C-2), 27.94 (C-11), 26.66 (C-17), 24.74 (C-6), 21.93 (C-18), 16.38 (C-19), 8.45 (C-20).

### Coniferaldehyde (**11**)

Amorphous light-brown powder

$\nu_{\max}$   $\text{cm}^{-1}$  (KBr): 3357  $\text{cm}^{-1}$ , 1691  $\text{cm}^{-1}$ ,  
1618  $\text{cm}^{-1}$

$m/z$  179 [M+H]<sup>+</sup>



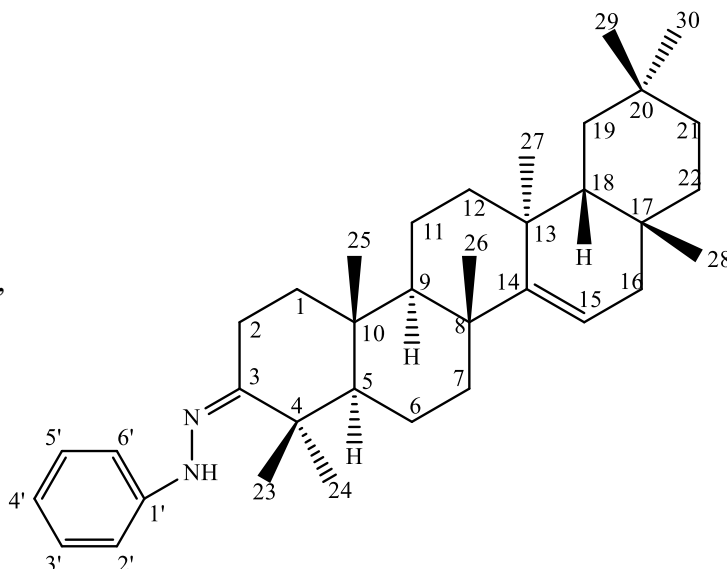
<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): 9.65 (1H, *d*, *J* = 7.7, H-1), 7.42 (1H, *d*, *J* = 15.7, H-3), 7.12 (1H, *dd*, *J* = 8.3, 2.0, H-9), 7.07 (1H, *d*, *J* = 1.9, H-5), 6.96 (1H, *d*, *J* = 8.1, H-8), 6.59 (1H, *dd*, *J* = 15.8, 7.7, H-2), 3.94 (3H, *s*, H-10).

<sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): 193,77 (C-1), 153,24 (C-3), 149,12 (C-7), 147,12 (C-6), 126,80 (C-4), 126,57 (C-2), 124,20 (C-9), 115,10 (C-8), 109,64 (C-5), 56,15 (C-10).

#### 1.8.3 Derivatization of taraxerone (**1**)

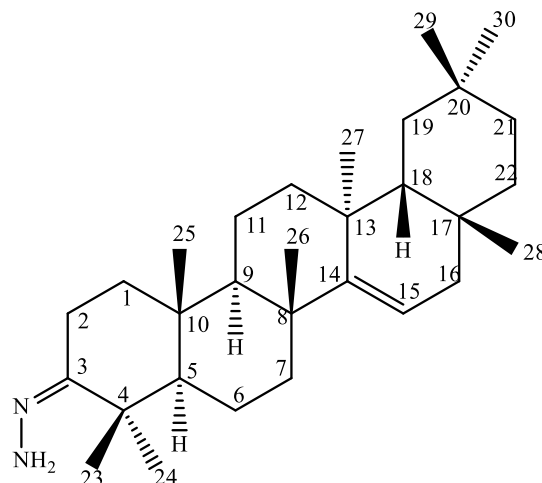
Taraxerone was derivatized by reaction with three different hydrazines: phenylhydrazine (reaction **1.1**), hydrazine monohydrate (reaction **1.2**) and 2,4-dinitrophenylhydrazine (reaction **1.3**). These reactions involved the mixture of 30 mg of taraxerone with approximately 5 equivalents of the corresponding reagent (40  $\mu\text{L}$  for **1.1**, 20  $\mu\text{L}$  for **1.2** and 60 mg for **1.3**), using methanol and dichloromethane (3:2, 10 mL) as solvent, with the addition of 200  $\mu\text{L}$  of a solution of acetic acid 10%, occurring overnight, under reflux.

