1 Production of novel biosurfactant by a new yeast species isolated

2 from Prunus mume Sieb. et Zucc.

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23 Abstract

Biosurfactants reduce surface and interfacial tension due to their amphiphilic properties, 24 and are an eco-friendly alternative for chemical surfactants. In this study, a novel yeast strain 25 JAF-11 that produces biosurfactant was selected using drop collapse method, and the properties 26 of the material were investigated. The nucleotide sequences of the strain were compared with 27 closely related strains and identified based on the D1/D2 domain of the large-subunit rDNA 28 (LSU) and internal transcribed spacer (ITS) regions. *Neodothiora populina* CPC 39399^T, the 29 closest species with strain JAF-11 in the phylogenetic tree, showed a sequence similarity of 30 97.75% for LSU and 94.27% for ITS, respectively. The result suggests that the strain JAF-11 31 32 represent a distinct species which cannot be assigned to any existing genus or species in the family Dothideaceae. Strain JAF-11 was able to produce biosurfactant reducing the surface 33 tension of medium to 34.5 mN/m on the 6th day of culture and the result of measuring the 34 critical micelle concentration (CMC) by extracting the crude biosurfactant was found to be 24 35 mg l⁻¹. The molecular weight 502 of the purified biosurfactant was confirmed by measuring 36 the fast atom bombardment mass spectrum (FAB-MS). The chemical structure was analyzed 37 by measuring ¹H nuclear magnetic resonance (NMR), ¹³C NMR, two-dimensional NMRs of 38 39 the compound. The molecular formula was $C_{26}H_{46}O_{9}$, and it was composed of one octanoyl group and two hexanovl group to myo-inositol moiety. The new biosurfactant is the first report 40 of a compound produced by a novel yeast strain JAF-11. This new biosurfactant is proposed 41 42 as potential candidate for use in a variety field.

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45 **KEYWORDS**: Yeast, novel species, novel biosurfactant, chemical structure

47 Introduction

48

The surfactants have amphiphilic properties possessing both non-polar (hydrophobic) 49 and polar (hydrophilic) moieties that allows reducing the surface and interfacial tension 50 between biphasic systems as liquid-liquid interface or solid-liquid boundaries [1, 2]. The 51 surfactants are one of the important compounds having potential commercial application of 52 detergents, cosmetics, food ingredients, agricultural, pharmaceutical, paint, textile and paper 53 etc. [3, 4]. However, surfactants are mainly chemically synthesized from petroleum-based 54 55 resources, which can cause environmental problems due to toxicity [5]. As environmental 56 awareness has gradually increased over the past decades, the demand for eco-friendly compounds has increased, and accordingly, there is an increasing interest in microbial 57 58 biosurfactants [6]. In particular, microbe-derived biosurfactants have advantages such as environmental compatibility, activity, stability, and lower toxicity compared to chemically 59 60 synthesized equivalents [7-11].

61 Biosurfactants are classified according to ionic charges (anionic, cationic, non-ionic and neutral biosurfactants), molecular weight (high molecular and low molecular weight 62 biosurfactants), secretion type (intracellular, extracellular and adhered to microbial cells), and 63 64 chemical structure (glycolipids, lipopeptides, fatty acids, phospholipids, neutral lipids and polymeric biosurfactants), and it has also been previously reported that various biosurfactants 65 are produced depending on the type of microorganisms [12]. Among them, glycolipids have 66 67 been best known as a class of biosurfactants that exhibit the highest commercial potential due to the high microbial productivity [13]. The microbe-derived biosurfactant rhamnolipids, also 68 69 known as glycolipids, was described for the first time in 1946 [14]. Despite their high productivity, most of them is produced by opportunistic pathogen Pseudomonas aeruginosa 70 [15]. The sophorolipids belonging to the group of glycolipid are one of the most promising 71

72 biosurfactant because they are produced by non-pathogen yeast Starmerella bombicola and other Candida spp. (Candida stellate, C. riodocensis, C. apicola, C. batistae, C. kuoi, and C. 73 floricola) [16, 17]. Mannosylerythritol lipids (MEL) by a variety of Pseudozyma yeasts and 74 trehalolipids by *Rhodococcus erythopolis* have been reported as representative biosurfactants 75 belonging to the glycolipid [18]. It has been reported that microorganisms such as the genera 76 Acinetobacter, Arthrobacter, Pseudomonas, Halomonas, Bacillus, Rhodococcus, Enterobacter, 77 and few yeast genera produce not only glycolipids but also other types of biosurfactants [19]. 78 In the 20th century, research on biosurfactants focused on understanding and 79 optimizing the production process, and since then, research on utilizing various renewable 80 resources, discovering new producer strains, or developing genetically modified strains has 81 82 been reported [6]. Since biosurfactants have a wide range of functional properties depending on their chemical structure [19], it is necessary to continuously secure a variety of 83 biosurfactants in order to be applied to various fields. The present study is the first report of 84 novel biosurfactant extracted from the novel yeast strain JAF-11. We isolated a new species of 85 yeast that produces biosurfactant from flower in order to secure various biosurfactants, and 86 87 chemical structure of the biosurfactant extracted from the yeast has been characterized and identified as a novel biosurfactant by nuclear magnetic resonance spectrometry techniques. 88

89

90 Materials and methods

91

92 Culture medium for biosurfactant production by yeast

93 The culture medium composition used in these studies for biosurfactant-producing
94 yeast was as follows (w/v): glucose (1.5%), soybean oil (1.5%), ammonium sulfate (0.1%),
95 potassium phosphate (0.25%), sodium phosphate (0.01%), magnesium sulfate (0.05%),

96 calcium chloride (0.01%), manganese sulfate (0.002%) and peptone (0.1%).

97

98 Isolation and screening of biosurfactant producing yeast

The biosurfactant-producing yeast used in this study was isolated from flower (*Prunus mume* Sieb. *et* Zucc.) collected from apricot village in Gwangyang, Republic of Korea during
March 2018. The biosurfactant was screened using modified drop collapse method as follows:
102 100µL of the culture supernatant and water (1:1, v/v) was pipetted and placed on parafilm [20,
21]. The selected strain JAF-11 was maintained for storage at -80°C in 15% (v/v) glycerol, and
was deposited with the patent depository as KACC 83047BP.

105

106 Identification of biosurfactant producing yeast

The identification of strain JAF-11 was conducted based on multigene phylogenetic analysis of the nucleotide sequences combined with the D1/D2 region of large-subunit (LSU) and the internal transcribed spacer (ITS) region ribosomal DNA (rDNA) genes. The DNA sequencing was performed by Macrogen Inc. (Seoul, Korea) and the gene sequences of related species were retrieved from GenBank database. The phylogenetic tree was inferred by using the maximum likelihood method with 1,000 bootstrap replicates and sequences analysis was performed in MEGA X software.

114

115 **Time course of the growth and determination of surface tension**

The strain JAF-11 was grown in 500 ml culture medium at 25℃ on a rotary shaker at 150 rpm, and then the optical density at 600 nm was measured for 8 days by a UV spectrophotometer. To confirm the production of biosurfactant, the culture medium of strain

119 JAF-11 was prepared through a 0.22 μ m filter and measured the surface tension(ST) every

day. For ST measurement, the Wilhelmy plate method was used at room temperature with a

121 force tensiometer K11 (Krüss, Germany). All tests were performed in triplicates.

122

Measurement of critical micelle concentration (CMC) of biosurfactant

To extract the crude biosurfactant, 8 L culture medium inoculated with freshly grown 125 strain JAF-11 was mixed with HP-20 non-polar resin (Mitsubishi chemical, Japan) and eluted 126 127 in methanol. After removing methanol by rotary vacuum evaporation, the concentrated solution was partitioned with an equal volume of ethyl acetate. The ethyl acetate was concentrated under 128 vacuum evaporator, and the product was used for CMC measurement after purification using 129 Flash silica gel column chromatography (CHCl₃:MeOH, 50:1 to 1:1, v/v, stepwise) (SK 130 chemical, Korea). The crude biosurfactant were dissolved in distilled water and serially diluted 131 to concentrations in the range of 0-250 mg l⁻¹. The CMC was determined by plotting the surface 132 tension against the log of the biosurfactant concentration using a force tensiometer K11 (Krüss, 133 Germany) [22]. 134