The compounds were purified through column chromatography (SiO<sub>2</sub>), using as eluent mixtures of *n*-hexane/EtOAc (1:0 to 0:1), as is the case with **1.3** (34 mg of product, with a yield of 79%), or through preparative chromatography, as is the case with **1.1** (22 mg of product, with a yield of 61%) and **1.2** (13.3 mg of product, with a yield of 42%).

**Compound 1.1****Yellow amorphous solid** $[\alpha]_{\text{D}}^{25} +225.9$  (CHCl<sub>3</sub>) $\nu_{\text{max}}\text{cm}^{-1}$  (KBr): 3664 cm<sup>-1</sup>,  
1631 cm<sup>-1</sup>, 684 cm<sup>-1</sup> $m/z$  513 [M+H]<sup>+</sup>

<sup>1</sup>H NMR ( 300 MHz, CDCl<sub>3</sub>):  $\delta$  7.72 (1H, *m*, H-4'), 7.49 (4H, *dd*, *J* = 5.1, 2.0 Hz, H-2', H-3', H-5', H-6'), 5.54 (1H, *dd*, *J* = 8.1, 3.2 Hz, H-15), 1.14 (3H, *s*, H-27), 1.09 (3H, *m*, H-24), 1.07 (3H, *s*, H-25), 0.96 (3H, *s*, H-29), 0.95 (3H, *s*, H-26), 0.92 (3H, *s*, H-30), 0.84 (3H, *s*, H-28).

<sup>13</sup>C NMR ( 75 MHz, CDCl<sub>3</sub>):  $\delta$  158.15 (C-14), 151.11 (C-3), 131.32 (C-1'), 129.31 (C-3', C-5'), 122.76 (C-4'), 117.02 (C-15), 107.29 (C-2', C-6'), 51.74 (C-5), 48.95 (C-18), 48.86 (C-9), 42.50 (C-4), 41.25 (C-19), 39.23 (C-8), 38.27 (C-1), 37.88 (C-12), 37.71 (C-13), 36.86 (C-17), 35.95 (C-16), 35.30 (C-10), 34.00 (C-7), 33.93 (C-2), 33.51 (C-29), 33.28 (C-21), 30.10 (C-22), 30.01 (C-28), 28.97 (C-26), 26.22 (C-20), 23.16 (C-23), 22.42 (C-27), 21.49 (C-30), 21.43 (C-24), 18.78 (C-6), 17.61 (C-11), 15.70 (C-25).

**Compound 1.2****White amorphous solid** $[\alpha]_{\text{D}}^{25} +19.1$  (CHCl<sub>3</sub>) $\nu_{\text{max}}\text{cm}^{-1}$  (KBr): 3490 cm<sup>-1</sup>, 1638 cm<sup>-1</sup> $m/z$  439 [M+H]<sup>+</sup>

**$^1\text{H}$  NMR ( 300 MHz,  $\text{CDCl}_3$ ):**  $\delta$  5.55 (1H, *dd*,  $J = 8.3, 3.4$  Hz, H-15), 1.21 (3H, *s*, H-27), 1.11 (3H, *s*, H-25), 1.04 (3H, *s*, H-24), 0.95 (3H, *s*, H-29), 0.91 (3H, *s*, H-26), 0.89 (3H, *s*, H-30), 0.83 (3H, *s*, H-28).

**$^{13}\text{C}$  NMR ( 75 MHz,  $\text{CDCl}_3$ ):**  $\delta$  158.08 (C-14), 148.31 (C-3), 117.13 (C-15), 56.35 (C-5), 49.07 (C-18), 48.95 (C-9), 41.80 (C-4), 41.11 (C-19), 39.12 (C-8), 38.25 (C-1), 38.08 (C-12), 37.86 (C-13), 37.70 (C-17), 36.83 (C-16), 35.93 (C-10), 35.27 (C-7), 33.51 (C-29), 33.25 (C-22), 30.08 (C-28), 30.01 (C-26), 28.95 (C-20), 27.80 (C-2), 25.84 (C-21), 23.67 (C-23), 23.43 (C-30), 21.47 (C-24), 20.00 (C-6), 17.91 (C-27), 17.60 (C-11), 15.06 (C-25).