135

136 Isolation and purification of biosurfactant compound

A yeast strain of JAF-11 was cultured in a medium described above for 5 days at 25°C on a rotary shaker incubator at 150 rpm. The culture broth was subjected to Diaion HP-20 column chromatography and eluted with 30% aq. MeOH, 70% aq. MeOH, MeOH, and acetone. The biosurfactant activity of eluates were evaluated by the drop collapse method, and the MeOH eluate was chosen and concentrated by rotary vacuum evaporator. The concentrate was

142	partitioned between ethyl acetate and water. The ethyl acetate portion showing biosurfactant
143	activity was concentrated and applied to silica gel column chromatography (Merck, Germany)
144	eluted with CHCl ₃ :MeOH (50:1, 20:1, 10:1, 5:1, 2:1, and 1:1, v/v) (Fig 1). An active fraction,
145	CHCl ₃ :MeOH (20:1), was subjected to Sephadex LH-20 (GE Healthcare, Sweden) column
146	chromatography eluted with $CHCl_3$:MeOH (1:1, v/v). The fractions showing biosurfactant
147	activity were combined and concentrated (Fig 1). The concentrate was further separated by the
148	medium pressure liquid chromatography (MPLC, Teledyne ISCO, USA) equipped with
149	Redisep Rf C ₁₈ reversed-phase column (Teledyne Isco, USA) using a gradient of $70\% \rightarrow 100\%$
150	aq. MeOH to yield compound 1 (13.4 mg).
151	
152	Fig 1. Purification scheme of the biosurfactant produced by strain JAF-11.
153	
153 154	Chemical structure analysis of biosurfactant
153 154 155	Chemical structure analysis of biosurfactant The fast atom bombardment mass spectrum (FAB-MS) and high-resolution fast atom
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166 **Result and discussion**

167 Isolation and identification of biosurfactant producing yeast

Strain JAF-11 was isolated from Prunus mume Sieb. et Zucc. in Republic of Korea 168 and was selected as a potential biosurfactant producer using modified drop collapse method 169 170 (S1 Fig). Phylogenetic relationships of strain JAF-11 and the closely related strains were inferred using concatenated LSU and ITS sequences [23, 24]. The LSU and ITS sequences 171 from the reference strains were obtained from GenBank database (Table 1). Phylogenetic 172 analysis revealed that JAF-11 belonged to the family *Dothideales* clade, and formed one 173 compact cluster with Neodothiora populina CPC 39399^T, Rhizosphaera macrospora CBS 174 208.79^T and *Phaeocryptopus nudus* CBS 268.37 (Fig 2). The LSU region sequence of strain 175 JAF-11 showed the highest similarity with those of *Rhizosphaera macrospora* CBS 208.79^T 176 (97.26%; 15nt substitutions in 548 nt) and Neodothiora populina CPC 39399^T (97.75%; 13nt 177 178 substitutions in 578 nt). In the ITS region sequences, strain JAF-11 had sequence similarity of 95.15% (26 nt substitutions in 536 nt) with *Rhizosphaera macrospora* CBS 208.79^T and 94.27% 179 (31 nt substitutions in 541 nt) with *Neodothiora populina* CPC 39399^T. According to Duong 180 Vu et al. (2016), the taxonomic threshold predicted to discriminate yeast species is 98.41% in 181 the LSU region and 99.51% in the ITS region [25]. Consequently, we propose that strain JAF-182 183 11 represents a novel yeast species of a new genus based on phylogenetic analysis and the taxonomic thresholds of gene sequence identities. 184

185

Table 1. GenBank accession numbers of species used in phylogenetic analyses

Creation	Strain no.	GenBank accession no.	
Species		LSU	ITS
Family Dothideaceae			
Coleophoma crateriformis	CBS 473.69	MH871117.1	MH859358.1
Coleophoma oleae	CBS 615.72	MH872293.1	KU728511.1

Cylindroseptoria ceratoniae	CBS 477.69 [™]	KF251655.1	KF251151.1
Delphinella strobiligena	CBS 735.71	MH872074.1	MH860318.1
Dothidea insculpta	CBS 189.58	MH869284.1	AF027764.1
Dothidea sambuci	AFTOL-ID 274 [™]	AY544681.1	DQ491505.1
Dothidea sambuci	DAOM 231303 ^T	NG_027611.1	AY883094.1
Dothiora cannabinae	CBS 737.71 [⊤]	MH872076.1	MH860320.1
Dothiora corymbiae	CBS 145060 [™]	MK047482.1	MK047431.1
Dothiora elliptica	CBS 736.71	MH872075.1	KU728502.1
Endoconidioma populi	UAMH 10902	HM185488.1	HM185487.1
Neocylindroseptoria pistaciae	CBS 471.69 [⊤]	MH871115.1	MH859357.1
Neodothiora populina	CPC 39399 [™]	MW175405.1	MW175365.1
Phaeocryptopus nudus	CBS 268.37	GU301856.1	EU700371.1
Plowrightia ribesia	MFLU 14-0040	KM388552.1	KM388544.1
Pringsheimia smilacis	CBS 873.71	FJ150970.1	MH860390.1
Rhizosphaera macrospora	CBS 208.79 [⊤]	MH872971.1	MH861202.1
Stylodothis puccinioides	CBS 193.58	MH869286.1	MH857753.1
Sydowia polyspora	CBS 116.29	MH866487.1	MH855019.1
Family Aureobasidiaceae			
Aureobasidium leucospermi	CPC 15180	JN712555.1	JN712489.1
Aureobasidium leucospermi	CBS 130593 [⊤]	MH877257.1	KT693727.1
Aureobasidium proteae	CPC 2825	JN712558.1	JN712492.1
Aureobasidium pullulans	MFLUCC 14-0288	KM461701.1	KM388542.1
Aureobasidium pullulans	CBS 584.75 [™]	DQ470956.1	FJ150906.1
Kabatiella lini	CBS 125.21 [⊤]	FJ150946.1	FJ150897.1
Family Cladosporiaceae			
Rachicladosporium cboliae	CBS 125424 ^T	MH875168.1	MH863703.1

187

Fig 2. Phylogenetic tree of concatenated LSU and ITS region sequences of closely related species. *Rachicladosporium choliae* was used as the outgroup in the phylogenetic tree. The phylogenetic tree was constructed using the maximum likelihood method and Tamura-Nei model with bootstrap values 1,000 replicates. The scale bar indicates substitutions per nucleotide position.

193

194 Surface-active properties of biosurfactant

195 Growth of strain JAF-11 in culture medium was detected by measuring absorbance at

196 660 nm and surface tensions of the aqueous supernatant was measured using force tensiometer

197	K11 (Krüss, Germany). While the growth of the strain increased continuously for 8 days, the
198	surface tension of the supernatant decreased from 53 mN/m to 34.5 mN/m for 6 days and
199	increased again after 7days. The surface tension recorded the lowest values of 34.5~34.6 mN/m
200	after 5-6 days of incubation (Fig 3). The above results indicated that the highest biosurfactant
201	production in strain JAF-11 is reached at 5-6 days before the stationary growth phase.
202	
203 204 205	Fig 3. Time course of growth kinetics and surface tension in culture medium during cultivation of strain JAF-11.
205	The artical miscalle concentration (CMC) is defined as the concentration of surfactant
206	The critical micelle concentration (CMC) is defined as the concentration of surfactant
207	required to start the micelles formation. It is determined by plotting the surface tension
208	measured according to the concentration of biosurfactant and identifying the point at which the
209	surface tension of the biosurfactant no longer decreases dramatically. As a results of measuring
210	surface tension of the water with the crude biosurfactants isolated from culture medium, the
211	values were from 72.23 mN/m to 32.80 mN/m and the minimum surface tension value was
212	32.80 mN/m. In particular, the value of CMC was 24 mg/L that the concentration of the
213	biosurfactant obtained from slope of the curve abruptly changed as shown in Figure 4.
214	
215 216	Fig 4. Determination of critical micelle concentration of producing crude biosurfactant from strain JAF-11.
217	
218	Chemical structure of the isolated compound
219	Chemical structure of the biosurfactant isolated was determined by mass and NMR
220	measurements. The molecular weight of 502 was determined by the FAB-MS measurement,

221 which showed a quasi-molecular ion peak at m/z 503 [M+H]⁺ (Fig 5). The molecular formula,

222 $C_{26}H_{46}O_9$, was determined by the HR-FAB-MS providing a molecular ion peak at m/z 503.3243