### Compound 1.3

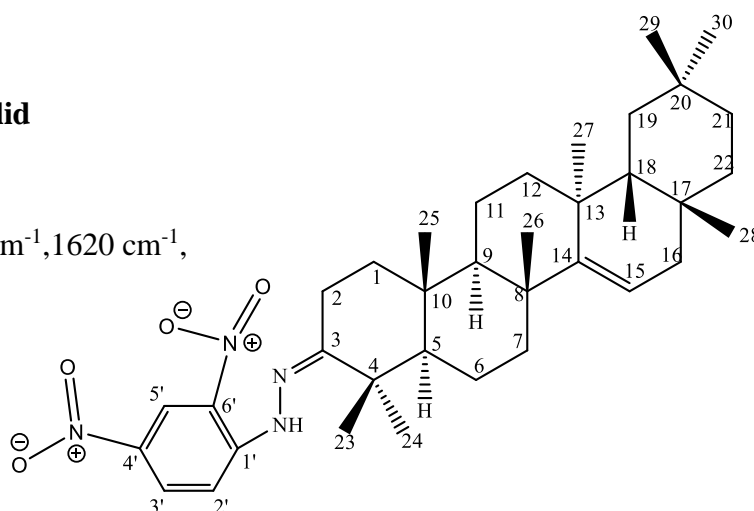
**Orange amorphous solid**

$[\alpha]_{\text{D}}^{25} +13.3$  ( $\text{CHCl}_3$ )

$\nu_{\text{max}} \text{cm}^{-1}$  (KBr): 3325  $\text{cm}^{-1}$ , 1620  $\text{cm}^{-1}$ ,

1541  $\text{cm}^{-1}$ , 1520  $\text{cm}^{-1}$

$m/z$  605  $[\text{M}+\text{H}]^+$



**$^1\text{H}$  NMR ( 300 MHz,  $\text{CDCl}_3$ ):**  $\delta$  9.12 (1H, *d*,  $J = 2.5$  Hz, H-5'), 8.29 (1H, *dd*,  $J = 9.6, 2.6$  Hz, H-3'), 7.95 (1H, *d*,  $J = 9.6$  Hz, H-2'), 5.56 (1H, *dd*,  $J = 8.2, 3.2$  Hz, H-15), 1.18 (3H, *s*, H-25), 1.15 (3H, *s*, H-27), 1.07 (3H, *s*, H-24), 0.95 (3H, *s*, H-29), 0.92 (3H, *s*, H-30), 0.90 (3H, *s*, H-26), 0.83 (3H, *s*, H-28).

**$^{13}\text{C}$  NMR ( 75 MHz,  $\text{CDCl}_3$ ):**  $\delta$  167.31 (C-3), 157.77 (C-14), 145.77 (C-4'), 137.65 (C-1'), 130.10 (C-6'), 129.11 (C-3'), 123.75 (C-2'), 117.36 (C-15), 116.64 (C-5'), 56.12 (C-5), 48.95 (C-18), 48.83 (C-9), 42.49 (C-4), 40.91 (C-19), 39.06 (C-8), 38.00 (C-1), 37.85 (C-12), 37.68 (C-13), 37.56 (C-17), 36.83 (C-16), 35.94 (C-10), 35.26 (C-7), 33.71 (C-21), 33.50 (C-29), 33.23 (C-22), 30.07 (C-28), 30.01 (C-26), 28.94 (C-20), 28.54 (C-2), 25.78 (C-23), 24.25 (C-27), 21.49 (C-30), 20.65 (C-24), 20.17 (C-6), 17.63 (C-11), 14.79 (C-25).

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