223	$[M+H]^+$ (calcd. for C ₂₆ H ₄₇ O ₉ , 503.3220), indicating four degrees of unsaturation. The ¹ H NMR
224	spectrum of 1 (Table 2) showed signals due to six oxygenated methines at $\delta_{\rm H}$ 5.50 (t, $J = 2.7$
225	Hz, H-2), 5.28 (t, J = 10.0 Hz, H-4) , 4.93 (dd, J = 10.0, 2.7 Hz, H-3), 3.66 (t, J = 9.5 Hz, H-
226	6), 3.65 (dd, $J = 10.0$, 2.7 Hz, H-1), and 3.45 (t, $J = 9.0$ Hz, H-5). It also showed signals
227	attributable to 14 methylenes at $\delta_{\rm H}$ 2.45 (m, H-2')/2.42 (m, H-2'), 2.36 (m, H-2'')/2.31 (m, H-
228	2"), 2.18 (m, H-2""), 1.68 (m, H-3"), 1.59 (m, H-3"), 1.55 (m, H-3""), and 1.35-1.25 (overlapped)
229	and three methyls at $\delta_{\rm H}$ 0.90 (overlapped). The ¹³ C NMR spectrum (Table 2) in combination
230	with HMQC spectrum displayed signals due to three carbonyl carbons at $\delta_{\rm C}$ 175.0 (C-1'), 174.6
231	(C-1"), and 174.2 (C-1""), six oxygenated methine carbons at $\delta_{\rm C}$ 74.6 (C-6), 74.2 (C-5), 73.2
232	(C-4), 72.6 (C-2), 71.7 (C-3), and 70.9 (C-1), 14 methylene carbons at $\delta_{\rm C}$ 23.4–35.2, and three
233	methyl carbons at $\delta_{\rm C}$ 14.4 (C-6''') and 14.2 (C-8' and C-6''). The ¹ H- ¹ H COSY correlations
234	among six oxygenated methine protons established the presence of an inositol moiety. Inositol
235	moiety was identified as a myo-inositol by the proton coupling constant. Except for an
236	equatorial proton at $\delta_{\rm H}$ 5.50 (H-2) with coupling constant of 2.7 Hz, other protons occupied an
237	axial position based on their proton coupling constants. The ¹ H- ¹ H COSY spectrum also
238	established six partial structures in three acyl chains, as shown in Fig 6B. Chemical structure
239	was unambiguously determined by the HMBC spectrum, which exhibited the long-range
240	correlations from two methyl protons at $\delta_{\rm H}$ 0.90 (H-8" and H-8"") to two methylene carbons at
241	$\delta_{\rm C}$ 32.4 (C-4" and C-4""), from the methylene proton at $\delta_{\rm H}$ 1.59 to the carbonyl carbon at $\delta_{\rm C}$
242	174.6 and the methylene carbons at $\delta_{\rm C}$ 32.4 and 23.4, and the methylene proton at $\delta_{\rm H}$ 1.55 to
243	the carbonyl carbon at $\delta_{\rm C}$ 174.2 and the methylene carbons at $\delta_{\rm C}$ 32.4 and 23.4, implying the
244	presence of two hexanoyl moieties. Other HMBC correlations from the methylene protons at
245	$\delta_{\rm H}$ 2.45/2.42 and 1.68 to the carbons at $\delta_{\rm C}$ 175.0 and 30.2 and from the methyl protons at $\delta_{\rm H}$
246	0.90 to the carbon at $\delta_{\rm C}$ 32.9, and 14 methylenes in the ¹ H and ¹³ C NMR spectra indicated the

247	presence of an octanoyl moiety. Finally, the long-range correlations from the oxygenated
248	methines at $\delta_{\rm H}$ 5.50, 4.93, and 5.28 to the carbonyl carbons at $\delta_{\rm C}$ 175.0, 174.2, and 174.6,
249	respectively, revealed that C-2, C-3, and C-4 in inositol moiety were acylated with one octanoyl
250	and two hexanoyl groups, as shown in Fig 6B. Taken together, the structure of compound 1
251	was determined to be a new myo-inositol derivative and was named JAF-11. This compound
252	was very similar to pullusurfactan E isolated from Aureobasidium pullulans strain A11211-4-
253	57 from fleabane flower, Erigeron annus (L.) pers. [26], except for the presence of octanoyl
254	moiety instead of hexanoyl moiety in pullusurfactan E (Fig 6A). Although the chemical
255	structure is similar to pullusurfactan E, this study reports the isolation of a new biosurfactant
256	from the novel yeast strain JAF-11 for the first time.
257	

Fig 5. Fast atom bombardment mass spectrum of purified biosurfactant in the positive ion mode.

260

No.	δ_{c}	δ_{H}
1	70.9	$3.65 (1H, dd, J = 10.0, 2.7)^{a}$
2	72.6	5.50 (1H, t, J = 2.7)
3	71.7	4.93 (1H, dd, <i>J</i> = 10.0, 2.7)
4	73.2	5.28 (1H, t, J = 10.0)
5	74.2	3.45 (1H, t, J = 9.0)
6	74.6	3.66 (1H, t, J = 9.5)
1'	175.0	
2'	35.2	2.42 (1H, m), 2.45 (1H, m)
3'	26.3	1.68 (2H, m)
4'	30.2	1.38 (2H, m)
5'	30.2	1.25-1.35 (2H, m)
6'	32.9	1.25-1.35 (2H, m)
7'	23.7	1.25-1.35 (2H, m)
8'	14.2	0.90 (3H, overlapped)
1"	174.6	
2"	35.1	2.31 (1H, m), 2.36 (1H, m)
3"	25.7	1.59 (2H, m)
4"	32.4	1.32 (2H, m)
5"	23.4	1.25-1.35 (2H, m)

261 **Table 2.** ¹H and ¹³C NMR spectral data of new biosurfactant in CD₃OD.

<i>(</i>)?	14.0	0.00 (211 arrandamend)
0	14.2	0.90 (3H, overlapped)
1'''	174.2	
2""	35.0	2.18 (2H, m)
3'''	25.4	1.55 (2H, m)
4""	32.4	1.28 (2H, m)
5""	23.4	1.25-1.35 (2H, m)
6'''	14.4	0.90 (3H, overlapped)

^a Proton resonance integral, multiplicity, and coupling constant (*J*=Hz) in parentheses

Fig 6. (A) Chemical structure of novel biosurfactant and (B) two-dimensional NMR correlations of novel biosurfactant.

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281	References
282 283	1. Pacwa-Płociniczak MP, Plaza GA. Piotrowska-Seget Z, Cameotra SS. Environmental
284	applications of biosurfactants: recent advances. Int J Mol Sci. 2011;12(1):633-54.
285	2. Ramani K, Jain SC, Mandal AB, Sekaran G. Microbial induced lipoprotein biosurfactant
286	from slaughterhouse lipid waste and its application to the removal of metal ions from aqueous
287	solution. Colloids Surf. B. 2012;97:254-63.
288	3. Vaz DA, Gudiña EJ, Alameda EJ, Teixeira JA, Rodrigues LR. Performance of a
289	biosurfactant produced by a Bacillus subtilis strain isolated from crude oil samples as compared
290	to commercial chemical surfactants. Colloids Surf. B. 2012;89:167-74.
291	4. Rebello S, Asok AK, Mundayoor S, Jisha MS. Surfactants: toxicity, remediation and green
292	surfactants. Environ Chem Lett. 2014;12(2):275-87.
293	5. Singh P, Tiwary BN. Isolation and characterization of glycolipid biosurfactant produced by
294	a Pseudomonas otitidis strain isolated from Chirimiri coal mines, India. Bioresour Bioprocess.
295	2016;3(1):42.
296	6. Claus S, Van Bogaert INA. Sophorolipid production by yeasts: a critical review of the
297	literature and suggestions for future research. Appl Microbiol Biotechnol. 2017;101:7811-21.
298	7. Kosaric N. Biosurfactants in industry. Pure and Applied Chemistry. 1992;64(11):1731-7.
299	8. Desai JD, Banat IM. Microbial production of surfactants and their commercial potential.
300	Microbiol Mol Biol Rev. 1997;61(1):47-64.
301	9. Rahman KSM, Rahman TJ, Kourkoutas Y, Petsas I, Marchant R, Banat IM. Enhanced
302	bioremediation of n-alkane in petroleum sludge using bacterial consortium amended with
303	rhamnolipid and micronutrients. Bioresour Technol. 2003;90(2):159-68.
304	10. Das K, Mukherjee AK. Comparison of lipopeptide biosurfactants production by Bacillus
305	subtilis strains in submerged and solid state fermentation systems using a cheap carbon source:

- 306 Some industrial applications of biosurfactants. Process Biochem. 2007;42(8):1191-9.
- 307 11. Das P, Mukherjee S, Sen R. Improved bioavailability and biodegradation of a model
- 308 polyaromatic hydrocarbon by a biosurfactant producing bacterium of marine origin.
- 309 Chemosphere. 2008;72(9):1229-34.
- 310 12. Mnif I, Ellouz-Chaabouni S, Ghribi D. Glycolipid biosurfactants, main classes, functional
- 311 properties and related potential applications in environmental biotechnology. J Polym Environ.
- 312 2018;26(5):2192-206.
- 313 13. Chen J, Wu Q, Hua Y, Chen J, Zhang H, Wang H. Potential applications of biosurfactant
- 314 rhamnolipids in agriculture and biomedicine. Appl Microbiol and Biotechnol.
 315 2017;101(23):8309-19.
- 14. Bergström S, Theorell H, Davide H. On a metabolic product of *Ps. Pyocyanea*, Pyolipic
 acid, active against *M. tuberculosis*. Arch Biochem. 1946;10:165-6.
- 15. Henkel M, Müller MM, Kügler JH, Lovaglio RB, Contiero J, Syldatk C, et al. Rhamnolipids
- as biosurfactants from renewable resources: Concepts for next-generation rhamnolipid
 production. Process Biochem. 2012;47(8):1207-19.
- 16. Van Bogaert INA, Zhang J, Soetaert W. Microbial synthesis of sophorolipids. Process
 Biochem. 2011;46(4):821-33.
- 323 17. Jezierska S, Claus S, Van Bogaert I. Yeast glycolipid biosurfactants. FEBS Lett.
 324 2018;592(8):1312-29.
- 18. Morita T, Fukuoka T, Imura T, Kitamoto D. Mannosylerythritol lipids: production and
 applications. J Oleo Sci. 2015;64(2):133-41
- 19. Shekhar S, Sundaramanickam A, Balasubramanian T. Biosurfactant producing microbes
 and their potential applications: A review. Crit Rev Environ Sci Technol. 2015;45(14):152254.
- 20. Jain DK, Collins-Thompson DL, Lee H, Trevors JT. A drop-collapsing test for screening

- 331 surfactant-producing microorganisms. Journal of Microbiological Methods. 1991;13(4):271-9.
- 332 21. Kim JS, Lee IK, Yun BS. A novel biosurfactant produced by Aureobasidium pullulans L3-
- GPY from a Tiger lily wild flower, *Lilium lancifolium* Thunb. PLoS One. 2015;10(4):e0122917.
- 334 22. Sheppard JD, Mulligan CN. The production of surfactin by *Bacillus subtilis* grown on peat
- hydrolysate. Appl Microbiol Biotechnol. 1987;27(2):110-6.
- 23. Thambugala KM, Ariyawansa HA, Li Y-M, Boonmee S, Hongsanan S, Tian Q, et al.
- 337 Dothideales. Fungal Divers. 2014;68(1):105-58.
- 24. Crous PW, Cowan DA, Maggs-Kölling G, Yilmaz N, Larsson E, Angelini C, et al. Fungal
- planet description sheets: 1112-1181. Persoonia. 2020;45:251-409.
- 340 25. Vu D, Groenewald M, Szöke S, Cardinali G, Eberhardt U, Stielow B, et al. DNA barcoding
- analysis of more than 9,000 yeast isolates contributes to quantitative thresholds for yeast
 species and genera delimitation. Stud Mycol. 2016;85:91-105.
- 343 26. Kim JS, Lee IK, Yun BS. Pullusurfactans A-E, new biosurfactants produced by
- 344 Aureobasidium pullulans A11211-4-57 from a fleabane, Erigeron annus (L.) pers. J Antibiot
- 345 (Tokyo). 2018;71:920-6.
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355 Supporting information

- **S1 Fig. Screening of biosurfactant producing yeast by drop collapse method**. Distilled water and fresh culture broth were used as a control.
- 358 **S2 Fig.** ¹H NMR spectrum of the purified biosurfactant
- 359 S3 Fig. ¹³C NMR spectrum of the purified biosurfactant
- 360 S4 Fig. HMQC spectrum of the purified biosurfactant
- 361 **S5 Fig. ¹H-¹H COSY spectrum of the purified biosurfactant**
- 362 S6 Fig. HMBC spectrum of the purified biosurfactant

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0.050



Surface tension









Fig6A



Fig6B